

Conversion factors for units

ICS 01.060; 17.020

Confirmed
January 2010

Committees responsible for this British Standard

The preparation of this British Standard was entrusted to Technical Committee SS/7, General metrology, quantities and units, upon which the following bodies were represented:

British Measurement and Testing Association
 City University
 Department of Trade and Industry, National Weights and Measures Laboratory
 Federation of Small Businesses
 Institute of Measurement and Control
 Institution of Electrical Engineers
 National Physical Laboratory
 Royal Society of Chemistry
 Society of Chemical Industry
 Trading Standards Institute
 Coopted members

This British Standard was published under the authority of the Standards Policy and Strategy Committee on 25 May 2004

© BSI 25 May 2004

First published June 1930
 Second edition July 1944
 Third edition, as BS 350-1, February 1959
 Fourth edition, as BS 350-1, March 1974
 Fifth edition, as BS 350, May 2004

The following BSI references relate to the work on this British Standard:
 Committee reference SS/7
 Draft for comment 03/659657 DC

ISBN 0 580 43516 4

Amendments issued since publication

Amd. No.	Date	Comments

Contents

	Page
Committees responsible	Inside front cover
Foreword	iv
<hr/>	
1 Scope	1
2 Number	1
3 Length	2
4 Area (length squared)	6
5 Volume and capacity (length cubed)	9
6 Modulus of section, first moment of area	12
7 Second moment of area, or geometrical moment of inertia	12
8 Plane angle	17
9 Solid angle	18
10 Time	18
11 Linear velocity (speed) (length/time)	19
12 Angular velocity (angle/time)	20
13 Frequency (number/time)	22
14 Acceleration (length/time squared)	23
15 Mass	24
16 Mass per unit length (or lineic mass) (formerly linear density) (mass/length)	29
17 Mass per unit area (areic mass) (mass/length squared)	31
18 Specific surface, or area per unit mass	31
19 Area per unit capacity	34
20 Density (volumic mass), (mass/volume)	34
21 Mass concentration (mass/volume)	36
22 Specific volume (volume/mass)	38
23 Mass rate of flow (mass/time)	38
24 Volume rate of flow (volume/time)	40
25 Traffic factors	42
26 Moment of inertia (mass \times length squared)	43
27 Momentum (linear) (mass \times velocity)	43
28 Angular momentum (mass \times velocity \times length)	43
29 Force (mass \times acceleration)	44
30 Weight	46
31 Moment of force, or torque (force \times length)	46
32 Force per unit length (force/length)	49
33 Pressure (force/area)	49
34 Stress (force/area)	52
35 Viscosity, dynamic (stress/velocity gradient)	56
36 Viscosity, kinematic (length squared/time)	58
37 Energy (work, heat, etc.)	60
38 Power (energy/time)	62
39 Temperature, including temperature difference or interval	66
40 Specific energy [(energy or heat)/mass]	67
41 Heat content, volume basis (heat/volume)	68
42 Specific heat capacity [heat/(mass \times temperature interval)]	72
43 Specific entropy [heat/(mass \times thermodynamic temperature)]	73
44 Heat capacity, volume basis [heat/(volume \times temperature interval)]	73

	Page
45 Heat flux density [heat/(area × time)]	74
46 Thermal conductance (heat transfer coefficient)	77
47 Thermal conductivity [heat × length/(area × time × temperature difference)]	78
48 Thermal resistivity [area × time × temperature difference/(heat × length)]	79
49 Heat release rate (e.g. as used in connection with furnaces) [heat/(volume × time)], or (power/volume)	80
50 Thermal diffusivity (area/time)	80
<hr/>	
Annex A (informative) Commentary on imperial and metric systems of measurement and units	81
<hr/>	
Index of symbols and abbreviations	84
Index of terms	89
<hr/>	
Bibliography	105
<hr/>	
Table 1 — Prefixes denoting decimal multiples or submultiples	1
Table 2 — Meaning of million, billion, trillion, etc.	2
Table 3 — Length	5
Table 4 — Area	8
Table 5 — Area of section of wire	9
Table 6 — Volume and capacity	13
Table 7 — Volume and capacity (<i>continued</i>)	14
Table 8 — Volume and capacity (<i>continued</i>)	15
Table 9 — Relationship between UK (imperial) and US units of capacity	16
Table 10 — Second moment of area	16
Table 11 — Plane angle	18
Table 12 — Linear velocity	21
Table 13 — Angular velocity and velocity of rotation	22
Table 14 — Acceleration	23
Table 15 — Mass	26
Table 16 — Mass (<i>continued</i>)	27
Table 17 — Mass (<i>continued</i>)	28
Table 18 — Mass per unit length (lineic mass)	30
Table 19 — Mass per unit area (areic mass)	32
Table 20 — Specific surface, or area per unit mass	33
Table 21 — Area per unit capacity	34
Table 22 — Density (volumic mass) (mass/volume)	35
Table 23 — Mass concentration	37
Table 24 — Specific volume	39
Table 25 — Mass rate of flow	40
Table 26 — Volume rate of flow	41
Table 27 — Fuel consumption (volume/distance)	42
Table 28 — Fuel consumption (distance/volume)	42
Table 29 — Moment of inertia	43
Table 30 — Force	45
Table 31 — Moment of force (torque)	48
Table 32 — Pressure	53
Table 33 — Pressure (<i>continued</i>)	54
Table 34 — Pressure (<i>continued</i>)	55

	Page
Table 35 — Viscosity (dynamic)	57
Table 36 — Viscosity (kinematic)	59
Table 37 — Energy	63
Table 38 — Energy (<i>continued</i>)	64
Table 39 — Power	65
Table 40 — Equivalent values on four temperature scales	67
Table 41 — Specific energy	69
Table 42 — Calorific value, volume basis	70
Table 43 — Calorific value of gases, volume basis (with differing reference conditions)	71
Table 44 — Conversion factors previously used by the UK Gas Industry	72
Table 45 — Specific heat, mass basis	75
Table 46 — Heat capacity, volume basis	76
Table 47 — Heat flux density, intensity of heat flow rate	76
Table 48 — Thermal conductance	77
Table 49 — Thermal conductivity	78
Table 50 — Thermal resistivity	79
Table 51 — Heat release rate	80
Table A.1 — Other metric systems	82
Table A.2 — Base and supplementary quantities, units and symbols in the SI system	82

Foreword

This British Standard has been prepared by Technical Committee SS/7. It supersedes BS 350-1:1974 which is withdrawn.

BS 350 was first published in 1930, and has been revised on a number of occasions since. It was split into two parts in 1959, Part 1 dealing with the basis of tables and conversion factors, and Part 2, which first appeared in 1962, giving detailed conversion tables for the more frequently used conversions. In 1967 a Supplement (PD 6203) was issued to Part 2, giving additional detailed tables for SI conversions. BS 350-2 was withdrawn in 1981 since many of the tables included in it had become inconsistent with the International System of Units (SI) and the increasing use of pocket calculators was considered to have made such tables, which often required interpolation, obsolete. PD 6203 was withdrawn in 1998.

This revision provides a comprehensive list of conversion factors and notes on their use. The units in about fifty quantities of measurement are given, together with such definitions and information on the derivation of conversion factors as are considered necessary for the purpose. However, as it is now the only part of the standard, it has been returned to its original numbering, BS 350.

In this revision, while interconversion factors between all the important units treated are given, the standpoint from which the various units and conversion factors are discussed is the SI. Very few imperial units remain in official use in the UK. Furthermore, the SI, under the custody of the General Conference of Weights and Measures (CGPM), forms the precise and natural basis for conversion information on units, and offers firm prospects of an international harmonization in unit practice, after which conversion factors will no longer be required. However, it is recognized that many older documents refer to units which are not recommended now, so conversion factors have been retained for many of these.

The standardization function of this standard lies in the provision of conversion factors reliable to a stated accuracy. Other important information is included to help the user to make conversions. BS 350 does not purport to define quantities or units, or to standardize the letter symbols or abbreviations used for units. These matters are dealt with elsewhere, but their mention is necessary here and has been updated with the latest international and national decisions.

Where conversion factors are given in bold type it is to show that they are exact; in general, factors have been rounded to include six significant figures, thus permitting accuracies satisfactory for most practical purposes. The computation of each factor has as far as possible been made from first principles, using eight or more significant figures to minimize the possibilities of errors in rounding. Six-figure factors are unnecessarily precise for many practical purposes, and may be rounded to fewer significant figures as appropriate. The Department of Trade and Industry has asked that users of these conversion factors be reminded that conversions for trade purposes have to be based on the statutory definitions of units in the Weights and Measures Act, 1985 [1].

Submission of additional units for inclusion in a future edition of the standard will be welcomed. These should be sent to the secretary of Technical Committee SS/7 at BSI, 389 Chiswick High Road, London W4 4AL.

This publication does not purport to include all the necessary provisions of a contract. Users are responsible for its correct application.

Compliance with a British Standard does not of itself confer immunity from legal obligations.

In particular, attention is drawn to the Weights and Measures Act 1985 [1] and associated Orders and Amendments.

Summary of pages

This document comprises a front cover, an inside front cover, pages i to iv, pages 1 to 105 and a back cover.

The BSI copyright notice displayed in this document indicates when the document was last issued.

1 Scope

This standard provides conversion factors for units of measurement for a number of quantities, which are, or have been, in general use in engineering, industry and trade. The subjects covered are, broadly, metrology, mechanics and heat; the standard does not deal with purely electrical units.

NOTE 1 SI units are normally recommended. However, in certain circumstances use of particular units is required by legislation.

NOTE 2 Electrical and light units are given in BS 5775-5 and in BS 5775-6, respectively.

NOTE 3 A commentary on imperial and metric systems of measurement and units is given in Annex A.

2 Number

2.1 The following prefixes, with meaning, name and symbol as shown in Table 1, are used to denote decimal multiples or submultiples of (metric) units. These prefixes developed in conjunction with the metric system, and are authorized as "SI prefixes".

Table 1 — Prefixes denoting decimal multiples or submultiples

Meaning	Name	Symbol
<i>To indicate multiples</i>		
$\times 10^{24}$	yotta	Y
$\times 10^{21}$	zetta	Z
$\times 10^{18}$	exa	E
$\times 10^{15}$	peta	P
$\times 10^{12}$	tera	T
$\times 10^9$	giga	G
$\times 10^6$	mega	M
$\times 10^3$	kilo	k
$\times 10^2$	hecto	h
$\times 10^1$	deca	da
<i>To indicate submultiples</i>		
$\times 10^{-1}$	deci	d
$\times 10^{-2}$	centi	c
$\times 10^{-3}$	milli	m
$\times 10^{-6}$	micro	μ
$\times 10^{-9}$	nano	n
$\times 10^{-12}$	pico	p
$\times 10^{-15}$	femto	f
$\times 10^{-18}$	atto	a
$\times 10^{-21}$	zepto	z
$\times 10^{-24}$	yocto	y

2.2 The scientific convention is now to use the system shown in Table 2.

Table 2 — Meaning of million, billion, trillion, etc.

Term	Meaning	Corresponding decimal factor
million	thousand × thousand	10 ⁶
billion	thousand × million	10 ⁹
trillion	million × million	10 ¹²
quadrillion	million × billion	10 ¹⁵

2.3 The former convention in the UK was that one billion was 10¹², one trillion 10¹⁸, one quadrillion 10²⁴ etc. Some people still use this convention. In view of the differences between the former and the current practice, ambiguities can easily arise with the words “billion”, “trillion” and “quadrillion”, etc., therefore their use should be avoided.

3 Length

3.1 The SI unit of length is the metre (symbol m). It is one of the base units of the SI and is the length of the path travelled by light in vacuum during a time interval of **1/299 792 458** of a second.

3.2 Multiples and submultiples of the metre are formed by using any of the SI prefixes given in 2.1; kilometre (km), decimetre (dm), centimetre (cm), millimetre (mm) and micrometre (µm) are common examples. An alternative term for the micrometre, abrogated by the CGPM but still in common use, is “micron”. The symbol µ, associated in the past with the micron, is incorrect; µm should be used.

3.3 Some units of length having associations with the metric system but not forming part of the SI are:

		Refer to note
1 ångström (Å)	= 10 ⁻¹⁰ m	—
1 fermi (f, fm)	= 10 ⁻¹⁵ m	—
1 nautical mile (international) (n mile)	= 1 852 m	—
1 astronomical unit (AU)	= 1.496 00 × 10 ¹¹ m	1
1 parsec (pc)	= 3.085 68 × 10 ¹⁶ m	2
	= 3.262 l.y.	
1 light year (l.y.)	= 9.460 528 × 10 ¹⁵ m	3

3.4 The definitive UK (or imperial) and US unit of length is the yard, legally defined (since 1959 in the USA and since 1963 in the UK) as follows:

$$1 \text{ yard} = \mathbf{0.914\ 4\ m}$$

3.5 The connection between multiples and submultiples of the yard is indicated in the following table of named UK and US units of length.

			Refer to note
	(1 in	= 0.025 4 m)	—
1 foot (ft)	= 12 inches (in)	(1 ft = 0.304 8 m)	4
1 yard (yd)	= 3 feet (ft)	(1 yd = 0.914 4 m)	—
1 chain	= 22 yards (yd)	(1 chain = 20.116 8 m)	5
1 furlong	= 10 chains	(1 furlong = 201.168 m)	—
1 mile	= 8 furlongs	(1 mile = 1 609.344 m)	6

3.6 Some less usual, or more specialized, UK and US named units of length, which are not recommended, are:

				Refer to note
1 micro-inch (μin)	= 10^{-6} in	= 0.025 4 μm	= 25.4 $\times 10^{-9}$ m	—
1 thou	= 10^{-3} in	= 25.4 μm	= 25.4 $\times 10^{-6}$ m	7
1 mil	= 10^{-3} in	= 25.4 μm	= 25.4 $\times 10^{-6}$ m	8
1 point	= $\frac{1}{72}$ in (approx)		= 0.351 mm (approx.)	9
1 iron	= $\frac{1}{48}$ in		= 0.529 167 mm	10
1 line	= $\frac{1}{40}$ in		= 0.635 mm	11
1 line or ligne	= $\frac{1}{12}$ in		= 2.116 67 mm	12
1 em	= $\frac{1}{6}$ in		= 4.233 33 mm	13
1 hand	= 4 in		= 10.16 cm	14
1 link	= $\frac{1}{100}$ chain	= 0.66 ft	= 0.201 168 m	—
1 US survey foot	= $\frac{1}{0.999\ 998}$ ft	= $\frac{12}{39.37}$ m	= 0.304 801 m	—
1 fathom	= 6 ft		= 1.828 8 m	—
1 rod, pole, or perch	= 5½ yd		= 5.029 2 m	15
1 engineer's chain	= 100 ft		= 30.48 m	—
1 cable-length	—		—	16
1 nautical mile (UK)	= 6 080 ft		= 1 853.18 m	17
1 telegraph nautical mile	= 6 087 ft		= 1 855.32 m	—

Notes on Clause 3

NOTE 1 Approximately the mean distance between the Sun and the Earth.

NOTE 2 The distance at which 1 AU subtends an angle of 1 second (1").

NOTE 3 Approximate distance travelled by light in 1 year.

NOTE 4 An exception is the US survey foot, shown in 3.6.

NOTE 5 Commonly called Gunter's chain in the USA.

NOTE 6 Also known as a statute mile. There is no recognized abbreviation for mile and the complete word "mile" is used as the unit symbol.

NOTE 7 Colloquial, for one-thousandth of an inch.

NOTE 8 Colloquial, for one-thousandth of an inch. For other meanings of mil see 4.5, 5.5 and Clause 8, Note 2.

NOTE 9 Printing trade. (Originally defined by 83 picas = 83 \times 12 points = 35 cm.)

NOTE 10 Boot and shoe trade.

NOTE 11 Button trade.

NOTE 12 Watch trade.

NOTE 13 Printing trade.

NOTE 14 Height of horses.

NOTE 15 Obsolescent.

NOTE 16 A nautical term not precisely defined. In its most general concept it is equal to one-tenth of an unspecified nautical mile, but other values have been used, including the "US Navy" value 120 fathoms (720 ft), the "Ordinary" definition 100 fathoms (600 ft), and "Royal Navy" 608 ft. In view of the scope for confusion, the use of this term is deprecated.

NOTE 17 Also known as the "nautical mile (British)" or "Admiralty nautical mile". In 1995 The Units of Measurement Regulations [2] redefined the nautical mile (UK) in metric terms as 1 853 m exactly (approximately 6 079.4 ft). The nautical mile (UK) is now obsolete as the international nautical mile has been adopted in the UK.

For conversion factors for a number of units of length see Table 3.

Table 3 — Length

Exact values are printed in bold type

	metre m	inch ^a in	foot ft	yard yd	chain	furlong	mile	fathom	nautical mile (UK) ^b	nautical mile (international) n mile
1 metre m	= 1	39.370 1	3.280 84	1.093 61	0.049 709 7	$4.970\ 97 \times 10^{-3}$	$6.213\ 71 \times 10^{-4}$	0.546 807	$5.396\ 12 \times 10^{-4}$	$5.399\ 57 \times 10^{-4}$
1 inch in	= 0.025 4	1	0.083 333 3	0.027 777 8	$1.262\ 63 \times 10^{-3}$	$1.262\ 63 \times 10^{-4}$	$1.578\ 28 \times 10^{-5}$	$1.388\ 89 \times 10^{-2}$	$1.370\ 61 \times 10^{-5}$	$1.371\ 49 \times 10^{-5}$
1 foot ft	= 0.304 8	12	1	0.333 333	0.015 151 5	$1.515\ 15 \times 10^{-3}$	$1.893\ 94 \times 10^{-4}$	0.166 667	$1.644\ 74 \times 10^{-4}$	$1.645\ 79 \times 10^{-4}$
1 yard yd	= 0.914 4	36	3	1	0.045 454 5	$4.545\ 45 \times 10^{-3}$	$5.681\ 82 \times 10^{-4}$	0.5	$4.934\ 21 \times 10^{-4}$	$4.937\ 37 \times 10^{-4}$
1 chain	= 20.116 8	792	66	22	1	0.1	0.012 5	11	$1.085\ 53 \times 10^{-2}$	$1.086\ 22 \times 10^{-2}$
1 furlong	= 201.168	7 920	660	220	10	1	0.125	110	0.108 553	0.108 622
1 mile	= 1 609.344	63 360	5 280	1 760	80	8	1	880	0.868 421	0.868 976
1 fathom	= 1.828 8	72	6	2	$9.090\ 91 \times 10^{-2}$	$9.090\ 91 \times 10^{-3}$	$1.136\ 36 \times 10^{-3}$	1	$9.868\ 42 \times 10^{-4}$	$9.874\ 73 \times 10^{-4}$
1 nautical mile (UK) ^b	= 1 853.18	72 960	6 080	2 026.67	92.121 2	9.212 12	1.151 52	1 013.33	1	1.000 64
1 nautical mile (international) n mile	= 1 852	72 913.4	6 076.12	2 025.37	92.062 4	9.206 24	1.150 78	1 012.69	0.999 361	1

^a For a detailed table of conversions from inches to millimetres and vice versa, see BS 2856. The conversions to inches are there given to the nearest 10^{-7} in.

^b The nautical mile (UK) is no longer used.

4 Area (length squared)

4.1 The coherent SI unit of area is the square metre (symbol m²), a derived unit.

4.2 Areas are also expressed in terms of the squares of any of the multiples and submultiples of the metre formed by the use of the SI prefixes, e.g. square millimetre (mm²), square centimetre (cm²), square decimetre (dm²), square kilometre (km²).

In accordance with the rule concerning prefixes attached to units raised to a power, the relationship between each of these and the square metre is as follows:

$$1 \text{ mm}^2 = \left(\frac{\text{m}}{1\,000}\right)^2 = 10^{-6} \text{ m}^2$$

$$1 \text{ cm}^2 = \left(\frac{\text{m}}{100}\right)^2 = 10^{-4} \text{ m}^2$$

$$1 \text{ dm}^2 = \left(\frac{\text{m}}{10}\right)^2 = 10^{-2} \text{ m}^2$$

$$1 \text{ km}^2 = (1\,000 \text{ m})^2 = 10^6 \text{ m}^2$$

4.3 A metric unit with a special name is the are (symbol a).

$$1 \text{ a} = 100 \text{ m}^2$$

This, and more especially its multiple the hectare (symbol ha), are used for land measurement of area.

$$1 \text{ ha} = 100 \text{ a} = 10\,000 \text{ m}^2$$

Another specially named metric unit (but not SI) is the barn, used in atomic physics in the measurement of cross sections.

$$1 \text{ barn} = 10^{-28} \text{ m}^2$$

4.4 The connection between various traditional UK and US units of area, and their relationship to the square metre, are as follows:

				Refer to note
1 square foot (ft ²)	= 144 square inches (in ²)	(1 in ²	= 6.451 6 × 10 ⁻⁴ m ²)	—
1 square yard (yd ²)	= 9 square feet	(1 ft ²	= 0.092 903 0 m ²)	—
1 rood	= 1 210 square yards	(1 yd ²	= 0.836 127 m ²)	1
1 acre = 4 roods	= 4 840 square yards	(1 rood	= 1 011.71 m ²)	—
		(1 acre	= 4 046.86 m ²)	—
1 square mile (mile ²)	= 640 acres	(1 mile ²	= 2.589 99 × 10 ⁶ m ²)	—
		(1 mile ²	= 258.999 ha)	—

4.5 A specialized UK and US unit of area (used in connection with sections of wire) is the “circular mil”.

Refer to
note

$$1 \text{ circular mil} = 7.853\,98 \times 10^{-7} \text{ in}^2 = 5.067\,07 \times 10^{-10} \text{ m}^2$$

2

Notes on Clause 4

NOTE 1 The rood is obsolete in the UK and rarely used in the USA.

NOTE 2 The circular mil has an area equal to that of a circle one-thousandth of an inch in diameter. For other meanings of mil see 3.6, 5.5 and Clause 8, Note 2.

For conversion factors for a number of units of area see Table 4 and Table 5.

Table 4 — Area

Exact values are printed in bold type

	square metre m ²	hectare ha	square inch in ²	square foot ft ²	square yard yd ²	rood	acre	square mile mile ²
1 square metre = m ²	1	1×10^{-4}	1 550.00	10.763 9	1.195 99	$9.884 22 \times 10^{-4}$	$2.471 05 \times 10^{-4}$	$3.861 02 \times 10^{-7}$
1 hectare = ha	10 000	1	$1 550.00 \times 10^4$	107 639	11 959.9	9.884 22	2.471 05	$3.861 02 \times 10^{-3}$
1 square inch = in ²	$6.451 6 \times 10^{-4}$	$6.451 6 \times 10^{-8}$	1	$6.944 44 \times 10^{-3}$	$7.716 05 \times 10^{-4}$	$6.376 90 \times 10^{-7}$	$1.594 23 \times 10^{-7}$	$2.490 98 \times 10^{-10}$
1 square foot = ft ²	0.092 903 0	$9.290 30 \times 10^{-6}$	144	1	0.111 111	$9.182 74 \times 10^{-5}$	$2.295 68 \times 10^{-5}$	$3.587 01 \times 10^{-8}$
1 square yard = yd ²	0.836 127	$8.361 27 \times 10^{-5}$	1 296	9	1	$8.264 46 \times 10^{-4}$	$2.066 12 \times 10^{-4}$	$3.228 31 \times 10^{-7}$
1 rood =	1 011.71	0.101 171	1 568 160	10 890	1 210	1	0.25	$3.906 25 \times 10^{-4}$
1 acre =	4 046.86	0.404 686	6 272 640	43 560	4 840	4	1	$1.562 5 \times 10^{-3}$
1 square mile = mile ²	$2.589 99 \times 10^6$	258.999	$4.014 49 \times 10^9$	$2.787 84 \times 10^7$	$3.097 6 \times 10^6$	2 560	640	1

Table 5 — Area of section of wire

Exact values are printed in bold type

		circular mil	square millimetre mm ²	square inch in ²
1 circular mil ^a	=	1	$5.067\ 07 \times 10^{-4}$	$7.853\ 98 \times 10^{-7}$
1 square millimetre mm ²	=	1 973.53	1	$1.550\ 00 \times 10^{-3}$
1 square inch in ²	=	$1.273\ 24 \times 10^6$	645.16	1

^a For definition of circular mil, see Clause 4, Note 2.

5 Volume and capacity (length cubed)

5.1 The coherent SI unit of volume is the cubic metre (symbol m³), a derived unit.

5.2 Volumes are also expressed in terms of the cubes of any of the multiples and submultiples of the metre formed by the use of the SI prefixes; of these the cubic decimetre (dm³), the cubic centimetre (cm³), and the cubic millimetre (mm³) are common examples.

The relationship between each of these and the cubic metre is as follows:

$$1 \text{ dm}^3 = \left(\frac{\text{m}}{10}\right)^3 = 10^{-3} \text{ m}^3$$

$$1 \text{ cm}^3 = \left(\frac{\text{m}}{100}\right)^3 = 10^{-6} \text{ m}^3$$

$$1 \text{ mm}^3 = \left(\frac{\text{m}}{1\ 000}\right)^3 = 10^{-9} \text{ m}^3$$

5.3 In the SI no distinction is drawn between units of volume and units of capacity. However, a metric unit with a special name, used in conjunction with the SI and commonly used for the measurement of liquids and fluids, is the litre (symbol l¹⁾).

$$1 \text{ litre} = 1 \text{ dm}^3 = 10^{-3} \text{ m}^3$$

(This definition has applied in the SI since 1964, but see 5.4.)

The SI prefixes are used with the litre, leading for example to the hectolitre (hl), centilitre (cl), millilitre (ml) and microlitre (μl).

$$1 \text{ hl} = 100 \text{ litre} = 10^{-1} \text{ m}^3$$

$$1 \text{ cl} = \left(\frac{1}{100}\right) \text{ litre} = 10^{-5} \text{ m}^3$$

$$1 \text{ ml} = \left(\frac{1}{1\ 000}\right) \text{ litre} = 10^{-6} \text{ m}^3 = 1 \text{ cm}^3$$

$$1 \text{ μl} = \left(\frac{1}{10^6}\right) \text{ litre} = 10^{-9} \text{ m}^3 = 1 \text{ mm}^3$$

¹⁾ Although in this standard the symbol used for the litre is the lower case letter “l”, it has long been recognized that in some typefaces it was difficult to distinguish between the lower case letter “l” and numeral 1. The 16th General Conference on Weights and Measures (1979) accordingly recognized the use of the upper case letter “L” as an alternative symbol for the litre. In British Standards “l” is the preferred symbol, but “L” is preferred by some other organizations.

5.4 Units of capacity for the measurement of liquids (and sometimes of dry goods also) have been treated as base units at various times in the past, and have been defined independently of length. Thus in the metric system from 1901 to 1964 the litre was defined as the volume occupied by a mass of one kilogram of water under specified conditions (at its temperature of maximum density and under a pressure of one standard atmosphere). Since 1964, however, the litre has been re-defined within the SI as a special name for the volume of one cubic decimetre (which is as it was before 1901). Since 1 November 1976, the 1964 definition has been embodied within the law of the United Kingdom. Because of these changes, where a very high degree of precision is called for, it is necessary to establish which definition of the litre is intended. In the tables which follow in this standard a litre as defined according to the 1901 definition is described as the “litre (1901)”, and the litre as it is now defined according to the SI is described simply as the “litre”.

$$1 \text{ litre (1901)} = 1.000\,028 \text{ litre}$$

5.5 In the French timber trade the volume of one cubic metre goes under the obsolescent name “stère” (symbol st). Similarly, in the timber trade in Germany the cubic metre has been described as the “Festmeter” (Fm) or “Raummeter” (Rm), and these two special names are also obsolescent. Another obsolete metric-based volume unit is the “mil”, once used in the UK in pharmaceutical work, particularly for prescriptions, to denote a millilitre. For other meanings of “mil” see **3.6**, **4.5**, and Clause **8**, Note **2**.

5.6 The connection between the traditional UK and US units of volume and their relationship to the cubic metre are as follows:

Symbol	Unit		Metric equivalent
yd ³	1 cubic yard	= 27 cubic feet	= 0.764 555 m ³
ft ³	1 cubic foot	= 1 728 cubic inches	= 0.028 316 8 m ³
in ³	1 cubic inch		= 1.638 71 × 10 ⁻⁵ m ³

5.7 As with the litre in the metric system, it is customary to regard certain UK and US volumetric units as units of capacity. These include the UK gallon and its multiples and submultiples, and the US gallon and US bushel, with their multiples and submultiples. The UK and US units of capacity differ markedly from each other²⁾ and it is therefore important to avoid confusion in their use. The prefixes UK and US are used for purpose of their identification in this standard but the qualifications UK or US are frequently omitted in practice. Care is particularly necessary with conversions of the gallon in order to identify which gallon is concerned.

5.8 *UK-units of capacity.* These are all based on the UK gallon (UKgal), defined in Schedule 1 of the Weights and Measures Act, 1985 [1], as 4.546 09 cubic decimetres.

Key conversion factors are:

$$\begin{aligned}
 1 \text{ UKgal} &= 4.546\,09 \text{ dm}^3 \text{ (Weights and Measures Act, 1985 [1])} \\
 &= 4.546\,09 \text{ litre} \\
 &= 4.545\,96 \text{ litre (1901)}
 \end{aligned}$$

²⁾ For a direct comparison of UK and US units of capacity see Table 9.

The connection between the UK gallon and its various multiples and submultiples is shown in the following list.

Symbol (if any)	Unit		Metric equivalent
UKmin	1 minim		= 0.059 193 9 cm ³
UK fl dr	1 fluid drachm	= 60 minim	= 3.551 63 cm ³
UK fl oz	1 fluid ounce	= 8 fluid drachms	= 28.413 1 cm ³
—	1 gill	= 5 fluid ounces	= 0.142 065 dm ³
UKpt	1 pint	= 4 gills (= 20 fluid ounces)	= 0.568 261 dm ³
UKqt	1 quart	= 2 pints	= 1.136 52 dm ³
UKgal	1 gallon	= 4 quarts (= 160 fluid ounces)	= 4.546 09 dm ³
—	1 peck	= 2 gallons	= 9.092 18 dm ³
—	1 bushel	= 4 pecks	= 36.368 7 dm ³

NOTE The minim, fluid drachm, peck and bushel were deleted from Schedule 1 to the UK Weights and Measures Act, 1963, and it is now illegal to use these units for trade purposes.

The following are, or have been, used in the brewing industry:

1 hogshead	= 54 gallons
1 barrel	= 36 gallons
1 kilderkin	= 18 gallons
1 firkin	= 9 gallons
1 pin	= 4.5 gallons

5.9 US units of capacity. The US units of capacity are defined in terms of a specified number of cubic inches. The US gallon is equal in volume to 231 cubic inches and is used for the measurement of liquids only. The US bushel is equal in volume to 2 150.42 cubic inches and is used for the measurement of dry commodities only.

5.10 US units of capacity (liquid measure only). The connection between the US gallon and its various multiples and submultiples is shown in the following list.

Symbol (if any)	Unit		Metric equivalent
—	1 US minim		= 0.061 611 5 cm ³
fl dr	1 US fluid dram ³⁾	= 60 minims	= 3.696 69 cm ³
US fl oz	1 US fluid ounce ⁴⁾	= 8 fluid drams	= 29.573 5 cm ³
gi	1 US gill	= 4 fluid ounces	= 0.118 294 dm ³
liq pt	1 US liquid pint	= 4 gills (= 16 fluid ounces)	= 0.473 176 dm ³
liq qt	1 US liquid quart	= 2 liquid pints	= 0.946 353 dm ³
USgal	1 US gallon	= 4 liquid quarts (= 128 fluid ounces)	= 3.785 41 dm ³
bbl	1 US barrel (for petroleum)	= 42 gallons	= 158.987 dm ³

³⁾ Sometimes also known as the liquid dram (liq dr) in the USA.

⁴⁾ Sometimes also known as the liquid ounce (liq oz) in the USA.

5.11 US units of capacity (dry measure only). The connection between the US bushel and its various multiples and submultiples is shown in the following list.

Symbol (if any)	Unit		Metric equivalent
—	1 US dry pint	=	0.550 610 dm ³
dry qt	1 US dry quart	= 2 dry pints	= 1.101 22 dm ³
pk	1 US peck	= 8 dry quarts	= 8.809 76 dm ³
bu	1 US bushel	= 4 pecks	= 35.239 1 dm ³
bb1 (dry)	1 US dry barrel	= 7 056 cubic inches	= 115.627 dm ³

Notes on Clause 5

NOTE 1 In the UK different values are used for the barrel for different purposes (e.g. the wine barrel is nominally 31½ UKgal and the beer barrel nominally 36 UKgal).

NOTE 2 The barrel (bb1) referred to in the list of US capacity units for dry measure only is the standard barrel in the US for fruits, vegetables and dried commodities, with the exception of cranberries. Cranberries are sold in the US by reference to a standard cranberry barrel containing 5 826 cubic inches.

NOTE 3 There are other bushels having different capacities from those mentioned in 5.8 and 5.11.

NOTE 4 Other specialized units of volume used in the UK timber trade are:

1 board foot	=	144 in ³	(= 2.359 74 dm ³)
1 cord	=	128 ft ³	(= 3.624 56 m ³)
1 standard	=	165 ft ³	(= 4.672 28 m ³)

This last is sometimes known as the “Petrograd standard”.

NOTE 5 The cran, once used in the UK fishing industry, is equal to 37½ UK gallons.

For conversion factors for a number of units of volume and units of capacity see Table 6, Table 7, Table 8 and Table 9.

6 Modulus of section, first moment of area

6.1 These quantities have the same dimensions as volume; the coherent SI unit is therefore the metre cubed (m³).

6.2 They may also be expressed in terms of the cube of any suitable submultiple of the metre; the centimetre cubed (cm³) and millimetre cubed (mm³) are commonly used.

6.3 The imperial units are the foot cubed (ft³) and the inch cubed (in³).

The relationship between the above-mentioned units can be seen in or inferred from Table 6.

7 Second moment of area, or geometrical moment of inertia

The coherent SI unit for this quantity is the metre to the fourth (m⁴). Other commonly used units are the centimetre to the fourth (cm⁴) and millimetre to the fourth (mm⁴). The imperial units are the foot to the fourth (ft⁴) and the inch to the fourth (in⁴). See Table 10 for conversion factors for these units.

Table 6 — Volume and capacity

Exact values are printed in bold type

	cubic metre m ³	cubic decimetre dm ³ litre ^a l	litre (1901) ^a	cubic inch in ³	cubic foot ft ³	cubic yard yd ³	UK bushel	US dry pint	US bushel
1 cubic metre m ³	1	1 000	999.972	61 023.7	35.314 7	1.307 95	27.496 1	1 816.17	28.377 6
1 cubic decimetre dm ³ 1 litre l ^a	0.001	1	0.999 972	61.023 7	0.035 314 7	1.307 95 × 10 ⁻³	0.027 496 1	1.816 17	0.028 377 6
1 litre (1901) ^a	1.000 028 × 10 ⁻³	1.000 028	1	61.025 5	0.035 315 7	1.307 99 × 10 ⁻³	0.027 496 9	1.816 22	0.028 378 4
1 cubic inch in ³	1.638 71 × 10 ⁻⁵	1.638 71 × 10 ⁻²	0.016 386 6	1	5.787 04 × 10 ⁻⁴	2.143 35 × 10 ⁻⁵	4.505 81 × 10 ⁻⁴	0.029 761 6	4.650 25 × 10 ⁻⁴
1 cubic foot ft ³	0.028 316 8	28.316 8	28.316 1	1 728	1	0.037 037 0	0.778 604	51.428 1	0.803 564
1 cubic yard yd ³	0.764 555	764.555	764.533	46 656	27	1	21.022 3	1 388.56	21.696 2
1 UK bushel	0.036 368 7	36.368 7	36.367 7	2 219.36	1.284 35	0.047 568 5	1	66.051 7	1.032 06
1 US dry pint	5.506 10 × 10 ⁻⁴	0.550 610	0.550 595	33.600 3	0.019 444 6	7.201 71 × 10 ⁻⁴	0.015 139 7	1	0.015 625
1 US bushel	0.035 239 1	35.239 1	35.238 1	2 150.42	1.244 46	0.046 091 0	0.968 939	64	1

^a For explanation of the litre and the litre (1901) see 5.3 and 5.4.

Table 7 — Volume and capacity (continued)

Exact values are printed in bold type

	cubic metre	cubic decimetre	litre (1901) ^a	cubic inch	cubic foot	UK pint ^b	UK gallon ^c	US liquid pint	US gallon
	m ³	dm ³ litre ^a l		in ³	ft ³	UKpt	UKgal	US liq pt	USgal
1 cubic metre m ³	1	1 000	999.972	61 023.7	35.314 7	1 759.75	219.969	2 113.38	264.172
1 cubic decimetre dm ³ 1 litre l ^a	0.001	1	0.999 972	61.023 7	0.035 314 7	1.759 75	0.219 969	2.113 38	0.264 172
1 litre (1901) ^a	$1.000\ 028 \times 10^{-3}$	1.000 028	1	61.025 5	0.035 315 7	1.759 80	0.219 975	2.113 44	0.264 179
1 cubic inch in ³	$1.638\ 71 \times 10^{-5}$	0.016 387 1	0.016 386 6	1	$5.787\ 04 \times 10^{-4}$	0.028 837 2	$3.604\ 65 \times 10^{-3}$	0.034 632 0	$4.329\ 00 \times 10^{-3}$
1 cubic foot ft ³	0.028 316 8	28.316 8	28.316 1	1 728	1	49.830 7	6.228 83	59.844 2	7.480 52
1 UK pint ^b UKpt	$0.568\ 261 \times 10^{-3}$	0.568 261	0.568 246	34.677 4	0.020 068 0	1	0.125	1.200 95	0.150 119
1 UK gallon ^c UKgal	$4.546\ 09 \times 10^{-3}$	4.546 09	4.545 96	277.420	0.160 544	8	1	9.607 60	1.200 95
1 US liquid pint US liq pt	$4.731\ 76 \times 10^{-4}$	0.473 176	0.473 163	28.875	0.016 710 1	0.832 674	0.104 084	1	0.125
1 US gallon USgal	$3.785\ 41 \times 10^{-3}$	3.785 41	3.785 31	231	0.133 681	6.661 39	0.832 674	8	1
<p>^a For explanation of the litre and the litre (1901) see 5.3 and 5.4.</p> <p>^b Also called an “imperial pint”.</p> <p>^c Also called an “imperial gallon”.</p>									

Table 8 — Volume and capacity (continued)

Exact values are printed in bold type

	cubic centimetre cm ³ millilitre ml	millilitre (1901)	cubic inch in ³	UK minim UKmin	UK fluid drachm UK fl dr	UK fluid ounce UK fl oz	US fluid ounce US fl oz
1 cubic centimetre cm ³ = 1 millilitre ml	1	0.999 972	0.061 023 7	16.893 6	0.281 561	0.035 195 1	0.033 814 0
1 millilitre (1901) ^a	1.000 028	1	0.061 025 5	16.894 1	0.281 568	0.035 196 1	0.033 815 0
1 cubic inch in ³	16.387 1	16.386 6	1	276.837	4.613 95	0.576 744	0.554 113
1 UK minim UKmin	0.059 193 9	0.059 192 2	$3.612\ 23 \times 10^{-3}$	1	0.016 666 7	$2.083\ 33 \times 10^{-3}$	$2.001\ 58 \times 10^{-3}$
1 UK fluid drachm UK fl dr	3.551 63	3.551 53	0.216 734	60	1	0.125	0.120 095
1 UK fluid ounce UK fl oz	28.413 1	28.412 3	1.733 87	480	8	1	0.960 760
1 US fluid ounce US fl oz	29.573 5	29.572 7	1.804 69	499.604	8.326 74	1.040 84	1
NOTE 1 mm ³ = 10 ⁻³ cm ³ = 1 µl (see also 5.2).							
^a For explanation of the litre and the litre (1901) see 5.3 and 5.4.							

Table 9 — Relationship between UK (imperial) and US units of capacity

1 UK minim	=	0.960 760 US minim
1 UK fluid drachm	=	0.960 760 US fluid ^a dram
1 UK fluid ounce	=	0.960 760 US fluid ^b ounce
1 UK gill	=	1.200 95 US gill
1 UK pint	=	1.200 95 US liquid pint
1 UK quart	=	1.200 95 US liquid quart
1 UK gallon	=	1.200 95 US gallon
1 UK pint	=	1.032 06 US dry pint
1 UK quart	=	1.032 06 US dry quart
1 UK peck	=	1.032 06 US peck
1 UK bushel	=	1.032 06 US bushel
1 US minim	=	1.040 84 UK minim
1 US fluid ^a dram	=	1.040 84 UK fluid drachm
1 US fluid ^b ounce	=	1.040 84 UK fluid ounce
1 US gill	=	0.832 674 UK gill
1 US liquid pint	=	0.832 674 UK pint
1 US liquid quart	=	0.832 674 UK quart
1 US gallon	=	0.832 674 UK gallon
1 US dry pint	=	0.968 939 UK pint
1 US dry quart	=	0.968 939 UK quart
1 US peck	=	0.968 939 UK peck
1 US bushel	=	0.968 939 UK bushel
^a Sometimes also known as the liquid dram in the USA.		
^b Sometimes also known as the liquid ounce in the USA.		

Table 10 — Second moment of area

Exact values are printed in bold type

	m ⁴	cm ⁴	ft ⁴	in ⁴
1 m ⁴ =	1	1 × 10 ⁸	115.862	2 402 510
1 cm ⁴ =	1 × 10 ⁻⁸	1	1.158 62 × 10 ⁻⁶	0.024 025 1
1 ft ⁴ =	0.863 097 × 10 ⁻²	863 097	1	20 736
1 in ⁴ =	41.623 1 × 10 ⁻⁸	41.623 1	4.822 53 × 10 ⁻⁵	1
NOTE 1 mm ⁴ = 10 ⁻⁴ cm ⁴ = 10 ⁻¹² m ⁴ .				

8 Plane angle

8.1 The coherent SI unit of plane angle is the radian (symbol rad), a coherent derived⁵⁾ unit. It is the angle between two radii of a circle, which cut an arc on the circumference equal in length to the radius.

Thus a complete circle subtends an angle of 2π rad at its centre, and a right angle (\perp)

$$\text{equals } \frac{2\pi}{4} \text{ rad} = \frac{\pi}{2} \text{ rad.}$$

8.2 Traditional angular units which are of such practical importance that they have been retained for general use in conjunction with the SI are the degree ($^\circ$), minute ($'$) and second ($''$) of arc.

The full circle subtends an angle of 360 degrees (360°) at its centre, and thus the right angle (\perp) = 90 degrees (90°).

$$1 \text{ degree } (1^\circ) = 60 \text{ minutes } (60') = \frac{\pi}{180} \text{ rad}$$

$$1 \text{ minute } (1') = 60 \text{ seconds } (60'') = \frac{\pi}{60 \times 180} \text{ rad}$$

$$1 \text{ second } (1'') = \frac{\pi}{3\,600 \times 180} \text{ rad}$$

It is often convenient to express sub-divisions of the degree in decimal form, rather than to use minutes and seconds. In navigation, it is now usual to quote degrees, minutes and decimals of minutes.

8.3 A unit of plane angle used in some mainland European countries is the grade (g) or, as it is called in Germany, the gon. This is a one-hundredth of a right angle.

$$1^g \text{ (or 1 gon)} = 0.9^\circ = \frac{\pi}{200} \text{ rad}$$

Notes on Clause 8

NOTE 1 Note the possibility of confusion between the hundredth part of a grade in angular measure and the term "Centigrade" (correctly called Celsius) in connection with temperature (see **39.2**).

NOTE 2 The unit "mil" is sometimes used in connection with angular measure. For some purposes the angular mil is taken to be one thousandth of a radian (10^{-3} rad), which is equivalent to $3' 26.25''$. There is, however, another concept in which an angular mil is equal to $360/6\,400$ degrees i.e. $3' 22.5''$. For other meanings of "mil" see **3.6**, **4.5**, and **5.5**.

NOTE 3 In English there is no commonly used expression for the "full angle" subtended by a circle. In German the term "Vollwinkel" is used.

For interconversion factors for the units mentioned in 8.1, 8.2 and 8.3 see Table 11.

⁵⁾ In October 1980 the International Committee of Weights and Measures decided to interpret the class of supplementary (now "coherent derived") units in the International System as a class of dimensionless derived units for which the General Conference of Weights and Measures leaves open the possibility of using these or not in expressions of derived units of the International System.

Table 11 — Plane angle

Exact values are printed in bold type

	radian rad	right angle └	degree °	minute '	second "	grade (or gon) g gon
1 radian rad =	1	0.636 620	57.295 8	3 437.75	206 265	63.662 0
1 right angle =	1.570 80	1	90	5 400	324 000	100
1 degree ° =	0.017 453 3	0.011 111 1	1	60	3 600	1.111 11
1 minute ' =	$2.908 88 \times 10^{-4}$	$1.851 85 \times 10^{-4}$	0.016 666 7	1	60	$1.851 85 \times 10^{-2}$
1 second " =	$4.848 14 \times 10^{-6}$	$3.086 42 \times 10^{-6}$	$2.777 78 \times 10^{-4}$	0.016 666 7	1	$3.086 42 \times 10^{-4}$
1 grade (or gon) g gon =	0.015 708 0	0.01	0.9	54	3 240	1

9 Solid angle

The coherent SI unit of solid angle, the only unit in common use for solid angle, is the steradian (symbol sr), a coherent derived⁶⁾ unit. It is the solid angle which, having its vertex at the centre of a sphere, cuts off an area of the surface of the sphere equal to that of a square with sides of length equal to the radius of the sphere.

A complete sphere subtends a solid angle of 4π sr at its centre.

10 Time

10.1 The SI unit of time is the second (symbol s), a base unit. It is now defined as the duration of 9 192 631 770 periods of the radiation corresponding to the transition between the two hyperfine levels of the ground state of the caesium 133 atom, as stated in *The International System of Units (SI)* 1998 [3].

10.2 Prior to 1967 the second was defined as a specified fraction of the time taken by the Earth to complete a particular orbit of the Sun. (This second, the “ephemeris second”, is retained for use as a special unit in astronomy, and is as nearly equal to the SI unit as the highest precision of measurement could permit in 1967.)

10.3 Other units of time of such practical importance that they are retained for general use in conjunction with the SI are:

minute (min)	1 min	=	60 s	
hour (h)	1 h	=	60 min	= 3 600 s
day (d)	1 d	=	24 h	= 86 400 s

10.4 Longer durations of time are conveniently expressed in terms of the week, month or year, but the last two of these cannot in general be explicitly related to the second (of time).

1 week	=	7 d = 604 800 s
1 month	=	28, 29, 30 or 31 days (according to calendar)
1 year ⁷⁾	=	12 months = 365 or 366 days (according to calendar) = 8 760 h or 8 784 h (according to calendar).

⁶⁾ In October 1980 the International Committee of Weights and Measures decided to interpret the class of supplementary (now “coherent derived”) units in the International System as a class of dimensionless derived units for which the General Conference of Weights and Measures leaves open the possibility of using these or not in expressions of derived units of the International System.

⁷⁾ The year referred to here is the “calendar year”. Calendar adjustments are based on the “tropical year”, the time interval between two consecutive passages (in the same direction) of the Sun through the Earth’s equatorial plane. In 1900 the duration of the “tropical year” was 365.242 198 78 d and it is decreasing at the rate of 6.14×10^{-6} days per century.

Notes on Clause 10

NOTE 1 The only time unit commonly used in conjunction with the SI prefixes is the second, e.g. the submultiples millisecond (ms), microsecond (μs) and nanosecond (ns), which are in wide technological use.

NOTE 2 The symbol “a” is used for year.

NOTE 3 The scale of International Atomic Time (TAI), based directly on the atomic radiation defining the second, is maintained by the Bureau International de l’Heure (BIH) in Paris. Legal time in the UK, as in most countries, is based on a related scale, that of Co-ordinated Universal Time (UTC), broadcast by an international network of radio stations. UTC is defined in such a manner that it differs from TAI by a whole number of seconds. The difference UTC – TAI was set equal to -10 s on 1 January 1972, the date of application of the reformulation of UTC (which previously involved a frequency offset). On 1 January 1999 the difference had risen to -31 s .

This difference can be changed in steps of 1 s, by the use of a positive or negative leap second at the end of a month of UTC, either at the end of December or of June, to keep UTC in agreement with the time defined by the rotation of Earth with an approximation better than 0.9 s. The decision on whether to add or remove a second is made by the International Earth Rotation Service (IERS), and notices are distributed well in advance. So far, all leap seconds have been positive. The legal times of most countries are offset from UTC by a whole number of hours (because of time zones and “daylight saving” arrangements). Note that since 1995, all member states of the European Union (EU) have used a common date and time for the beginning and end of “summer time”.

11 Linear velocity (speed) (length/time)

11.1 The coherent SI unit of linear velocity is the metre per second (symbol m/s), a coherent derived unit.

11.2 Multiples and submultiples of the metre per second are formed by using any of the SI prefixes in conjunction with the metre.

11.3 A metric unit often used for speed is the kilometre per hour (km/h).

$$1\text{ km/h} = 0.277\ 778\text{ m/s}$$

11.4 Various speed units used in the imperial system are:

foot per second	1 ft/s	=	0.304 8 m/s
inch per second	1 in/s	=	0.025 4 m/s
foot per minute	1 ft/min	=	0.005 08 m/s
mile per hour ⁸⁾	1 mile/h	=	0.447 04 m/s

11.5 The knot, one nautical mile per hour, is a unit used for speed in nautical and aeronautical contexts. The international knot (kn) is metric-based, being equal to one international nautical mile per hour.

$$1\text{ kn} = 1\ 852\text{ m/h} = 0.514\ 444\text{ m/s}$$

The UK knot is imperial-based and obsolescent, being equal to one UK nautical mile per hour.

$$1\text{ UK knot} = 6\ 080\text{ ft/h} = 0.514\ 773\text{ m/s}$$

For interconversion factors for the above units see Table 12.

⁸⁾ Traditionally indicated by the abbreviation m.p.h.

12 Angular velocity⁹⁾ (angle/time)

12.1 The coherent SI unit of angular velocity is the radian per second (rad/s), a derived unit.

12.2 Other units used are:

radian per minute	(rad/min)
revolution per minute	(rev/min) or (r/min)
revolution per second	(rev/s) or (r/s)
degree per minute	(°/min)
degree per second	(°/s)

For interconversion factors for the above units see Table 13.

⁹⁾ The terms “rotational velocity”, “rotational speed” and “speed of rotation” are commonly used as alternative terms for angular velocity, but are also often thought of as a frequency, particularly when being expressed in revolutions per minute, or per second. When frequency is meant, the revolution should not be identified with angle as it is so identified in Table 13 (1 revolution = 2π radians = 360°), but should be thought of as a number, and a clearer term for expressing this concept is “rotational frequency”. (See also Clause 13, Frequency.)

Table 12 — Linear velocity

Exact values are printed in bold type

	metre per second m/s	kilometre per hour km/h	foot per second ft/s	foot per minute ft/min	inch per second in/s	mile per hour mile/h	knot (international) kn	UK knot
1 metre per second m/s	1	3.6	3.280 84	196.850	39.370 1	2.236 94	1.943 84	1.942 60
1 kilometre per hour km/h	0.277 778	1	0.911 344	54.680 7	10.936 1	0.621 371	0.539 957	0.539 612
1 foot per second ft/s	0.304 8	1.097 28	1	60	12	0.681 818	0.592 484	0.592 105
1 foot per minute ft/min	0.005 08	0.018 288	0.016 666 7	1	0.2	0.011 363 6	$9.874\ 73 \times 10^{-3}$	$9.868\ 42 \times 10^{-3}$
1 inch per second in/s	0.025 4	0.091 44	0.083 333 3	5	1	0.056 818 2	$4.937\ 37 \times 10^{-2}$	$4.934\ 21 \times 10^{-2}$
1 mile per hour mile/h	0.447 04	1.609 344	1.466 67	88	17.6	1	0.868 976	0.868 421
1 knot (international) kn	0.514 444	1.852	1.687 81	101.269	20.253 7	1.150 78	1	0.999 361
1 UK knot	0.514 773	1.853 18	1.688 89	101.333	20.266 7	1.151 52	1.000 64	1

Table 13 — Angular velocity and velocity of rotation

Exact values are printed in bold type

	radian per second rad/s	radian per minute rad/min	revolution per second rev/s	revolution per minute rev/min	degree per second °/s	degree per minute °/min
1 radian per second = rad/s	1	60	0.159 155	9.549 30	57.295 8	3 437.75
1 radian per minute = rad/min	0.016 666 7	1	0.002 652 58	0.159 155	0.954 930	57.295 8
1 revolution per second = rev/s	6.283 19	376.991	1	60	360	21 600
1 revolution per minute = rev/min	0.104 720	6.283 19	0.016 666 7	1	6	360
1 degree per second = °/s	0.017 453 3	1.047 20	0.002 777 78	0.166 667	1	60
1 degree per minute = °/min	$2.908 88 \times 10^{-4}$	0.017 453 3	$4.629 63 \times 10^{-5}$	$2.777 78 \times 10^{-3}$	0.016 666 7	1

13 Frequency (number/time)

13.1 The coherent SI unit of frequency (of a wave or periodic phenomenon) is the hertz (symbol Hz), a derived unit with a special name. Formerly, in the UK, the hertz was called the cycle per second (c/s). Expressed in terms of base units of the SI both the hertz and the cycle per second are the inverse second,

$$\text{i.e. } \frac{1}{\text{s}} \quad (\text{or } \text{s}^{-1}).$$

13.2 The coherent SI unit of rotational frequency (e.g. a frequency associated with the mechanical rotation of a shaft) is also the inverse second,

$$\text{i.e. } \frac{1}{\text{s}} \quad (\text{or } \text{s}^{-1}).$$

It is commonly known as the revolution per second (rev/s or r/s).

13.3 Another very commonly used unit of rotational frequency is the revolution per minute (rev/min or r/min, but traditionally indicated by the abbreviation r.p.m.).

$$\begin{aligned} 1 \text{ rev/min} &= \frac{1}{60} \text{ rev/s} \\ &= (\text{in SI terms}) \quad \frac{1}{60 \text{ s}} \quad \text{or} \quad \frac{1}{60} \text{ s}^{-1} \end{aligned}$$

13.4 Corresponding angular velocities are obtainable from Table 13,

using 1 rev/s as corresponding to 1 Hz, $\frac{1}{\text{s}}$ (or 1 s^{-1}).

NOTE See also footnote to Clause 12, Angular velocity.

14 Acceleration (length/time squared)

14.1 The coherent SI unit of acceleration is the metre per second squared (symbol m/s^2), a derived unit.

14.2 The centimetre per second squared (cm/s^2), a submultiple of the above, is also called the galileo or gal (symbol Gal).

$$1 \text{ Gal} = 1 \text{ cm/s}^2 = 10^{-2} \text{ m/s}^2$$

A unit that has been commonly used in geodesy is the milligal (mGal).

$$1 \text{ mGal} = 10^{-3} \text{ Gal} = 10^{-5} \text{ m/s}^2$$

14.3 The unit in the imperial system for acceleration is the foot per second squared (ft/s^2).

$$1 \text{ ft/s}^2 = \mathbf{0.304\ 8} \text{ m/s}^2$$

14.4 The standard acceleration of $\mathbf{9.806\ 65} \text{ m/s}^2$ plays an important part in the definition of certain units in the older technical systems. When the acceleration of free fall has this value, this is the standard acceleration due to gravity, also known as “standard gravity”, the associated symbol for this quantity being g_n .

The acceleration due to gravity is sometimes used as a unit of acceleration, and called “ g ”, particularly in aeronautical engineering and centrifuge technology. For the sake of precision the standard value $\mathbf{9.806\ 65} \text{ m/s}^2$ should be taken for this unit. A close approximation in imperial units is $32.174\ 0 \text{ ft/s}^2$. These are frequently rounded to 9.81 m/s^2 and 32.2 ft/s^2 .

Interconversion factors for the above units can be seen in or inferred from Table 14.

Table 14 — Acceleration

Exact values are printed in bold type

	metre per second squared m/s^2	foot per second squared ft/s^2	standard acceleration due to gravity (standard gravity) g_n
1 metre per second squared m/s^2	1	3.280 84	0.101 972
1 foot per second squared ft/s^2	0.304 8	1	0.031 081 0
standard acceleration due to gravity = (“standard gravity”) g_n	9.806 65	32.174 0	1
NOTE 1 Gal = $1 \text{ cm/s}^2 = 10^{-2} \text{ m/s}^2$ 1 mGal = 10^{-5} m/s^2 (see 14.2).			

15 Mass

15.1 The coherent SI unit of mass is the kilogram (symbol kg), a base unit. It is defined as equal to the mass of the international prototype of the kilogram (which is in the custody of the International Bureau of Weights and Measures at Sèvres near Paris). In view of the variability of the standard kilogram, due to dirt, moisture etc., coupled with more accurate balances, it is planned to replace it with a theoretical, electronic kilogram.

15.2 Because the name of the base unit of mass already contains the SI prefix “kilo”, multiples and submultiples are formed by adding SI prefixes to the word “gram”. Examples are megagram (Mg), gram (g), milligram (mg) and microgram (μg), as follows:

	Refer to note
1 Mg = 1 000 kg	—
1 g = $\frac{1}{1\,000}$ kg	—
1 mg = 10^{-6} kg	—
1 μg = 10^{-9} kg	1

In practice the megagram is usually referred to by the special name “tonne” (symbol t), and is often called the “metric ton” in the UK and in the USA.

15.3 Some other units of mass having associations with the metric system are:

	Refer to note
1 metric carat = 200 milligrams = 2×10^{-4} kg	2
1 quintal (q) = 100 kg	—
1 atomic mass unit (u) = $1.660\,53 \times 10^{-27}$ kg	—

15.4 The primary unit of mass in the imperial system and in the USA is the pound (lb). In the UK Weights and Measures Act 1985 [1], and in similar legislation in the USA, it is defined exactly as follows:

$$1 \text{ lb} = 0.453\,592\,37 \text{ kg}$$

15.5 The connection between multiples and submultiples of the pound is indicated in the following lists of named UK and US units of mass.

Avoirdupois units

a) UK and US units

1 pound	= 16 ounces (oz)	(1 oz = 28.349 5 g)
	= 16×16 drams (dr)	(1 dr = 1.771 85 g)
	= 7 000 grains (gr)	(1 gr = 0.064 798 91 g)

b) UK units only

1 stone	= 14 pounds	(= 6.350 29 kg)
1 quarter (qr)	= 28 pounds	(= 12.700 6 kg)
1 cental (ctl)	= 100 pounds	(= 45.359 2 kg)
1 hundredweight (cwt)	= 112 pounds	(= 50.802 3 kg)
1 ton (ton)	= 2 240 pounds	(= 1 016.05 kg)

c) US units only

1 short hundredweight (sh cwt)	= 100 pounds	(= 45.359 2 kg)
1 short ton (sh ton)	= 2 000 pounds	(= 907.185 kg)

(In the USA the word ton refers to the “short ton” of 2 000 lb unless otherwise specified. The terms “long ton” or “gross ton” are sometimes used, referring to the ton of 2 240 lb. The hundredweight of 112 lb is often called the “long hundredweight”. The use of the “long” units is decreasing in the USA.)

Apothecaries’ units (formerly used in the UK¹⁰) and the USA)

Both British and US Pharmacopoeias use SI units, but the following units may appear in older documents.

1 scruple ¹⁰	=	20 grains	(= 1.295 98 gram)
1 drachm ¹⁰ (in UK)	=	3 scruples	(= 3.887 93 gram)
1 dram (in USA)	=	3 scruples	(= 3.887 93 gram)
1 ounce ¹⁰ = 24 scruples (oz apoth in UK oz ap in USA)	=	480 grains	(= 31.103 5 gram)

Troy units (used in the UK and the USA)

1 ounce troy (oz tr in UK oz t in USA)	=	1 apothecaries’ ounce	=	480 grains	(= 31.103 5 gram)
--	---	-----------------------	---	------------	-------------------

(The apothecaries’ ounce¹⁰) and the ounce troy are identical in mass and differ from the avoirdupois ounce. Unless otherwise qualified, the term ounce and its abbreviation oz signify the avoirdupois ounce. The pound troy has no legal basis in the UK but was legalized in the USA, where it was defined as a mass equal to 5 760 grains.

1 pound troy (USA only) = 12 ounces troy = 5 760 grains (= 0.373 242 kg)

The grain has the same value in the avoirdupois, troy, and apothecaries’ systems, and is abbreviated to gr in the UK.)

15.6 Some more specialized UK and/or USA named units of mass are:

			Refer to note
1 assay ton (UK)	=	32.667 g	3
1 assay ton (US)	=	29.166 g	4
1 slug	=	32.174 0 lb	(= 14.593 9 kg) 5
1 international corn bushel	=	60 lb	(= 27.215 5 kg) 6

For interconversion factors for many of the units of mass mentioned above see Table 15, Table 16 and Table 17.

¹⁰ The apothecaries’ units (scruple, drachm, and apothecaries’ ounce) have been illegal since 1 January 1971 for use in the United Kingdom.

Notes on Clause 15

NOTE 1 The alternative name “gamma” (symbol γ) is sometimes used to indicate a microgram.

NOTE 2 The metric carat has international sanction for use in trade in diamonds, fine pearls, and precious stones. In the UK the legal abbreviation for this unit is CM.

NOTE 3 The number of milligrams in a UK assay ton is equal to the number of ounces troy in a UK ton.

NOTE 4 The number of milligrams in a US assay ton is equal to the number of ounces troy in a US (short) ton.

NOTE 5 The slug is the British technical unit of mass. One pound-force acting on this mass produces an acceleration of 1 foot per second squared.

NOTE 6 Used for the sale of wheat under International Wheat Agreement, 1949. However, this was replaced by the International Grains Agreement in 1995, which uses only metric units.

Table 15 — Mass

Exact values are printed in bold type

		kilogram kg	pound lb	slug
1 kilogram kg	=	1	2.204 62	0.068 521 8
1 pound lb	=	0.453 592 37	1	0.031 081 0
1 slug	=	14.593 9	32.174 0	1

Table 16 — Mass (continued)

Exact values are printed in bold type

	gram g (0.001 kg)	metric carat	grain^a gr	dram (avoirdupois) dr	drachm (apoth.)	ounce (avoirdupois) oz	ounce (troy or apoth.) oz tr or oz apoth
1 gram (0.001 kg) g	= 1	5	15.432 4	0.564 383	0.257 206	0.035 274 0	0.032 150 7
1 metric carat	= 0.2	1	3.086 47	0.112 877	0.051 441 2	$7.054\ 79 \times 10^{-3}$	$6.430\ 15 \times 10^{-3}$
1 grain ^a gr	= 0.064 798 9	0.323 995	1	0.036 571 4	0.016 666 7	$2.285\ 71 \times 10^{-3}$	$2.083\ 33 \times 10^{-3}$
1 dram (avoirdupois) dr	= 1.771 85	8.859 23	27.343 75	1	0.455 729	0.062 5	0.056 966 1
1 drachm (apoth.)	= 3.887 93	19.439 7	60	2.194 29	1	0.137 143	0.125
1 ounce (avoirdupois) oz	= 28.349 5	141.748	437.5	16	7.291 67	1	0.911 458
1 ounce (troy or apoth.) oz tr or oz apoth	= 31.103 5	155.517	480	17.554 3	8	1.097 14	1
^a The grain has the same value in the avoirdupois, troy and apothecaries' systems. The apothecaries' units (drachm and apothecaries' ounce) are no longer legal for use in the United Kingdom.							

Table 17 — Mass (continued)

Exact values are printed in bold type

		tonne (1 000 kg) (megagram Mg)	pound lb	UK hundredweight cwt	short hundredweight ^a sh cwt	UK ton ton	short ton ^a sh ton
1 tonne (1 000 kg) (1 Mg) t	=	1	2 204.62	19.684 1	22.046 2	0.984 207	1.102 31
1 pound lb	=	$4.535\ 92 \times 10^{-4}$	1	$8.928\ 57 \times 10^{-3}$	0.01	$4.464\ 29 \times 10^{-4}$	0.000 5
1 hundredweight cwt	=	0.050 802 3	112	1	1.12	0.05	0.056
1 short hundredweight ^a sh cwt	=	0.045 359 2	100	0.892 857	1	0.044 642 9	0.05
1 UK ton ton	=	1.016 05	2 240	20	22.4	1	1.12
1 short ton ^a sh ton	=	0.907 185	2 000	17.857 1	20	0.892 857	1

^a US units. The short hundredweight (of 100 lb) is, in the UK, sometimes called the “cental” (symbol ctl).

16 Mass per unit length (or lineic mass) (formerly linear density) (mass/length)

16.1 The coherent SI unit of mass per unit length is the kilogram per metre (symbol kg/m), a derived unit. The term “lineic mass” is recommended by the International Organization for Standardization (ISO).

16.2 Two specialized units of linear density used in the textile industry and which have an association with the metric system are:

1 tex	=	1 gram per kilometre	=	10^{-6} kg/m
1 denier	=	1 gram per 9 kilometres	=	$0.111\ 112 \times 10^{-6}$ kg/m

16.3 A selection of imperial units used in industry, often for wires, rods etc. is:

1 pound per inch (lb/in)	(= 17.858 0 kg/m)
1 pound per foot (lb/ft)	(= 1.488 16 kg/m)
1 pound per yard (lb/yd)	(= 0.496 055 kg/m)
1 pound per mile (lb/mile)	(= $2.818\ 49 \times 10^{-4}$ kg/m)
1 UK ton per 1 000 yards (ton/1 000 yd)	(= 1.111 16 kg/m)
1 UK ton per mile (ton/mile)	(= 0.631 342 kg/m)

Interconversion factors for the units in 16.1 and 16.3 are given in Table 18.

For further information on **16.2** reference should be made to BS 947 which gives tables for calculating the tex values of numbers or counts in other systems, including denier.

Table 18 — Mass per unit length (lineic mass)
(applicable to wires, rods, etc.)

Exact values are printed in bold type

	kilogram per metre kg/m	pound per inch lb/in	pound per foot lb/ft	pound per yard lb/yd	pound per mile lb/mile	UK ton per 1 000 yards ton/1 000 yd	UK ton per mile ton/mile
1 kilogram per metre ^a kg/m	1	0.055 997 4	0.671 969	2.015 91	3 548.00	0.899 958	1.583 93
1 pound per inch lb/in	17.858 0	1	12	36	63 360	16.071 4	28.285 7
1 pound per foot lb/ft	1.488 16	0.083 333 3	1	3	5 280	1.339 29	2.357 14
1 pound per yard lb/yd	0.496 055	0.027 777 8	0.333 333	1	1 760	0.446 429	0.785 714
1 pound per mile lb/mile	$2.818\ 49 \times 10^{-4}$	$1.578\ 28 \times 10^{-5}$	$1.893\ 94 \times 10^{-4}$	$5.681\ 82 \times 10^{-4}$	1	$2.536\ 53 \times 10^{-4}$	$4.464\ 29 \times 10^{-4}$
1 UK ton per 1 000 yards ton/1 000 yd	1.111 16	0.062 222 2	0.746 667	2.24	3 942.4	1	1.76
1 UK ton per mile ton/mile	0.631 342	0.035 353 5	0.424 242	1.272 73	2 240	0.568 182	1
^a 1 kilogram per metre (kg/m) = 1 tonne per kilometre (t/km).							

17 Mass per unit area (areic mass) (mass/length squared)

(applicable for example to sheet metal, plating etc., and in agriculture)

17.1 The coherent SI unit is the kilogram per square metre (kg/m^2), a derived unit. ISO recommend the term “areic mass”.

17.2 Other commonly used metric units are:

gram per square metre (g/m^2)	(= 0.001 kg/m^2)
milligram per square centimetre (mg/cm^2)	(= 0.01 kg/m^2)
milligram per square millimetre (mg/mm^2)	(= 1 kg/m^2)
kilogram per hectare (kg/ha)	(= 0.000 1 kg/m^2)

17.3 A selection of imperial units is:

pound per thousand square feet ($\text{lb}/1\ 000\ \text{ft}^2$)	(= 4.882 43 $\times 10^{-3}\ \text{kg/m}^2$)
ounce per square yard (oz/yd^2)	(= 3.390 57 $\times 10^{-2}\ \text{kg/m}^2$)
ounce per square foot (oz/ft^2)	(= 0.305 152 kg/m^2)
pound per acre (lb/acre)	(= 1.120 85 $\times 10^{-4}\ \text{kg/m}^2$)
UK ton per square mile (ton/mile^2)	(= 3.922 98 $\times 10^{-4}\ \text{kg/m}^2$)

For interconversion factors for the above see Table 19.

18 Specific surface, or area per unit mass

(applicable to sheet metal, plating, etc., and in agriculture)

18.1 The coherent SI unit is the square metre per kilogram (m^2/kg), a derived unit.

18.2 Other commonly used metric units are:

square metre per gram (m^2/g)	(= 1 000 m^2/kg)
square centimetre per milligram (cm^2/mg)	(= 100 m^2/kg)
square millimetre per milligram (mm^2/mg)	(= 1 m^2/kg)
hectare per kilogram (ha/kg)	(= 10 000 m^2/kg)

18.3 A selection of imperial units is:

thousand square feet per pound ($1\ 000\ \text{ft}^2/\text{lb}$)	(= 204.816 m^2/kg)
square yard per ounce (yd^2/oz)	(= 29.493 5 m^2/kg)
square foot per ounce (ft^2/oz)	(= 3.277 06 m^2/kg)
acre per pound (acre/lb)	(= 8 921.79 m^2/kg)
square mile per UK ton (mile^2/ton)	(= 2 549.08 m^2/kg)

For interconversion factors for the above see Table 20.

Table 19 — Mass per unit area (areic mass)
(applicable to sheet metal, plating, etc., and in agriculture)

Exact values are printed in bold type

	kilogram per square metre ^a kg/m ²	pound per thousand square feet lb/1 000 ft ²	ounce per square yard oz/yd ²	ounce per square foot oz/ft ²	pound per acre lb/acre	UK ton per square mile ton/mile ²	kilogram per hectare kg/ha
1 kilogram per square metre ^a kg/m ²	1	204.816	29.493 5	3.277 06	8 921.79	2 549.08	1 × 10 ⁴
1 pound per thousand square feet lb/1 000 ft ²	4.882 43 × 10 ⁻³	1	0.144	0.016	43.56	12.445 7	48.824 3
1 ounce per square yard oz/yd ²	0.033 905 7	6.944 44	1	0.111 111	302.5	86.428 6	339.057
1 ounce per square foot oz/ft ²	0.305 152	62.5	9	1	2 722.5	777.857	3 051.52
1 pound per acre lb/acre	1.120 85 × 10 ⁻⁴	0.022 956 8	3.305 79 × 10 ⁻³	3.673 09 × 10 ⁻⁴	1	0.285 714	1.120 85
1 UK ton per square mile ton/mile ²	3.922 98 × 10 ⁻⁴	0.080 348 9	0.011 570 2	1.285 58 × 10 ⁻³	3.5	1	3.922 98
1 kilogram per hectare kg/ha	1 × 10 ⁻⁴	0.020 481 6	2.949 35 × 10 ⁻³	3.277 06 × 10 ⁻⁴	0.892 179	0.254 908	1
^a 1 kilogram per square metre (kg/m ²) = 1 000 g/m ² = 100 mg/cm ² = 1 mg/mm ² .							

Table 20 — Specific surface, or area per unit mass
(applicable to sheet metal, plating, etc., and in agriculture)

Exact values are printed in bold type

	square metre per kilogram ^a m ² /kg	thousand square feet per pound 1 000 ft ² /lb	square yard per ounce yd ² /oz	square foot per ounce ft ² /oz	acre per pound acre/lb	square mile per UK ton mile ² /ton	hectare per kilogram ha/kg
1 square metre per kilogram ^a m ² /kg	1	4.882 43 × 10 ⁻³	0.033 905 7	0.305 152	1.120 85 × 10 ⁻⁴	3.922 98 × 10 ⁻⁴	1 × 10 ⁻⁴
1 thousand square feet per pound 1 000 ft ² /lb	204.816	1	6.944 44	62.5	0.022 956 8	0.080 348 9	0.020 481 6
1 square yard per ounce yd ² /oz	29.493 5	0.144	1	9	3.305 79 × 10 ⁻³	0.011 570 2	2.949 35 × 10 ⁻³
1 square foot per ounce ft ² /oz	3.277 06	0.016	0.111 111	1	3.673 09 × 10 ⁻⁴	1.285 58 × 10 ⁻³	3.277 06 × 10 ⁻⁴
1 acre per pound acre/lb	8 921.79	43.56	302.5	2 722.5	1	3.5	0.892 179
1 square mile per UK ton mile ² /ton	2 549.08	12.445 7	86.428 6	777.857	0.285 714	1	0.254 908
1 hectare per kilogram ha/kg	1 × 10 ⁴	48.824 3	339.057	3 051.52	1.120 85	3.922 98	1
^a 1 square metre per kilogram (m ² /kg) = 10 ⁻³ m ² /g = 10 ⁻² cm ² /mg = 1 mm ² /mg.							

19 Area per unit capacity

Another combination with somewhat similar application is “area per unit capacity” (used for the “covering power” of paints, etc.). Table 21 gives interconversion factors for square metres per litre, square yards per UK gallon, and square feet per UK gallon.

Table 21 — Area per unit capacity

Exact values are printed in bold type

	square metre per litre m ² /l	square yard per gallon yd ² /gal	square foot per gallon ft ² /gal
1 square metre per litre m ² /l	1	5.437 08	48.933 7
1 square yard per gallon yd ² /gal	0.183 992	1	9
1 square foot per gallon ft ² /gal	0.020 435 8	0.111 111	1

20 Density¹¹⁾ (volumic mass), (mass/volume)

20.1 The coherent SI unit of density is the kilogram per cubic metre (kg/m³), a derived unit. ISO now prefer the term “volumic mass” to density.

20.2 Other commonly used metric units are:

$$\left. \begin{array}{l} \text{gram per cubic centimetre (g/cm}^3\text{)} \\ \text{or} \\ \text{gram per millilitre (g/ml)}^{12)} \end{array} \right\} (= 1\,000 \text{ kg/m}^3)$$

20.3 A selection of imperial units is:

$$\begin{array}{ll} \text{pound per cubic inch (lb/in}^3\text{)} & (= 27\,679.9 \text{ kg/m}^3) \\ \text{pound per cubic foot (lb/ft}^3\text{)} & (= 16.018\,5 \text{ kg/m}^3) \\ \text{UK ton per cubic yard (UKton/yd}^3\text{)} & (= 1\,328.94 \text{ kg/m}^3) \\ \text{pound per UK gallon (lb/UKgal)} & (= 99.776\,3 \text{ kg/m}^3) \\ \text{pound per US gallon (lb/USgal)} & (= 119.826 \text{ kg/m}^3) \end{array}$$

For interconversion factors for the above see Table 22. See also Clause 21, Mass concentration.

¹¹⁾ It should be noted that “relative density” (i.e. density/reference density) is a dimensionless quantity. The relative density of a substance is defined as the ratio of the mass of a given volume of that substance to the mass of an equal volume of a reference substance, under conditions which should be specified for both substances. When the reference substance is water the term “specific gravity” is commonly used for relative density. For conversions of readings of hydrometers on different density and specific gravity bases see BS 718.

¹²⁾ 1 gram per millilitre (1901) = 999.972 kg/m³. See 5.3 and 5.4.

Table 22 — Density (volumic mass) (mass/volume)

Exact values are printed in bold type

	kilogram per cubic metre kg/m ³	gram per cubic centimetre (or gram per millilitre) g/cm ³ (g/ml)	gram per millilitre (1901) g/ml (1901)	pound per cubic inch lb/in ³	pound per cubic foot lb/ft ³	UK ton per cubic yard UKton/yd ³	pound per UK gallon lb/UKgal	pound per US gallon lb/USgal
1 kilogram per cubic metre = kg/m ³	1	0.001	1.000 028 × 10 ⁻³	3.612 73 × 10 ⁻⁵	6.242 80 × 10 ⁻²	7.524 80 × 10 ⁻⁴	1.002 24 × 10 ⁻²	0.834 540 × 10 ⁻²
1 gram per cubic centimetre g/cm ³ = (g/ml)	1 000	1	1.000 028	0.036 127 3	62.428 0	0.752 480	10.022 4	8.345 40
1 gram per millilitre (1901) = g/ml (1901)	999.972	0.999 972	1	0.036 126 3	62.426 2	0.752 459	10.022 1	8.345 17
1 pound per cubic inch = lb/in ³	27 679.9	27.679 9	27.680 7	1	1 728	20.828 6	277.420	231
1 pound per cubic foot = lb/ft ³	16.018 5	0.016 018 5	0.016 018 9	5.787 04 × 10 ⁻⁴	1	0.012 053 6	0.160 544	0.133 681
1 UK ton per cubic yard = UKton/yd ³	1 328.94	1.328 94	1.328 98	0.048 011 0	82.963 0	1	13.319 2	11.090 5
1 pound per UK gallon = lb/UKgal	99.776 3	0.099 776 3	0.099 779 1	3.604 65 × 10 ⁻³	6.228 83	0.075 079 7	1	0.832 674
1 pound per US gallon = lb/USgal	119.826	0.119 826	0.119 830	4.329 00 × 10 ⁻³	7.480 52	0.090 167 0	1.200 95	1

21 Mass concentration (mass/volume)

21.1 In practice “concentration” is “amount of substance concentration”, e.g. “mol/l”. Following discussions between IUPAP (International Union of Pure and Applied Physics) and IUPAC (International Union of Pure and Applied Chemistry), agreement was reached to assign the value 12 exactly to the relative atomic mass (commonly and erroneously called the “atomic weight”) of the isotope of carbon with mass number 12 (carbon 12, ^{12}C). This unified the scale of relative atomic masses. The amount of substance was then defined by fixing the mass of carbon 12 to 0.012 kg, and the “amount of substance” was given the name “mole” (symbol mol). After proposals by IUPAP, IUPAC and ISO, the CIPM (International Committee for Weights and Measures) defined the mole as follows.

1. The mole is the amount of substance of a system which contains as many elementary entities as there are atoms in 0.012 kilogram of carbon 12; its symbol is “mol”.
2. When the mole is used, the elementary entities must be specified and may be atoms, molecules, ions, electrons, other particles, or specified groups of such particles.

In 1980 the CIPM approved the report of the CCU (Consultative Committee for Units) (1980) which specified the following.

“In this definition, it is understood that unbound atoms of carbon 12, at rest and in their ground state, are referred to.”

21.2 The coherent SI unit for the expression of mass concentration¹³⁾ (in the sense of the mass of a substance per unit volume of a solution, or the like) is the kilogram per cubic metre (kg/m^3), a derived unit. This unit is equal to 1 gram per cubic decimetre, and is commonly expressed as 1 gram per litre¹⁴⁾.

$$1 \text{ kg/m}^3 = 1 \text{ g/dm}^3 = 1 \text{ g/l}$$

21.3 Some imperial and US units for the statement of mass concentration are:

grain per cubic foot (gr/ft^3)	(= $0.228\ 835 \times 10^{-2} \text{ kg/m}^3$)
grain per UK gallon (gr/UKgal)	(= $0.014\ 253\ 8 \text{ kg/m}^3$)
grain per US gallon (gr/USgal)	(= $0.017\ 118\ 1 \text{ kg/m}^3$)
ounce per UK gallon (oz/UKgal)	(= $6.236\ 02 \text{ kg/m}^3$)
ounce per US gallon (oz/USgal)	(= $7.489\ 15 \text{ kg/m}^3$)

For interconversion factors for the above see Table 23. See also Clause 20, Density (volumic mass).

¹³⁾ Concentration is sometimes expressed in other ways, for example, mass (of a substance) per unit mass (of a solution), or, in physical chemistry, in terms of moles per unit volume.

¹⁴⁾ 1 gram per litre (1901) = $0.999\ 972 \text{ kg/m}^3$. See 5.3 and 5.4.

Table 23 — Mass concentration

Exact values are printed in bold type

	kg/m ³ g/dm ³ g/l	g/l (1901) ^a	gr/ft ³	gr/UKgal	gr/USgal	oz/UKgal	oz/USgal
1 kilogram per cubic metre kg/m ³ 1 gram per cubic decimetre g/dm ³ 1 gram per litre g/l	1	1.000 028	436.996	70.156 9	58.417 8	0.160 359	0.133 526
1 gram per litre (1901) ^a g/l (1901)	0.999 972	1	436.983	70.154 9	58.416 2	0.160 354	0.133 523
1 grain per cubic foot gr/ft ³	0.228 835 × 10 ⁻²	0.228 842 × 10 ⁻²	1	0.160 544	0.133 681	3.669 57 × 10 ⁻⁴	3.055 56 × 10 ⁻⁴
1 grain per UK gallon gr/UKgal	0.014 253 8	0.014 254 2	6.228 83	1	0.832 674	2.285 71 × 10 ⁻³	1.903 25 × 10 ⁻³
1 grain per US gallon gr/USgal	0.017 118 1	0.017 118 5	7.480 52	1.200 95	1	2.745 03 × 10 ⁻³	2.285 71 × 10 ⁻³
1 ounce per UK gallon oz/UKgal	6.236 02	6.236 20	2 725.11	437.5	364.295	1	0.832 674
1 ounce per US gallon oz/USgal	7.489 15	7.489 36	3 272.73	525.416	437.5	1.200 95	1
NOTE 1 g indicates gram. gr indicates grain.							
NOTE 2 See also Table 22, Density (volumic mass).							
^a 1 litre (1901) per kilogram = 1.000 028 × 10 ⁻³ m ³ /kg. See 5.3 and 5.4.							

22 Specific volume (volume/mass)

22.1 The coherent SI unit of specific volume (which is the reciprocal of density) is the cubic metre per kilogram (m^3/kg), a derived unit.

22.2 Another commonly used metric unit is:

litre¹⁵ per kilogram (l/kg) = $0.001 \text{ m}^3/\text{kg}$.

22.3 A selection of imperial units is:

cubic foot per pound (ft^3/lb)	(= $0.062\ 428\ 0 \text{ m}^3/\text{kg}$)
cubic inch per pound (in^3/lb)	(= $3.612\ 73 \times 10^{-5} \text{ m}^3/\text{kg}$)
cubic foot per UK ton (ft^3/UKton)	(= $2.786\ 96 \times 10^{-5} \text{ m}^3/\text{kg}$)
UK gallon per pound (UKgal/lb)	(= $0.010\ 022\ 4 \text{ m}^3/\text{kg}$)

For interconversion factors for the above see Table 24.

23 Mass rate of flow (mass/time)

23.1 The coherent SI unit of mass rate of flow is the kilogram per second (kg/s), a derived unit.

23.2 Another commonly used metric unit is the kilogram per hour (kg/h).

$1 \text{ kg}/\text{h} = 2.777\ 78 \times 10^{-4} \text{ kg}/\text{s}$

23.3 A selection of imperial units is:

pound per second (lb/s)	(= $0.453\ 592 \text{ kg}/\text{s}$)
pound per hour (lb/h)	(= $1.259\ 98 \times 10^{-4} \text{ kg}/\text{s}$)
UK ton per hour (UKton/h)	(= $0.282\ 235 \text{ kg}/\text{s}$)

For interconversion factors for the above see Table 25.

¹⁵ 1 litre (1901) per kilogram = $1.000\ 028 \times 10^{-3} \text{ m}^3/\text{kg}$. See 5.3 and 5.4.

Table 24 — Specific volume

Exact values are printed in bold type

		cubic metre per kilogram m ³ /kg	litre ^a per kilogram l/kg	cubic foot per pound ft ³ /lb	cubic inch per pound in ³ /lb	cubic foot per UK ton ft ³ /ton	UK gallon per pound UKgal/lb
1 cubic metre per kilogram m ³ /kg	=	1	1 000	16.018 5	27 679.9	35 881.4	99.776 3
1 litre ^a per kilogram l/kg	=	0.001	1	0.016 018 5	27.679 9	35.881 4	0.099 776 3
1 cubic foot per pound ft ³ /lb	=	0.062 428 0	62.428 0	1	1 728	2 240	6.228 83
1 cubic inch per pound in ³ /lb	=	3.612 73 × 10 ⁻⁵	0.036 127 3	5.787 04 × 10 ⁻⁴	1	1.296 30	3.604 65 × 10 ⁻³
1 cubic foot per UK ton ft ³ /ton	=	2.786 96 × 10 ⁻⁵	0.027 869 6	4.464 29 × 10 ⁻⁴	0.771 429	1	2.780 73 × 10 ⁻³
1 UK gallon per pound UKgal/lb	=	0.010 022 4	10.022 4	0.160 544	277.420	359.618	1
^a 1 litre (1901) = 1.000 028 litre. See 5.3 and 5.4.							

Table 25 — Mass rate of flow

Exact values are printed in bold type

		kilogram per second kg/s	kilogram per hour kg/h	pound per second lb/s	pound per hour lb/h	UK ton per hour UKton/h
1 kilogram per second kg/s	=	1	3 600	2.204 62	7 936.64	3.543 14
1 kilogram per hour kg/h	=	$2.777\ 78 \times 10^{-4}$	1	$6.123\ 95 \times 10^{-4}$	2.204 62	$9.842\ 07 \times 10^{-4}$
1 pound per second lb/s	=	0.453 592	1 632.93	1	3 600	1.607 14
1 pound per hour lb/h	=	$1.259\ 98 \times 10^{-4}$	0.453 592	$2.777\ 78 \times 10^{-4}$	1	$4.464\ 29 \times 10^{-4}$
1 UK ton per hour UKton/h	=	0.282 235	1 016.05	0.622 222	2 240	1

24 Volume rate of flow¹⁶⁾ (volume/time)

24.1 The coherent SI unit of volume rate of flow is the cubic metre per second¹⁷⁾ (m^3/s), a derived unit.

24.2 Some other commonly used metric units are:

cubic metre per hour (m^3/h)	(= $2.777\ 78 \times 10^{-4} \text{ m}^3/\text{s}$)
litre ¹⁸⁾ per second (l/s)	(= $0.001 \text{ m}^3/\text{s}$)
litre ¹⁸⁾ per minute (l/min)	(= $1.666\ 67 \times 10^{-5} \text{ m}^3/\text{s}$)
litre ¹⁸⁾ per hour (l/h)	(= $2.777\ 78 \times 10^{-7} \text{ m}^3/\text{s}$)

24.3 A selection of imperial units is:

cubic foot per second ¹⁹⁾ (ft^3/s)	(= $0.028\ 316\ 8 \text{ m}^3/\text{s}$)
cubic foot per hour (ft^3/h)	(= $7.865\ 79 \times 10^{-6} \text{ m}^3/\text{s}$)
UK gallon per second (UKgal/s)	(= $4.546\ 09 \times 10^{-3} \text{ m}^3/\text{s}$)
UK gallon per minute (UKgal/min)	(= $7.576\ 82 \times 10^{-5} \text{ m}^3/\text{s}$)
UK gallon per hour (UKgal/h)	(= $1.262\ 80 \times 10^{-6} \text{ m}^3/\text{s}$)

For interconversion factors for the above see Table 26.

¹⁶⁾ For gases, the conversion factors given here are based on the assumption that the reference conditions of temperature, pressure and humidity remain unchanged.

¹⁷⁾ The cubic metre per second is sometimes known as the “cume²”.

¹⁸⁾ The litre (1901) = 1.000 028 litre (see 5.3 and 5.4).

¹⁹⁾ The cubic foot per second is sometimes known as the “cusec”.

Table 26 — Volume rate of flow

Exact values are printed in bold type

	cubic metre per second m ³ /s	cubic metre per minute m ³ /min	cubic metre per hour m ³ /h	litre^a per second l/s	litre^a per minute l/min	litre^a per hour l/h	cubic foot per second ft ³ /s	cubic foot per minute ft ³ /min	cubic foot per hour ft ³ /h	UK gallon per second UKgal/s	UK gallon per minute UKgal/min	UK gallon per hour UKgal/h
1 cubic metre per second = m ³ /s	1	60	3 600	1 000	60 000	3.6 × 10⁶	35.314 7	2.118 88 × 10 ³	127 133	219.969	13 198.2	791 889
1 cubic metre per minute = m ³ /min	0.016 666 7	1	60	16.666 7	1 000	60 000	0.588 578	35.314 7	2.118 88 × 10 ³	3.666 15	219.969	13 198.2
1 cubic metre per hour = m ³ /h	2.777 78 × 10 ⁻⁴	0.016 666 7	1	0.277 778	16.666 7	1 000	9.809 63 × 10 ⁻³	0.588 578	35.314 7	0.061 102 6	3.666 15	219.969
1 litre ^a per second = l/s	0.001	0.06	3.6	1	60	3 600	0.035 314 7	2.118 88	127.133	0.219 969	13.198 2	791.889
1 litre ^a per minute = l/min	1.666 67 × 10 ⁻⁵	0.001	0.06	0.016 667	1	60	5.885 78 × 10 ⁻⁴	0.035 314 7	2.118 88	3.666 15 × 10 ⁻³	0.219 969	13.198 2
1 litre ^a per hour = l/h	0.277 778 × 10 ⁻⁶	1.666 67 × 10 ⁻⁵	0.001	0.277 778 × 10 ⁻³	0.016 666 7	1	9.809 63 × 10 ⁻⁶	5.885 78 × 10 ⁻⁴	0.035 314 7	6.110 26 × 10 ⁻⁵	3.666 15 × 10 ⁻³	0.219 969
1 cubic foot per second = ft ³ /s	0.028 316 8	1.699 01	101.941	28.316 8	1 699.01	101 941	1	60	3 600	6.228 84	373.730	2.242 38 × 10 ⁴
1 cubic foot per minute = ft ³ /min	4.719 47 × 10 ⁻⁴	0.028 316 8	1.699 01	0.471 947	28.316 8	1.699 01 × 10 ³	0.016 666 7	1	60	0.103 814	6.228 84	373.730
1 cubic foot per hour = ft ³ /h	7.865 79 × 10 ⁻⁶	4.719 47 × 10 ⁻⁴	0.028 316 8	7.865 79 × 10 ⁻³	0.471 947	28.316 8	0.277 778 × 10 ⁻³	0.016 666 7	1	1.730 23 × 10 ⁻³	0.103 814	6.228 84
1 UK gallon per second = UKgal/s	4.546 09 × 10 ⁻³	0.272 765	16.365 9	4.546 09	272.765	16 365.9	0.160 544	9.632 62	577.957	1	60	3 600
1 UK gallon per minute = UKgal/min	7.576 82 × 10 ⁻⁵	4.546 09 × 10 ⁻³	0.272 765	0.075 768 2	4.546 09	272.765	2.675 73 × 10 ⁻³	0.160 544	9.632 62	0.016 666 7	1	60
1 UK gallon per hour = UKgal/h	1.262 80 × 10 ⁻⁶	7.576 82 × 10 ⁻⁵	4.546 09 × 10 ⁻³	1.262 80 × 10 ⁻³	0.075 768 2	4.546 09	4.459 55 × 10 ⁻⁵	2.675 73 × 10 ⁻³	0.160 544	0.277 778 × 10 ⁻³	0.016 666 7	1

NOTE For gases, the conversion factors given in this table are based on the assumption that the reference conditions of temperature, pressure and humidity remain unchanged.

^a The litre used here is equal to 1 cubic decimetre. 1 litre (1901) = 1.000 028 dm³ (see 5.3 and 5.4).

25 Traffic factors

(in connection with volume of fuel consumed, distance run and load carried)

It should be noted that in mainland European countries fuel consumptions are usually expressed in terms of litres per kilometre, or litres per 100 kilometres, i.e. volume of fuel per distance run (see Table 27).

In the UK the reciprocal factor (distance/volume) in terms of miles per gallon is used (see Table 28). As fuel in the UK is now sold in litres, consumption is sometimes calculated in miles per litre.

Table 27 — Fuel consumption (volume/distance)

Exact values are printed in bold type

		litre per kilometre ^a l/km	UK gallon per mile UKgal/mile	US gallon per mile USgal/mile	litre per mile l/mile
1 litre per kilometre ^a l/km	=	1	0.354 006	0.425 144	1.609 344
1 UK gallon per mile UKgal/mile	=	2.824 81	1	1.200 95	4.546 09
1 US gallon per mile USgal/mile	=	2.352 15	0.832 674	1	3.785 41
1 litre per mile l/mile	=	0.621 371	0.219 969	0.264 172	1

^a Several mainland European countries use the factor "litre per 100 kilometres".

Table 28 — Fuel consumption (distance/volume)

Exact values are printed in bold type

		kilometre per litre km/l	mile per UK gallon mile/UKgal	mile per US gallon mile/USgal	mile per litre mile/l
1 kilometre per litre km/l	=	1	2.824 81	2.352 15	0.621 371
1 mile per UK gallon mile/UKgal	=	0.354 006	1	0.832 674	0.219 969
1 mile per US gallon mile/USgal	=	0.425 144	1.200 95	1	0.264 172
1 mile per litre mile/l	=	1.609 344	4.546 09	3.785 41	1

Mass carried × distance

1 tonne kilometre = 0.611 558 UKton mile
1 UKton mile = 1.635 17 tonne kilometre

Mass carried × distance/volume

1 tonne kilometre per litre = 2.780 20 UKton mile per UK gallon
1 UKton mile per UK gallon = 0.359 687 tonne kilometre per litre

26 Moment of inertia (mass × length squared)

26.1 The coherent SI unit of moment of inertia is the kilogram metre squared ($\text{kg}\cdot\text{m}^2$), a derived unit.

26.2 Some other metric units which have been used are:

$$\begin{aligned} \text{kilogram millimetre squared (kg}\cdot\text{mm}^2) &= 10^{-6} \text{ kg}\cdot\text{m}^2 \\ \text{gram centimetre squared (g}\cdot\text{cm}^2) &= 10^{-7} \text{ kg}\cdot\text{m}^2 \end{aligned}$$

26.3 A selection of imperial units is:

$$\begin{aligned} \text{pound foot squared (lb}\cdot\text{ft}^2) &= 0.042\,140\,1 \text{ kg}\cdot\text{m}^2 \\ \text{pound inch squared (lb}\cdot\text{in}^2) &= 2.926\,40 \times 10^{-4} \text{ kg}\cdot\text{m}^2 \\ \text{ounce (avoir) inch squared (oz}\cdot\text{in}^2) &= 1.829\,00 \times 10^{-5} \text{ kg}\cdot\text{m}^2 \end{aligned}$$

For interconversion factors for the above see Table 29.

Table 29 — Moment of inertia

Exact values are printed in bold type

		kilogram metre squared $\text{kg}\cdot\text{m}^2$	pound foot squared $\text{lb}\cdot\text{ft}^2$	pound inch squared $\text{lb}\cdot\text{in}^2$	ounce inch squared $\text{oz}\cdot\text{in}^2$
1 kilogram metre squared $\text{kg}\cdot\text{m}^2$	=	1	23.730 4	3 417.17	54 674.8
1 pound foot squared $\text{lb}\cdot\text{ft}^2$	=	0.042 140 1	1	144	2 304
1 pound inch squared $\text{lb}\cdot\text{in}^2$	=	$2.926\,40 \times 10^{-4}$	$6.944\,44 \times 10^{-3}$	1	16
1 ounce inch squared $\text{oz}\cdot\text{in}^2$	=	$1.829\,00 \times 10^{-5}$	$4.340\,28 \times 10^{-4}$	0.062 5	1

NOTE 1 $\text{kg}\cdot\text{m}^2 = 10^6 \text{ kg}\cdot\text{mm}^2 = 10^7 \text{ g}\cdot\text{cm}^2$.

27 Momentum (linear) (mass × velocity)

The coherent SI unit of momentum is the kilogram metre per second.

Some key conversion factors are:

$$\begin{aligned} 1 \text{ kg}\cdot\text{m/s} &= 7.233\,01 \text{ lb}\cdot\text{ft/s} \\ 1 \text{ lb}\cdot\text{ft/s} &= 0.138\,255 \text{ kg}\cdot\text{m/s} \\ (1 \text{ kg}\cdot\text{m/s}) &= 10^5 \text{ g}\cdot\text{cm/s} \end{aligned}$$

28 Angular momentum (mass × velocity × length)

The coherent SI unit of angular momentum is the kilogram metre squared per second.

Some key conversion factors are:

$$\begin{aligned} 1 \text{ kg}\cdot\text{m}^2/\text{s} &= 23.730\,4 \text{ lb}\cdot\text{ft}^2/\text{s} \\ 1 \text{ lb}\cdot\text{ft}^2/\text{s} &= 0.042\,140\,1 \text{ kg}\cdot\text{m}^2/\text{s} \end{aligned}$$

29 Force (mass × acceleration)

29.1 The coherent SI unit of force is the newton (N), a derived unit with a special name. Expressed in terms of base units of the SI, the newton is the kilogram metre per second squared ($\text{kg}\cdot\text{m}/\text{s}^2$) and is that force which, when applied to a body having a mass of one kilogram, gives it an acceleration of one metre per second squared.

29.2 Other metric units of force of historical or practical importance are:

the dyne (dyn), the force unit in the centimetre-gram-second system;

the sthène (sn), the force unit in the metre-tonne-second system; and

the kilogram-force (kgf), which is often described as the metric technical unit of force. In Germany and some other mainland European countries the kilogram-force is called the kilopond (symbol kp).

$$1 \text{ dyn} = 1 \text{ g}\cdot\text{cm}/\text{s}^2 \quad (= 10^{-5} \text{ N})$$

$$1 \text{ sn} = 1 \text{ t}\cdot\text{m}/\text{s}^2 \quad (= 10^3 \text{ N})$$

The kilogram-force (or kilopond) is that force which, when applied to a body having a mass of one kilogram, gives it the standard acceleration²⁰⁾ due to gravity (i.e. **9.806 65** m/s^2).

Thus:

$$1 \text{ kgf (or kp)} = \mathbf{9.806\ 65} \text{ kg}\cdot\text{m}/\text{s}^2 \quad (= \mathbf{9.806\ 65} \text{ N})$$

29.3 In the foot-pound-second system the coherent force unit is the poundal (pdl).

$$\begin{aligned} 1 \text{ pdl} &= 1 \text{ lb}\cdot\text{ft}/\text{s}^2 \\ &= \mathbf{0.453\ 592\ 37} \times \mathbf{0.304\ 8} \text{ kg}\cdot\text{m}/\text{s}^2 \\ & (= 0.138\ 255 \text{ N}) \text{ (approximately)} \end{aligned}$$

The corresponding technical force unit in the UK and USA is the pound-force (lbf). It is that force which, when applied to a body having a mass of one pound, gives it the standard acceleration²⁰⁾ due to gravity.

Thus:

$$\begin{aligned} 1 \text{ lbf} &= \frac{\mathbf{9.806\ 65}}{\mathbf{0.304\ 8}} \text{ lb}\cdot\text{ft}/\text{s}^2 \\ &= 32.174\ 0 \text{ pdl (approximately)} \\ & (= 4.448\ 22 \text{ N})^{21)} \end{aligned}$$

Further technical force units associated with the pound-force are the ounce-force (ozf), the UK ton-force (tonf) and the US ton-force. In the USA a unit of 1 000 lbf named the “kip” is often used.

$$\begin{aligned} 1 \text{ ozf} &= \frac{1}{16} \text{ lbf} \quad (= 0.278\ 014 \text{ N}) \\ 1 \text{ tonf} &= 2\ 240 \text{ lbf} \quad (= 9\ 964.02 \text{ N}) \\ 1 \text{ US ton-force} &= 2\ 000 \text{ lbf} \quad (= 8\ 896.44 \text{ N}) \\ 1 \text{ kip (USA only)} &= 1\ 000 \text{ lbf} \quad (= 4\ 448.22 \text{ N}) \end{aligned}$$

29.4 The kilogram-force (kgf), and the pound-force (lbf) and its associated units, are both exactly defined in terms of the standard acceleration due to gravity. Because local acceleration due to gravity usually differs slightly from standard acceleration, it follows that the forces exerted by gravity on bodies having a mass of 1 kg or 1 lb are rarely exactly equal to 1 kgf or 1 lbf respectively, and account has to be taken of this when very high precision is required. See also Clause 30, Weight.

Interconversion factors for the above units are given in, or can be readily inferred from, Table 30.

²⁰⁾ See 14.4

²¹⁾ In exact terms, $0.453\ 592\ 37 \times 9.806\ 65 \text{ N}$.

Table 30 — Force

Exact values are printed in bold type

		newton N	kilogram-force kgf	poundal pdl	pound-force lbf	UK ton-force tonf	ounce-force ozf
1 newton N	=	1	0.101 972	7.233 01	0.224 809	$1.003\ 61 \times 10^{-4}$	3.596 94
1 kilogram-force kgf	=	9.806 65	1	70.931 6	2.204 62	$9.842\ 07 \times 10^{-4}$	35.274 0
1 poundal pdl	=	0.138 255	0.014 098 1	1	0.031 081 0	$1.387\ 54 \times 10^{-5}$	0.497 295
1 pound-force lbf	=	4.448 22	0.453 592	32.174 0	1	$4.464\ 29 \times 10^{-4}$	16
1 UK ton-force tonf	=	9 964.02	1 016.05	72 069.9	2 240	1	35 840
1 ounce-force ozf	=	0.278 014	0.028 349 5	2.010 88	0.062 5	$2.790\ 18 \times 10^{-5}$	1
NOTE							
1 dyne (dyn) = 10^{-5} N (see 29.2)							
1 sthène (sn) = 10^3 N (see 29.2)							
1 kip (USA only) = 1 000 lbf (see 29.3)							
1 US ton-force = 2 000 lbf (see 29.3)							

30 Weight

30.1 Meaning of “weight”

The term “weight” is commonly used to denote either mass or force, i.e. the mass of a body or the force of gravity acting upon it. It is used in the sense of “mass” in the UK Weights and Measures Act, 1985 [1] and in common parlance; it is used in the sense of force by the CGPM and in scientific and some technical work. *The International System of Units (SI)*, 7th edition 1998 [3] states that “the word ‘weight’ denotes a quantity of the same nature as a ‘force’: the weight of a body is the product of its mass and the acceleration due to gravity: in particular, the standard weight of a body is the product of its mass and the standard acceleration due to gravity.” The UK Weights and Measures Act 1985 [1], refers to “measurement of mass or weight”, but specifies that the kilogram is the unit of mass.

As weight values may be found quoted in either mass or force units both usages are accommodated in the conversion tables. To convert weight units when using weight in the mass sense Table 15, Table 16 or Table 17 should be used; to convert weight units when using weight as a force Table 30 should be used.

30.2 Relationship between force of gravity and mass

The force of gravity (for example, expressed in newtons) is equal to the mass (in kilograms) multiplied by the local gravitational acceleration (in metres per second squared). For most practical purposes variations in local gravitational acceleration can be ignored and the standard value of **9.806 65** m/s² is assumed (usually rounded to 9.81 m/s²).

It is the standard value of **9.806 65** m/s² that is used in defining with precision the technical force units, the kilogram-force and the pound-force.

30.3 Accurate weight conversions

Conversions from one system of units to another on a mass to mass basis, or on a force to force basis, can be made with good accuracy by using the tables for Clause 15 (Table 15, Table 16 and Table 17) and Clause 29 (Table 30), respectively. However, to obtain the accurate relationship of a mass to its associated gravitational force account has to be taken of the exact local value of the earth’s gravitational field. The downward force on the mass is also affected by the buoyancy of any displaced atmosphere.

31 Moment of force, or torque (force × length)

31.1 The coherent SI unit of moment (of force) is the newton metre (N·m), a derived unit. See Note on Clause 31 concerning the energy unit, which has a different physical significance.

31.2 A metric unit often used for moment, or torque, in mainland European countries is the kilogram-force metre (kgf·m).

$$1 \text{ kgf} \cdot \text{m} = \mathbf{9.806\ 65} \text{ N} \cdot \text{m}$$

This unit is called the kilopond metre (kp·m) in Germany.

31.3 A selection of imperial units is:

poundal foot (pdl·ft)	(= 0.042 140 1 N·m)
pound-force foot (lbf·ft)	(= 1.355 82 N·m)
pound-force inch (lbf·in)	(= 0.112 985 N·m)
UK ton-force foot (tonf·ft)	(= 3 037.03 N·m)
ounce-force inch (ozf·in)	(= 7.061 55 × 10 ⁻³ N·m)

Note on Clause 31

NOTE The product newton \times metre (N·m) also expresses the SI unit for work done, or energy, a unit having the special name joule (J), (see Clause 37, Energy). However, torque and energy are different physical quantities; both are dimensionally force \times length but in the former the directions of the force and length components are perpendicular to each other while in the latter they are in line with each other.

With imperial units a distinction between torque units and energy units is made (by convention) by reversing the order of the units; e.g. the foot pound-force (ft·lbf) is an energy unit and the pound-force foot (lbf·ft) a torque unit. There is no similar convention used, or advisable, with metric (including SI) units; it can be seen for example that m·N (the reverse of N·m) would easily be mistaken for millinewton. However, in practice both ft·lb and N·m may be seen indicating torque.

Metric moment or torque units should be expressed as indicated in 31.1 and 31.2.

It may be useful to point out that because both torque and energy are dimensionally the same (force \times length) there is a numerical correspondence between energy conversion tables and torque conversion tables.

For interconversion factors for the above see Table 31.

Table 31 — Moment of force (torque)

Exact values are printed in bold type

		newton metre N·m	kilogram-force ^a metre kgf·m	poundal foot pdl·ft	pound-force foot lbf·ft	pound-force inch lbf·in	UK ton-force foot tonf·ft	ounce-force inch ozf·in
1 newton metre N·m	=	1	0.101 972	23.730 4	0.737 562	8.850 75	$3.292\ 69 \times 10^{-4}$	141.612
1 kilogram-force ^a metre kgf·m	=	9.806 65	1	232.715	7.233 01	86.796 2	$3.229\ 02 \times 10^{-3}$	1 388.74
1 poundal foot pdl·ft	=	0.042 140 1	$4.297\ 10 \times 10^{-3}$	1	0.031 081 0	0.372 971	$1.387\ 54 \times 10^{-5}$	5.967 54
1 pound-force foot lbf·ft	=	1.355 82	0.138 255	32.174 0	1	12	$4.464\ 29 \times 10^{-4}$	192
1 pound-force inch lbf·in	=	0.112 985	0.011 521 2	2.681 17	0.083 333 3	1	$3.720\ 24 \times 10^{-5}$	16
1 UK ton-force foot tonf·ft	=	3 037.03	309.691	72 069.9	2 240	26 880	1	430 080
1 ounce-force inch ozf·in	=	$7.061\ 55 \times 10^{-3}$	$7.200\ 78 \times 10^{-4}$	0.167 573	$5.208\ 33 \times 10^{-3}$	0.062 5	$2.325\ 15 \times 10^{-6}$	1
NOTE 1 newton millimetre (N·mm) = 10^{-3} newton metre (N·m). 1 dyne centimetre (dyn·cm) = 10^{-7} newton metre (N·m).								
^a The kilogram-force is called the kilopond (kp) in Germany. 1 kgf·m = 1 kp·m								

32 Force per unit length²²⁾ (force/length)

32.1 The coherent SI unit is the newton per metre (N/m).

32.2 Another metric unit that may still be encountered is the dyne per centimetre (dyn/cm).

$$1 \text{ dyn/cm} = 10^{-3} \text{ N/m}$$

No interconversion tables are provided for force per unit length. The main reason for mentioning it here is to show the distinction from torque.

33 Pressure (force/area)

33.1 General

33.1.1 The coherent SI unit of pressure is the newton per square metre, N/m^2 , for which the special name pascal (symbol Pa) was approved by the CGPM in 1971.

One pascal represents a very small pressure, and its multiples kilopascal (kPa) (or kN/m^2) and megapascal (MPa) (or MN/m^2) are therefore frequently used.

33.1.2 Arising from the historical evolution of the SI from the CGS system, some pressure units have a decimal relationship with the pascal. These are the dyne per square centimetre (dyn/cm^2) (sometimes called the barye), the pièze (pz), and the bar ($\text{bar}^{23)$). Only the last of these, with its multiples, persists in common use.

The dyne per square centimetre is also a very small pressure:

$$1 \text{ dyn/cm}^2 = \frac{1 \times 10^{-5} \text{ N}}{(\text{m}/100)^2} = 0.1 \text{ N/m}^2 = 0.1 \text{ Pa}$$

The pièze is the coherent pressure unit in the metre-tonne-second system, being equal to one sthène per square metre:

$$1 \text{ pz} = 1 \text{ sn/m}^2 = 1 \times 10^3 \text{ N/m}^2 = 1 \text{ kN/m}^2 = 1 \text{ kPa}.$$

The bar, 10^6 dyn/cm^2 , is legally recognized in EU countries and has a magnitude not far removed from that of usual atmospheric pressure at sea level.

$$1 \text{ bar} = 10^6 \text{ dyn/cm}^2 = 10^6 \times 0.1 \text{ N/m}^2 = 10^5 \text{ N/m}^2 = 10^5 \text{ Pa}.$$

One of its submultiples, the millibar, is widely used in the expression of barometric pressures.

33.1.3 Also in common use in mainland European countries are the technical pressure units, the kilogram-force per square metre, and, in particular, the kilogram-force per square centimetre:

$$\begin{aligned} 1 \text{ kgf/m}^2 &= 9.806 \ 65 \text{ N/m}^2 \text{ (exactly)} &= 9.806 \ 65 \text{ Pa} \\ 1 \text{ kgf/cm}^2 &= 0.980 \ 665 \times 10^5 \text{ N/m}^2 \text{ (exactly)} &= 0.098 \ 066 \ 5 \text{ MPa} \end{aligned}$$

In Germany and some other mainland European countries the kilopond (kp) is used in place of the kilogram-force (kgf), e.g. $1 \text{ kp/cm}^2 = 1 \text{ kgf/cm}^2$.

These technical units have a simple relationship with conventional columns of water expressed in metric terms (see **33.2**).

²²⁾ For example, surface tension.

²³⁾ The internationally recognized unit symbol for the bar is the same as the unit name. In meteorology, however, the commonly used symbol for the millibar is simply mb.

33.1.4 Some imperial units that are expressed directly in terms of force per unit area, are as follows. All except the first are technical units.

poundal per square foot (pdl/ft²), the coherent unit in the foot-pound-second system

pound-force per square foot (lbf/ft²)

pound-force per square inch (lbf/in²)

UK ton-force per square foot (tonf/ft²)

UK ton-force per square inch (tonf/in²)

$$1 \text{ pdl/ft}^2 = \frac{0.453\ 592\ 37 \times (0.304\ 8) \text{ N}}{(0.304\ 8 \text{ m})^2}$$

$$= 1.488\ 16 \text{ Pa (or N/m}^2\text{)} \quad \text{approx.}$$

$$1 \text{ lbf/ft}^2 = \frac{0.453\ 592\ 37 \times 9.806\ 65 \text{ N}}{(0.304\ 8 \text{ m})^2}$$

$$= 47.880\ 3 \text{ Pa (or N/m}^2\text{)} \quad \text{approx.}$$

$$1 \text{ lbf/in}^2 = 144 \text{ lbf/ft}^2 = 6\ 894.76 \text{ Pa (or N/m}^2\text{)} \quad \text{approx.}$$

$$1 \text{ UKtonf/ft}^2 = 2\ 240 \text{ lbf/ft}^2 = 1.072\ 52 \times 10^5 \text{ Pa (or N/m}^2\text{)} \quad \text{approx.}$$

$$1 \text{ UKtonf/in}^2 = 2\ 240 \text{ lbf/in}^2 = 1.544\ 43 \times 10^7 \text{ Pa (or N/m}^2\text{)} \quad \text{approx.}$$

The pound-force per square inch (lbf/in²) is often known and shown by the abbreviation p.s.i., but, although widely used in the UK and USA, this abbreviation is inconsistent with the internationally recognized symbology for units. In the USA, the expression “k.s.i.” is often used to signify kips per square inch (i.e. 1 000 lbf/in²).

33.2 Liquid columns

Pressures are often measured in terms of the height of a column of liquid, e.g. of mercury or of water. The pressure associated with a given height is dependent upon the density of the liquid and the local acceleration due to gravity. The following pressure units are based upon conventional density and gravity conditions:

the conventional millimetre of mercury (symbol mmHg);

the conventional inch of mercury (symbol inHg);

the conventional millimetre of water (symbol mmH₂O);

the conventional metre of water (symbol mH₂O);

the conventional inch of water (symbol inH₂O);

the conventional foot of water (symbol ftH₂O).

$$1 \text{ mmH}_2\text{O} = 0.001 \text{ m} \times 1\,000 \text{ kg/m}^3 \times 9.806\,65 \text{ m/s}^2 = 9.806\,65 \text{ N/m}^2 = \mathbf{9.806\,65 \text{ Pa}}$$

(This is the pressure due to an ideal column of water of length 1 mm and of uniform density 1 g/cm^3 , when under the standard condition $g_n = \mathbf{9.806\,65 \text{ m/s}^2}$)

$$\begin{aligned} 1 \text{ mmHg} &= 13.595\,1 \text{ mmH}_2\text{O} &= \mathbf{13.595\,1} \times \mathbf{9.806\,65 \text{ Pa}} \\ & &= 133.322 \text{ Pa (approx.)}^{24)} \\ 1 \text{ inH}_2\text{O} &= 25.4 \text{ mmH}_2\text{O} &= \mathbf{9.806\,65} \times \mathbf{25.4 \text{ Pa}} \\ & &= 249.089 \text{ Pa (approx.)} \\ 1 \text{ inHg} &= 25.4 \text{ mmHg} &= \mathbf{9.806\,65} \times \mathbf{13.595\,1} \times \mathbf{25.4 \text{ Pa}} \\ & &= 3\,386.39 \text{ Pa (approx.)} \\ 1 \text{ ftH}_2\text{O} &= 304.8 \text{ mmH}_2\text{O} &= \mathbf{304.8} \times \mathbf{9.806\,65 \text{ Pa}} \\ & &= 2\,989.07 \text{ Pa (approx.)} \end{aligned}$$

Another pressure unit in common use, known as the torr, is equal, within one part in 7 million, to the conventional millimetre of mercury (mmHg). It is, however, precisely defined in terms of the pascal as follows:

$$\begin{aligned} 1 \text{ torr} &= \frac{\mathbf{101\,325.0}}{\mathbf{760}} \text{ Pa} \\ &= 133.322 \text{ Pa (approx.)} \end{aligned}$$

Because of its size, the pascal is well-suited to vacuum technology. However, in addition to the millibar and torr and their submultiples, the term “micron” meaning micrometre of mercury (μmHg), is still common in this field. The symbol μmHg is sometimes (incorrectly) contracted to μ .

Following from its definition, the conventional millimetre of water is exactly equal to the kilogram-force per square metre:

$$\begin{aligned} 1 \text{ mmH}_2\text{O} &= 1 \text{ kgf/m}^2 \text{ (or kp/m}^2\text{)} \\ \text{Similarly, } 10 \text{ mmH}_2\text{O} &= 1 \text{ kgf/cm}^2 \text{ (or kp/cm}^2\text{)}. \end{aligned}$$

33.3 Atmospheres

Attention is called to the significance of the following terms and symbols.

Standard atmosphere (atm). This is an internationally established reference for pressure of 101 325 Pa, being equal to 760 mmHg within one part in 7 million. It should not be regarded or used as a unit, but it is of great importance and in widespread use as a reference.

Technical atmosphere (at). This unit, which is used in mainland European countries, is equal to the kilogram-force per square centimetre, or kilopond per square centimetre:

$$1 \text{ at} = 1 \text{ kgf/cm}^2 \text{ (or kp/cm}^2\text{)} = \mathbf{98\,066.5 \text{ Pa}}$$

33.4 “Absolute” and “gauge” pressure

33.4.1 All the pressure units mentioned in **33.1**, **33.2** and **33.3** may be used to state the magnitude of an absolute pressure or of a pressure difference, and misunderstandings in interpretation and conversion may arise if the quantity concerned is not clearly expressed.

33.4.2 It has been internationally recommended that pressure units themselves should not be modified to indicate whether the pressure value is “absolute” (i.e. with zero pressure as the datum) or “gauge” (i.e. with atmospheric pressure as the datum).

²⁴⁾ For detailed information on barometer conventions see BS 2520.

33.4.3 Both in the UK and USA it was common practice to use the abbreviation p.s.i. to indicate lbf/in², and to differentiate between gauge and absolute pressures by adding the further letters “g” and “a” to make “p.s.i.g.” and “p.s.i.a.”, respectively. A similar situation existed in German practice, where the symbol for the technical atmosphere (at) was modified to atü or ata to indicate the expression of “gauge” (über) or “absolute” pressure respectively²⁵⁾. Of these, only “at” was an internationally recognized unit symbol; furthermore the modifications did not change the units of measurement, but were in fact an indication of the quantity being expressed.

33.4.4 From the recommendation in **33.4.2** it follows that, if the context leaves any doubt as to which quantity is meant, the word “pressure” should be qualified appropriately:

- e.g. “at a gauge pressure of 12.5 bar”
- or “at a gauge pressure of 1.25 MPa”
- or “at an absolute pressure of 2.34 bar”
- or “at an absolute pressure of 234 kPa”.

33.4.5 Absolute pressures are always positive, but gauge pressures are shown as negative when indicating a pressure less than the datum pressure.

It is common practice in the power and process industries to refer to “vacuum” values, e.g. “1 mmHg vacuum” represents a gauge pressure of –1 mmHg, and “one per cent of vacuum” represents a gauge pressure of minus one per cent of the datum atmosphere in use.

Interconversion factors for the above units are given in, or can be deduced from, Table 32, Table 33 and Table 34. See also Clause 34, Stress.

34 Stress (force/area)

Though it is a different physical quantity, stress is naturally treated with pressure, since it is also force divided by area. Many, but not all, of the units mentioned in connection with pressure are used for stress, so the conversion factors in Table 32, Table 33 and Table 34 might be found useful.

The coherent SI unit of stress is again the pascal (Pa), i.e. the newton per square metre (N/m²).

Technical units that have been widely used for stresses in metals and some other materials are the kilogram-force per square millimetre (kgf/mm²), pound-force per square inch (lbf/in²) and UKton-force per square inch (UKtonf/in²)²⁶⁾. In the change to SI, a practical unit of similar size to the kgf/mm² was sought, the first proposal being the hectobar (hbar), which came into some use. However, the hbar has been superseded by the N/mm², which can be otherwise stated as MN/m² or MPa:

$$\begin{array}{lll}
 1 \text{ N/mm}^2 & = & 1 \text{ MN/m}^2 & = & 1 \text{ MPa} \\
 1 \text{ kgf/mm}^2 & = & \mathbf{9.806\ 65} \text{ N/mm}^2 & = & \mathbf{9.806\ 65} \text{ MPa} \\
 1 \text{ hbar} & = & 100 \text{ bar} = 10^2 \times 10^5 \text{ N/m}^2 & = & 10 \text{ MPa}
 \end{array}$$

Interconversion factors for these and other units used for stresses are given in, or can be deduced from, Table 32, Table 33 and Table 34. See also Clause 33, Pressure.

²⁵⁾ With the atü, the datum is an absolute pressure of 1 at.

²⁶⁾ In the USA, the expression “k.s.i.” is often used to signify kips per square inch (i.e. 1 000 lbf/in²).

Table 32 — Pressure

Exact values are printed in bold type

	pascal Pa (or newton per square metre N/m ²)	newton per square millimetre ^a N/mm ²	hectobar hbar	kilogram- force per square centimetre ^b kgf/cm ²	poundal per square foot pdl/ft ²	pound-force per square inch lbf/in ²	pound- force per square foot lbf/ft ²	UKton-force per square inch tonf/in ²	UKton-force per square foot tonf/ft ²
1 pascal = 1 newton per square metre Pa N/m ²	1	1 × 10 ⁻⁶	1 × 10 ⁻⁷	1.019 72 × 10 ⁻⁵	0.671 969	1.450 38 × 10 ⁻⁴	0.020 885 4	6.474 90 × 10 ⁻⁸	9.323 85 × 10 ⁻⁶
1 newton per square millimetre ^a N/mm ²	1 × 10⁶	1	0.1	10.197 2	671 969	145.038	20 885.4	6.474 90 × 10 ⁻²	9.323 85
1 hectobar hbar	1 × 10⁷	10	1	101.97 2	6 719 690	1 450.38	208 854	0.647 490	93.238 5
1 kilogram-force per square centimetre ^b kgf/cm ²	9.806 65 × 10⁴	9.806 65 × 10⁻²	9.806 65 × 10⁻³	1	65 897.6	14.223 3	2 048.16	6.349 71 × 10 ⁻³	0.914 358
1 poundal per square foot pdl/ft ²	1.488 16	1.488 16 × 10 ⁻⁶	1.488 16 × 10 ⁻⁷	1.517 50 × 10 ⁻⁵	1	2.158 40 × 10 ⁻⁴	0.031 081 0	9.635 71 × 10 ⁻⁸	1.387 54 × 10 ⁻⁵
1 pound-force per square inch lbf/in ²	6.894 76 × 10 ³	6.894 76 × 10 ⁻³	6.894 76 × 10 ⁻⁴	0.070 307 0	4 633.06	1	144	4.464 29 × 10 ⁻⁴	0.064 285 7
1 pound-force per square foot lbf/ft ²	47.880 3	4.788 03 × 10 ⁻⁵	4.788 03 × 10 ⁻⁶	4.882 43 × 10 ⁻⁴	32.174 0	6.944 44 × 10 ⁻³	1	3.100 20 × 10 ⁻⁶	4.464 29 × 10 ⁻⁴
1 UKton-force per square inch tonf/in ²	1.544 43 × 10 ⁷	15.444 3	1.544 43	157.488	1.037 81 × 10 ⁷	2 240	322 560	1	144
1 UKton-force per square foot tonf/ft ²	1.072 52 × 10 ⁵	0.107 252	1.072 52 × 10 ⁻²	1.093 66	72 069.9	15.555 6	2 240	6.944 44 × 10 ⁻³	1

^a 1 N/mm² (used for stress in metals) = 1 MN/m² = 1 MPa.
^b Also called the kilopond per square centimetre (kp/cm²) and sometimes known as the “technical atmosphere” (at). It is equal to 10 mH₂O (see 33.2).
 1 kgf/mm² = 100 kgf/cm²

Table 33 — Pressure (continued)

Exact values are printed in bold type

	pascal Pa (or newton per square metre N/m ²)	bar bar	millibar mbar ^a	standard atmosphere atm	kilogram-force per square centimetre ^b kgf/cm ²	pound-force per square inch lbf/in ²	torr ^c	(conventional) inch of mercury inHg	
1 pascal = Pa	1 newton per square metre N/m ²	1	1 × 10⁻⁵	0.01	9.869 23 × 10 ⁻⁶	1.019 72 × 10 ⁻⁵	1.450 38 × 10 ⁻⁴	0.750 062 × 10 ⁻²	2.953 00 × 10 ⁻⁴
1 bar bar		1 × 10⁵	1	1 000	0.986 923	1.019 72	14.503 8	750.062	29.530 0
1 millibar mbar ^a		100	0.001	1	9.869 23 × 10 ⁻⁴	1.019 72 × 10 ⁻³	0.014 503 8	0.750 062	0.029 530 0
1 standard atmosphere atm		101 325.0	1.013 25	1 013.250	1	1.033 23	14.695 9	760	29.921 3
1 kilogram-force per square centimetre ^b kgf/cm ²		98 066.5	0.980 665	980.665	0.967 841	1	14.223 3	735.559	28.959 0
1 pound-force per square inch lbf/in ²		6 894.76	0.068 947 6	68.947 6	0.068 046 0	0.070 307 0	1	51.714 9	2.036 02
1 torr ^c		133.322	0.001 333 22	1.333 22	1.315 79 × 10 ⁻³	1.359 51 × 10 ⁻³	0.019 336 8	1	0.039 370 1
1 (conventional) inch of mercury inHg		3 386.39	0.033 863 9	33.863 9	0.033 421 1	0.034 531 6	0.491 154	25.400 0	1
^a The abbreviation mb is used in meteorology. ^b This unit is equal to 10 (conventional) metres of water (10 mH ₂ O), see 33.2. ^c To within 1 part in 7 million, the torr is equal to the (conventional) millimetre of mercury (mmHg).									

Table 34 — Pressure (continued)

Exact values are printed in bold type

	pascal Pa (or newton per square metre N/m ²)	millibar mbar ^a	kilogram-force per square metre^b kgf/m ²	pound-force per square foot lbf/ft ²	(conventional) inch of water inH ₂ O	(conventional) foot of water ftH ₂ O	(conventional) millimetre of mercury mmHg	(conventional) inch of mercury inHg	
1 pascal = Pa	1 newton per square metre = N/m ²	1	0.01	0.101 972	$2.088\ 54 \times 10^{-2}$	$4.014\ 63 \times 10^{-3}$	$3.345\ 53 \times 10^{-4}$	$7.500\ 62 \times 10^{-3}$	$2.953\ 00 \times 10^{-4}$
1 millibar mbar ^a	=	100	1	10.197 2	2.088 54	0.401 463	0.033 455 3	0.750 062	0.029 530 0
1 kilogram-force per square metre ^b kgf/m ²	=	9.806 65	0.098 066 5	1	0.204 816	0.039 370 1	$3.280\ 84 \times 10^{-3}$	0.073 555 9	$2.895\ 90 \times 10^{-3}$
1 pound-force per square foot lbf/ft ²	=	47.880 3	0.478 803	4.882 43	1	0.192 222	0.016 018 5	0.359 131	0.014 139 0
1 (conventional) inch of water inH ₂ O	=	249.089	2.490 89	25.4	5.202 33	1	0.083 333 3	1.868 32	0.073 555 9
1 (conventional) foot of water ftH ₂ O	=	2 989.07	29.890 7	304.8	62.428 0	12	1	22.419 8	0.882 671
1 (conventional) millimetre of mercury mmHg	=	133.322	1.333 22	13.595 1	2.784 50	0.535 240	0.044 603 3	1	0.039 370 1
1 (conventional) inch of mercury inHg	=	3 386.39	33.863 9	345.316	70.726 2	13.595 1	1.132 92	25.4	1
^a The abbreviation mb is used in meteorology. ^b This unit is equal to the conventional millimetre of water (mmH ₂ O), see 33.2.									

35 Viscosity, dynamic (stress/velocity gradient)

35.1 The coherent SI unit of dynamic viscosity is the pascal second (Pa·s), which may also be expressed as the newton second per square metre (N·s/m²), or as the kilogram per metre second [kg/(m·s)].

This unit has also been called the poiseuille (Pl) in France. (It should be noted that this is not the same as the poise (P), described in **35.2**.)

35.2 The poise (P) is the corresponding CGS unit.

$$1 \text{ P} = 1 \text{ dyn}\cdot\text{s}/\text{cm}^2 = 10^{-1} \text{ N}\cdot\text{s}/\text{m}^2 = 10^{-1} \text{ Pa}\cdot\text{s}.$$

The commonly used submultiple is the centipoise (cP).

$$1 \text{ cP} = 10^{-2} \text{ P} = 10^{-3} \text{ Pa}\cdot\text{s}.$$

35.3 Other metric and imperial units that have been used for dynamic viscosity are:

kilogram-force second per square metre		(kgf·s/m ²)
poundal second per square foot	= pound per foot second, lb/(ft·s)	(pdl·s/ft ²)
pound-force second per square foot	= slug per foot second, slug/(ft·s)	(lbf·s/ft ²)
pound-force hour per square foot	= slug hour per foot second squared, slug·h/(ft·s ²)	(lbf·h/ft ²)
pound-force second per square inch		(lbf·s/in ²)
pound per foot hour		(lb/ft·h)

$$1 \text{ kgf}\cdot\text{s}/\text{m}^2 = 9.806 65 \text{ Pa}\cdot\text{s}$$

$$1 \text{ pdl}\cdot\text{s}/\text{ft}^2 = 1.488 16 \text{ Pa}\cdot\text{s}$$

$$1 \text{ lbf}\cdot\text{s}/\text{ft}^2 = 47.880 3 \text{ Pa}\cdot\text{s}$$

$$1 \text{ lbf}\cdot\text{h}/\text{ft}^2 = 1.723 69 \times 10^5 \text{ Pa}\cdot\text{s}$$

$$1 \text{ lbf}\cdot\text{s}/\text{in}^2 = 6 894.76 \text{ Pa}\cdot\text{s}$$

$$1 \text{ lb}/\text{ft}\cdot\text{h} = 4.133 79 \times 10^{-4} \text{ kg}/(\text{m}\cdot\text{s}) = 4.133 79 \times 10^{-4} \text{ Pa}\cdot\text{s}$$

NOTE 1 The pound-force second per square inch is sometimes called the “reyn”.

NOTE 2 For reference to frequently used but empirical units of viscosity, such as the Redwood second, see Clause 36, Viscosity, kinematic.

Interconversion factors for most of these units are given in Table 35.

Table 35 — Viscosity (dynamic)

Exact values are printed in bold type

		pascal second Pa·s	centipoise cP	kilogram-force second per square metre kgf·s/m ²	poundal second per square foot pdl·s/ft ²	pound-force second per square foot lbf·s/ft ²	pound-force hour per square foot lbf·h/ft ²
1 pascal second Pa·s	=	1	1 000	0.101 972	0.671 969	$2.088\ 54 \times 10^{-2}$	$5.801\ 51 \times 10^{-6}$
1 centipoise cP	=	0.001	1	$1.019\ 72 \times 10^{-4}$	$6.719\ 69 \times 10^{-4}$	$2.088\ 54 \times 10^{-5}$	$5.801\ 51 \times 10^{-9}$
1 kilogram-force second per square metre kgf·s/m ²	=	9.806 65	9 806.65	1	6.589 76	0.204 816	$5.689\ 34 \times 10^{-5}$
1 poundal second per square foot [=1 lb/(ft·s)] pdl·s/ft ²	=	1.488 16	1 488.16	0.151 750	1	0.031 081 0	$8.633\ 60 \times 10^{-6}$
1 pound-force second per square foot [=1 slug/(ft·s)] lbf·s/ft ²	=	47.880 3	47 880.3	4.882 43	32.174 0	1	$2.777\ 78 \times 10^{-4}$
1 pound-force hour per square foot lbf·h/ft ²	=	$1.723\ 69 \times 10^5$	$1.723\ 69 \times 10^8$	$1.757\ 67 \times 10^4$	$1.158\ 27 \times 10^5$	3 600	1
NOTE 1 lb/(ft·h) = $4.133\ 79 \times 10^{-4}$ kg/(m·s) = $4.133\ 79 \times 10^{-4}$ Pa·s (see 35.3).							

36 Viscosity, kinematic (length squared/time)

36.1 The coherent SI unit of kinematic viscosity (which is dynamic viscosity divided by density) is the metre squared per second (m^2/s).

36.2 The corresponding CGS unit is the stokes (St).

$$1 \text{ St} = 1 \text{ cm}^2/\text{s} = 10^{-4} \text{ m}^2/\text{s}$$

The common submultiple is the centistokes (cSt).

$$1 \text{ cSt} = 10^{-2} \text{ St} = 10^{-6} \text{ m}^2/\text{s} (= 1 \text{ mm}^2/\text{s})$$

36.3 Another metric unit sometimes used is the metre squared per hour (m^2/h).

$$1 \text{ m}^2/\text{h} = 2.777\ 78 \times 10^{-4} \text{ m}^2/\text{s}$$

36.4 A selection of imperial units is:

inch squared per second in^2/s

foot squared per second ft^2/s

inch squared per hour in^2/h

foot squared per hour ft^2/h

$$1 \text{ in}^2/\text{s} = 6.451\ 6 \times 10^{-4} \text{ m}^2/\text{s}$$

$$1 \text{ ft}^2/\text{s} = 9.290\ 30 \times 10^{-2} \text{ m}^2/\text{s}$$

$$1 \text{ in}^2/\text{h} = 1.792\ 11 \times 10^{-7} \text{ m}^2/\text{s}$$

$$1 \text{ ft}^2/\text{h} = 2.580\ 64 \times 10^{-5} \text{ m}^2/\text{s}$$

36.5 The units referred to in **36.1** to **36.4** are absolute units with physical dimensions, as distinct from values on frequently used but empirical scales such as the Redwood second, Saybolt Universal scale, and Engler degrees. For tables from which viscosity values in these empirical scales may be converted to centistokes see ESDU Item No. 68036 [4].

Interconversion factors for the units in 36.1 to 36.4 are given in Table 36.

Note that this table may be used for the conversion of values of thermal diffusivity, which also has the dimensions of length squared/time.

Table 36 — Viscosity (kinematic)

Exact values are printed in bold type

	metre squared per second m ² /s	centistokes cSt	inch squared per second in ² /s	foot squared per second ft ² /s	inch squared per hour in ² /h	foot squared per hour ft ² /h	metre squared per hour m ² /h
1 metre squared per second = m ² /s	1	1 × 10 ⁶	1.550 00 × 10 ³	10.763 9	5.580 01 × 10 ⁶	3.875 01 × 10 ⁴	3 600
1 centistokes = cSt	1 × 10 ⁻⁶	1	1.550 00 × 10 ⁻³	1.076 39 × 10 ⁻⁵	5.580 01	3.875 01 × 10 ⁻²	0.003 6
1 inch squared per second = in ² /s	6.451 6 × 10 ⁻⁴	645.16	1	6.944 44 × 10 ⁻³	3 600	25	2.322 58
1 foot squared per second = ft ² /s	9.290 30 × 10 ⁻²	92 903.0	144	1	518 400	3 600	334.451
1 inch squared per hour = in ² /h	1.792 11 × 10 ⁻⁷	0.179 211	2.777 78 × 10 ⁻⁴	1.929 01 × 10 ⁻⁶	1	6.944 44 × 10 ⁻³	6.451 6 × 10 ⁻⁴
1 foot squared per hour = ft ² /h	2.580 64 × 10 ⁻⁵	25.806 4	0.04	2.777 78 × 10 ⁻⁴	144	1	0.092 903 0
1 metre squared per hour = m ² /h	2.777 78 × 10 ⁻⁴	277.778	0.430 556	2.989 98 × 10 ⁻³	1 550.00	10.763 9	1

NOTE This table may also be used for the conversion of values of thermal diffusivity (area/time), for which the conversion factors employed are the same as those for kinematic viscosity.

37 Energy (work, heat, etc.)

37.1 General

37.1.1 The coherent SI unit for the expression of all forms of energy is the joule (symbol J). Just as energy arises in many ways, the connection between the joule and other SI units may be indicated in different ways, e.g:

$$\begin{aligned} 1 \text{ J} &= 1 \text{ N}\cdot\text{m} \text{ (force} \times \text{distance, newton metre)} \\ &= 1 \text{ W}\cdot\text{s} \text{ (electrical energy, watt second)} \\ &= 1 \text{ Pa}\cdot\text{m}^3 \text{ (pressure} \times \text{volume, pascal cubic metre)} \end{aligned}$$

This unit was, prior to the SI, known as the absolute joule, but it is now simply the joule (J).

The "international" joule, which became obsolete in 1948, was approximately equal to 1.000 19 J.

37.1.2 Arising from the historical development of the SI from the CGS system, the unit of energy in the CGS system (the erg) is decimally related to the joule.

$$\begin{aligned} 1 \text{ erg} &= 1 \text{ dyn}\cdot\text{cm} &= 1 \times 10^{-5} \text{ N} \times 0.01 \text{ m} \\ & &= 10^{-7} \text{ N}\cdot\text{m} = 10^{-7} \text{ J} \end{aligned}$$

37.1.3 A unit in extensive use for the expression of electrical energy is the kilowatt hour (kW·h).

$$\begin{aligned} 1 \text{ kW}\cdot\text{h} &= 1 \times 1\,000 \text{ W} \times 3\,600 \text{ s} \\ &= 3.6 \times 10^6 \text{ W}\cdot\text{s} = 3.6 \text{ MJ} \end{aligned}$$

37.1.4 Two other metric units used for the expression of energy are the kilogram-force metre²⁷⁾ (kgf·m) and the litre atmosphere.

$$\begin{aligned} 1 \text{ kgf}\cdot\text{m} &= 9.806\,65 \text{ N}\cdot\text{m} &= \mathbf{9.806\,65 \text{ J}} \\ 1 \text{ litre atmosphere} &= 1 \text{ dm}^3 \times 101\,325 \text{ Pa} &= 101.325 \text{ Pa}\cdot\text{m}^3 &= \mathbf{101.325 \text{ J}} \end{aligned}$$

[The litre used here is equal to 1 decimetre cubed (see 5.3 and 5.4), and the atmosphere used is the standard atmosphere (see 33.3).]

37.1.5 Some corresponding imperial units for the statement of energy are the foot poundal (ft·pdl), the foot pound-force (ft·lbf) and the horsepower hour (hp·h).

$$\begin{aligned} 1 \text{ ft}\cdot\text{pdl} &= 1 \times 0.304\,8 \text{ m} \times 0.453\,592\,37 \times 0.304\,8 \text{ N} \text{ (see 31.3)} \\ &= 0.042\,140\,1 \text{ J (approx.)} \\ 1 \text{ ft}\cdot\text{lbf} &= 1 \times 0.304\,8 \text{ m} \times 0.453\,592\,37 \times 9.806\,65 \text{ N} \text{ (see 31.3)} \\ &= 1.355\,82 \text{ J (approx.)} \\ 1 \text{ hp}\cdot\text{h} &= 550 \text{ ft lbf/s} \times 3\,600 \text{ s} \text{ (see 38.3)} \\ &= 1.98 \times 10^6 \text{ ft}\cdot\text{lbf} \\ &= 2.684\,52 \times 10^6 \text{ J (approx.)} \end{aligned}$$

²⁷⁾ Known as the kilopond metre (kp·m) in Germany.

37.2 Heat units

Heat is one of the forms of energy and, as stated in 37.1.1, the SI unit for all forms is the joule. The following heat units originally arose from the concept of the heat required to warm unit mass of water through unit temperature, but some of these are now precisely defined in terms of the joule:

- the various *calories* (originally relating to the gram of water and the degree Celsius);
- the various *British thermal units*, now obsolete (originally relating to the pound of water and degree Fahrenheit); and
- the various *Centigrade heat units* (based on the pound of water and the degree Celsius).

The specific heat capacity of water changes with temperature and a number of different calories, British thermal units, and Centigrade heat units came into use according to their means of definition.

Three of the calories, when used in precise work, need to be separately identified.

These are the International Table calorie (cal_{IT}), the thermochemical calorie (cal_{th}) and the 15 °C calorie (cal_{15}) described below:

$1 \text{ cal}_{\text{IT}} =$	4.186 8 J	(as defined at the Fifth International Conference on Properties of Steam, London 1956).
$1 \text{ cal}_{\text{th}} =$	4.184 0 J	(a “defined” calorie).
$1 \text{ cal}_{15} =$	4.185 5 J (approx.)	(This is defined as the amount of heat required to warm 1 g of air-free water from 14.5 °C to 15.5 °C at a constant pressure of 1 atm. The joule equivalent shown above was adopted by the CIPM in 1950 as being the most accurate value which could then be deduced from experiment.)

Associated with the cal_{15} are the thermie (th), also sometimes described as the “tonne-calorie” and the frigorie, used in connection with the extraction of heat.

$1 \text{ thermie} =$	10^6 cal_{15}	$=$	4.185 5 MJ (approx.)
$1 \text{ frigorie} =$	-10^3 cal_{15}	$=$	-4.185 5 kJ (approx.)

The “calorie” commonly referred to in nutritional science is in fact a kilocalorie, which is sometimes called a “kilogram-calorie” or “large calorie”. In this standard, if the symbol cal is used without qualification, it refers to the International Table calorie (cal_{IT}). (The dietitians calorie is based on the cal_{15} .) The use of the joule is strongly recommended, because of the wide variety of meanings of the “calorie”.

The British thermal unit (Btu) used in this standard is the one corresponding to the International Table calorie and it is defined by the equation:

$$1 \text{ Btu/lb} = \mathbf{2.326 \text{ J/g}}$$

Thus:

$$1 \text{ Btu} = 2.326 \times 453.592 \text{ 37 J} = \mathbf{1 \text{ 055.06 J (approx.)}}$$

Other British thermal units formerly in use but now obsolete are the following.

The “60 °F British thermal unit” (heat required to warm 1 lb of water from 60 °F to 61 °F).

$$1 \text{ Btu}_{60/61} = \mathbf{1 \text{ 054.5 J (approx.)}}$$

The “mean British thermal unit” (1/180 of the heat required to warm 1 lb of liquid water from 32 °F to 212 °F).

$$1 \text{ Btu}_{\text{mean}} = \mathbf{1 \text{ 055.8 J (approx.)}}$$

The British thermal unit once used for most purposes by the British Gas Industry relates to the 15 °C calorie and is equal to:

$$2.326 \times 453.592 \text{ 37 J} \times \frac{4.185 \text{ 5}}{4.186 \text{ 8}} = 1 \text{ 054.73 J (approx.)}$$

Associated with the Btu is the therm, once used as an energy unit by the Gas Industry.

$$1 \text{ therm} = 100 \text{ 000 Btu} = 105.5 \text{ MJ (approx.)}$$

The “Centigrade heat unit” (C.H.U.), based on the lb of water and the °C, is still sometimes used.

$$1 \text{ C.H.U.} = 1.8 \text{ Btu (but to each British thermal unit there corresponds a Centigrade heat unit)}$$

$$1 \text{ C.H.U.}_{\text{mean}} = 1.8 \text{ Btu}_{\text{mean}} = 1 \text{ 900.4 J (approx.)}$$

For interconversion factors for most of the above units see Table 37 and Table 38.

38 Power (energy/time)

38.1 The coherent SI unit for all forms of power, including heat flow rate, is the watt (symbol W), which is equal to the joule per second.

$$1 \text{ W} = 1 \text{ J/s}$$

The kilowatt (kW) is a commonly-used multiple of the watt.

38.2 Two metric technical units of power are the kilogram-force metre per second (kgf·m/s) and the metric horsepower²⁸⁾.

$$\begin{aligned} 1 \text{ kgf} \cdot \text{m/s} &= 9.806 \text{ 65 J/s} = 9.806 \text{ 65 W} \\ 1 \text{ metric horsepower} &= 75 \text{ kgf} \cdot \text{m/s} = 735.499 \text{ W} \end{aligned}$$

38.3 Similar technical units in the imperial system are the foot pound-force per second (ft·lbf/s) and the horsepower (hp).

$$\begin{aligned} 1 \text{ ft} \cdot \text{lbf/s} &= 1.355 \text{ 82 J/s (see 37.1.5)} = 1.355 \text{ 82 W} \\ 1 \text{ hp} &= 550 \text{ ft} \cdot \text{lbf/s} = 745.700 \text{ W} \end{aligned}$$

38.4 The following is a selection of heat flow units shown in terms of the watt (see 37.2):

$$\begin{aligned} \text{calorie per second} &1 \text{ cal/s} = 4.186 \text{ 8 W} \\ \text{kilocalorie per hour} &1 \text{ kcal/h} = 1.163 \text{ W} \\ \text{British thermal unit per hour} &1 \text{ Btu/h} = 0.293 \text{ 071 W} \\ \text{“ton of refrigeration”} &= 12 \text{ 000 Btu/h} = 3.516 \text{ 85 kW} \end{aligned}$$

For interconversion factors for most of the above units see Table 39.

²⁸⁾ The metric horsepower goes under the name “cheval vapeur” in France and sometimes the symbols ch or CV are used. In Germany it is called the “Pferdestärke” (symbol PS).

Table 37 — Energy

Exact values are printed in bold type

	joule J	kilowatt hour kW·h	kilogram-force metre kgf·m	litre ^a atmosphere	foot poundal ft·pdl	foot pound- force ft·lbf	horsepower hour hp·h
1 joule J	= 1	$2.777\ 78 \times 10^{-7}$	0.101 972	$0.986\ 923 \times 10^{-2}$	23.730 4	0.737 562	$3.725\ 06 \times 10^{-7}$
1 kilowatt hour kW·h	= 3.6×10^6	1	$3.670\ 98 \times 10^5$	$3.552\ 92 \times 10^4$	$8.542\ 93 \times 10^7$	$2.655\ 22 \times 10^6$	1.341 02
1 kilogram-force metre kgf·m	= 9.806 65	$2.724\ 07 \times 10^{-6}$	1	0.096 784 1	232.715	7.233 01	$3.653\ 04 \times 10^{-6}$
1 litre ^a atmosphere	= 101.325	$2.814\ 58 \times 10^{-5}$	10.332 3	1	2 404.48	74.733 5	$3.774\ 42 \times 10^{-5}$
1 foot poundal ft·pdl	= 0.042 140 1	$1.170\ 56 \times 10^{-8}$	0.004 297 10	$4.158\ 91 \times 10^{-4}$	1	0.031 081 0	$1.569\ 74 \times 10^{-8}$
1 foot pound-force ft·lbf	= 1.355 82	$3.766\ 16 \times 10^{-7}$	0.138 255	$1.338\ 09 \times 10^{-2}$	32.174 0	1	$5.050\ 51 \times 10^{-7}$
1 horsepower hour hp·h	= $2.684\ 52 \times 10^6$	0.745 700	$2.737\ 45 \times 10^5$	$2.649\ 41 \times 10^4$	$6.370\ 46 \times 10^7$	1.98 × 10 ⁶	1
^a The litre used here is equal to one decimetre cubed. (See 5.3.)							

Table 38 — Energy (continued)

Exact values are printed in bold type

	joule J	kilowatt hour kW·h	foot pound- force ft·lbf	horsepower hour hp·h	calorie ^a cal (cal _{IT})	thermochemical calorie cal _{th}	15 °C calorie cal ₁₅	British thermal unit Btu
1 joule J	1	$2.777\ 78 \times 10^{-7}$	0.737 562	$3.725\ 06 \times 10^{-7}$	0.238 846	0.239 006	0.238 920	$9.478\ 17 \times 10^{-4}$
1 kilowatt hour kW·h	3.6×10^6	1	$2.655\ 22 \times 10^6$	1.341 02	859 845	860 421	860 112	3 412.14
1 foot pound-force ft·lbf	1.355 82	$3.766\ 16 \times 10^{-7}$	1	$5.050\ 51 \times 10^{-7}$	0.323 832	0.324 048	0.323 932	$1.285\ 07 \times 10^{-3}$
1 horsepower hour hp·h	$2.684\ 52 \times 10^6$	0.745 700	1.98×10^6	1	641 186	641 616	641 386	2 544.43
1 calorie ^a cal (cal _{IT})	4.186 8	1.163×10^{-6}	3.088 03	$1.559\ 61 \times 10^{-6}$	1	1.000 67	1.000 31	$3.968\ 32 \times 10^{-3}$
1 thermochemical calorie cal _{th}	4.184	$1.162\ 22 \times 10^{-6}$	3.085 96	$1.558\ 57 \times 10^{-6}$	0.999 331	1	0.999 642	$3.965\ 67 \times 10^{-3}$
1 15 °C calorie cal ₁₅	4.185 5	$1.162\ 64 \times 10^{-6}$	3.087 07	$1.559\ 12 \times 10^{-6}$	0.999 690	1.000 36	1	$3.967\ 09 \times 10^{-3}$
1 British thermal unit Btu	1 055.06	$2.930\ 71 \times 10^{-4}$	778.169	$3.930\ 15 \times 10^{-4}$	251.996	252.164	252.074	1

^a This is the International Table calorie. For a description of the three calories mentioned here, see 37.2.
1 thermie (th) = 10⁶ cal₁₅.

Table 39 — Power

Exact values are printed in bold type

	watt	kilogram-force metre per second	metric horsepower	foot pound- force per second	horsepower	calorie ^a per second	kilocalorie ^a per hour	British thermal unit per hour
	W	kgf·m/s		ft·lbf/s	hp	cal/s	kcal/h	Btu/h
1 watt W	= 1	0.101 972	$1.359\ 62 \times 10^{-3}$	0.737 562	$1.341\ 02 \times 10^{-3}$	0.238 846	0.859 845	3.412 14
1 kilogram-force metre per second kgf·m/s	= 9.806 65	1	0.013 333 3	7.233 01	0.013 150 9	2.342 28	8.432 20	33.461 7
1 metric horsepower	= 735.499	75	1	542.476	0.986 320	175.671	632.415	2 509.63
1 foot pound-force per second ft·lbf/s	= 1.355 82	0.138 255	$1.843\ 40 \times 10^{-3}$	1	$1.818\ 18 \times 10^{-3}$	0.323 832	1.165 79	4.626 24
1 horsepower hp	= 745.700	76.040 2	1.013 87	550	1	178.107	641.186	2 544.43
1 calorie ^a per second cal/s	= 4.186 8	0.426 935	$5.692\ 46 \times 10^{-3}$	3.088 03	$5.614\ 59 \times 10^{-3}$	1	3.6	14.286 0
1 kilocalorie ^a per hour kcal/h	= 1.163	0.118 593	$1.581\ 24 \times 10^{-3}$	0.857 785	$1.559\ 61 \times 10^{-3}$	0.277 778	1	3.968 32
1 British thermal unit per hour Btu/h	= 0.293 071	$2.988\ 49 \times 10^{-2}$	$3.984\ 66 \times 10^{-4}$	0.216 158	$3.930\ 15 \times 10^{-4}$	0.069 998 8	0.251 996	1

^a This refers to the International Table calorie. See also 37.2.

39 Temperature, including temperature difference or interval

39.1 The SI unit of temperature is the kelvin (K). It is one of the base units of the SI and is defined as a specified fraction (1/273.16) of the thermodynamic temperature of the triple point²⁹⁾ of water. The kelvin is used for the expression of thermodynamic temperature, for which the datum is absolute zero; it can also be used for the expression of any temperature difference or temperature interval.

39.2 The temperature unit recognized for use in conjunction with the SI, is the degree Celsius (°C). The degree Centigrade (°C) was renamed the degree Celsius in 1948 by the 9th Conference of the CGPM mainly to prevent confusion with an angular measure, the centigrade (equal to one hundredth of a grade; see 8.3 and Clause 8, Note 1). The term “Centigrade”, although now incorrect, remains in widespread use for Celsius. The zero datum for Celsius temperature (0 °C) is now exactly defined by the thermodynamic temperature 273.16 K; formerly it was defined by the melting point of ice at 1 atm. The units of temperature difference, one degree Celsius and one kelvin, are exactly equal, by definition. In this sense:

$$1\text{ °C} = 1\text{ K}$$

and any temperature difference therefore has the same numerical value when expressed in °C as it has when expressed in K.

For formulae showing the interrelationships between Celsius temperatures and thermodynamic temperatures expressed in kelvins, and some other temperatures mentioned below, see Table 40.

39.3 Traditional in practical use in the UK and USA is Fahrenheit temperature, which has now been displaced in the UK by Celsius. The Fahrenheit scale is not formally defined, but it is generally recognized that:

32 °F is the ice point;

212 °F is the boiling point of water at 1 atm;

and that the unit of temperature difference one degree Fahrenheit (1 °F) is equal to five ninths of the unit of temperature difference the degree Celsius (1 °C). In this sense:

$$1\text{ °F} = \left(\frac{5}{9}\text{ °C}\right) = \left(\frac{5}{9}\text{ K}\right)$$

For formulae giving the interrelationship between Fahrenheit, Celsius, and other temperatures see Table 40.

39.4 For thermodynamic temperatures, the degree Rankine (°R) is still occasionally used. The unit interval of the degree Rankine is equal to 1 °F, a thermodynamic temperature of 0 °R being absolute zero. See Table 40.

39.5 In the 1959 edition of this standard, for *temperature interval* the letters “deg” were recommended, instead of the degree sign (°) which was reserved for *temperature*. In 1967-68, the 13th Conference of the CGPM considered the arguments for and against this practice and decided to recommend that the use of “deg” should be discontinued.

39.6 For the purpose of practical measurements the CIPM adopted in 1968 the “International Practical Temperature Scale of 1968”, IPTS - 68, based on reproducible fixed points and interpolation instruments and procedures. The IPTS - 68 was designed so that the International Practical Kelvin and Celsius temperatures closely approximate the kelvin and Celsius temperatures described in 39.1 and 39.2. The IPTS - 68 is defined only from a thermodynamic temperature of 13.81 K upwards. 13.81 K is the triple point of hydrogen. Some amendments were made subsequently in “The International Temperature Scale of 1990” (ITS-90). The triple point of water (273.16 K) is used as the defining point, instead of the freezing point of water (273.15 K). The scale now extends to lower temperatures, viz. 0.65 K, and the range of the platinum resistance thermometer as defining instrument has been extended from 630 °C up to the silver point 962 °C.

²⁹⁾ The temperature at the triple point of water, (where water, ice, and water vapour are in equilibrium) is very slightly removed from the temperature of the melting point of ice at atmospheric pressure (the ice point).

Table 40 — Equivalent values on four temperature scales

Exact values are printed in bold type

For the same temperature, if $[T]$ K, $[\theta]$ °C, $[t]$ °F and $[r]$ °R represent that temperature on the kelvin, Celsius, Fahrenheit and Rankine scales respectively, then the formulae relating the pure numbers $[T]$, $[\theta]$, $[t]$ and $[r]$ are as shown below.				
(kelvins)	$[T]$	$=$	$[\theta] + 273.15$	$= \frac{5}{9} ([t] + 459.67) = \frac{5}{9} [r]$
(degrees Celsius)	$[\theta]$	$=$	$[T] - 273.15$	$= \frac{5}{9} ([t] - 32) = \frac{5}{9} ([r] - 491.67)$
(degrees Fahrenheit)	$[t]$	$=$	$\frac{9}{5} [T] - 459.67$	$= \frac{9}{5} [\theta] + 32 = [r] - 459.67$
(degrees Rankine)	$[r]$	$=$	$\frac{9}{5} [T]$	$= \frac{9}{5} [\theta] + 491.67 = [t] + 459.67$
For the same temperature difference: (temperature difference in °C or K) = $\frac{5}{9}$ (temperature difference in °F or °R).				

40 Specific energy [(energy or heat)/mass]

40.1 There are several different terms for energy per unit mass which are used in different contexts, e.g.:

- specific enthalpy
- specific latent heat
- calorific value, mass basis

The SI unit for all such quantities is the joule per kilogram (J/kg).

40.2 Other units are:

kilocalorie per kilogram (kcal/kg) (see 37.2)

kilogram-force metre per kilogram (kgf·m/kg)

$$1 \text{ kcal}_{\text{T}}/\text{kg} = 4 \text{ 186.8 J/kg}$$

$$1 \text{ kcal}_{\text{th}}/\text{kg} = 4 \text{ 184 J/kg}$$

$$1 \text{ kcal}_{15}/\text{kg} = 4 \text{ 185.5 J/kg (approx.)}$$

$$1 \text{ kgf}\cdot\text{m}/\text{kg} = 9.806 \text{ 65 J/kg}$$

40.3 Corresponding imperial units are:

British thermal unit per pound (Btu/lb)

foot pound-force per pound (ft·lbf/lb)

$$1 \text{ Btu/lb} = 2 \text{ 326 J/kg}$$

$$1 \text{ ft}\cdot\text{lbf/lb} = 2.989 \text{ 07 J/kg (approx.)}$$

For interconversion factors see Table 41.

41 Heat content, volume basis (heat/volume)

(e.g. calorific value, volume basis)

41.1 The SI unit for this quantity, which is mainly used in connection with the combustion of gaseous or liquid fuels, is the joule per cubic metre (J/m^3). For most practical purposes either the kJ/m^3 or MJ/m^3 are suitable multiples.

41.2 Units that have been in common use are:

kilocalorie per cubic metre ($kcal/m^3$) (see **37.2** for the various calories)

thermie per litre (th/litre)

$$1 \text{ kcal}_{IT}/m^3 = 4 \text{ 186.8 } J/m^3$$

$$1 \text{ kcal}_{th}/m^3 = 4 \text{ 184 } J/m^3$$

$$1 \text{ kcal}_{15}/m^3 = 4 \text{ 185.5 } J/m^3$$

$$1 \text{ thermie/litre} = 4 \text{ 185.5 } \times 10^6 \text{ } J/m^3 \text{ (see } \mathbf{5.3} \text{ and } \mathbf{5.4} \text{ for litre)}$$

41.3 Corresponding imperial units are:

British thermal unit per cubic foot (Btu/ft^3)

therm per UK gallon (therm/UKgal)

$$1 \text{ Btu}/ft^3 = 37 \text{ 258.9 } J/m^3$$

$$1 \text{ therm}/UKgal = 2.320 \text{ 80 } \times 10^{10} \text{ } J/m^3$$

41.4 In **41.2** and **41.3** and in Table 42 it is assumed that, where gases are concerned, the volumes involved in the conversion have the same reference conditions of temperature, pressure and humidity. For some conversions of calorific value (volume basis) when the reference conditions are different, see Table 43 and Table 44.

Table 41 — Specific energy (e.g. calorific value, mass basis; specific latent heat; specific enthalpy)

Exact values are printed in bold type

	joule per kilogram J/kg	kilocalorie ^a per kilogram kcal _{IT} /kg	thermochemical kilocalorie per kilogram kcal _{th} /kg	15 °C kilocalorie per kilogram kcal ₁₅ /kg	British thermal unit per pound Btu/lb	foot pound-force per lb ft·lbf/lb	kilogram-force metre per kilogram kgf·m/kg
1 joule per kilogram J/kg	= 1	0.238 846 × 10 ⁻³	0.239 006 × 10 ⁻³	0.238 920 × 10 ⁻³	0.429 923 × 10 ⁻³	0.334 553	0.101 972
1 kilocalorie ^a per kilogram kcal _{IT} /kg	= 4 186.8	1	1.000 67	1.000 31	1.8	1 400.70	426.935
1 thermochemical kilocalorie per kilogram kcal _{th} /kg	= 4 184	0.999 331	1	0.999 642	1.798 80	1 399.77	426.649
1 15 °C kilocalorie per kilogram kcal ₁₅ /kg	= 4 185.5	0.999 690	1.000 36	1	1.799 44	1 400.27	426.802
1 British thermal unit per pound Btu/lb	= 2 326	0.555 556	0.555 927	0.555 728	1	778.169	237.186
1 foot pound-force per pound ft·lbf/lb	= 2.989 07	7.139 26 × 10 ⁻⁴	7.144 04 × 10 ⁻⁴	7.141 48 × 10 ⁻⁴	1.285 07 × 10 ⁻³	1	0.304 8
1 kilogram-force metre per kilogram kgf·m/kg	= 9.806 65	2.342 28 × 10 ⁻³	2.343 85 × 10 ⁻³	2.343 01 × 10 ⁻³	4.216 10 × 10 ⁻³	3.280 84	1
^a This is the International Table kilocalorie. For a description of the three calories mentioned here see 37.2 .							

Table 42 — Calorific value, volume basis

Exact values are printed in bold type

		joule per cubic metre J/m ³	kilocalorie^a per cubic metre kcal _{IT} /m ³	thermochemical kilocalorie per cubic metre kcal _{th} /m ³	15 °C kilocalorie per cubic metre kcal ₁₅ /m ³	British thermal unit per cubic foot^b Btu/ft ³	therm per UK gallon therm/UKgal	thermie per litre^c th/litre
1 joule per cubic metre J/m ³	=	1	0.238 846 × 10 ⁻³	0.239 006 × 10 ⁻³	0.238 920 × 10 ⁻³	26.839 2 × 10 ⁻⁶	4.308 86 × 10 ⁻¹¹	2.389 20 × 10 ⁻¹⁰
1 kilocalorie ^a per cubic metre kcal _{IT} /m ³	=	4 186.8	1	1.000 67	1.000 31	0.112 370	1.804 04 × 10 ⁻⁷	1.000 31 × 10 ⁻⁶
1 thermochemical kilocalorie per cubic metre kcal _{th} /m ³	=	4 184	0.999 331	1	0.999 642	0.112 295	1.802 83 × 10 ⁻⁷	0.999 642 × 10 ⁻⁶
1 15 °C kilocalorie per cubic metre kcal ₁₅ /m ³	=	4 185.5	0.999 690	1.000 36	1	0.112 335	1.803 47 × 10 ⁻⁷	1 × 10 ⁻⁶
1 British thermal unit per cubic foot ^b Btu/ft ³	=	37 258.9	8.899 15	8.905 10	8.901 91	1	1.605 44 × 10 ⁻⁶	8.901 91 × 10 ⁻⁶
1 therm per UK gallon therm/UKgal	=	2.320 80 × 10 ¹⁰	5.543 13 × 10 ⁶	5.546 84 × 10 ⁶	5.544 85 × 10 ⁶	6.228 83 × 10 ⁵	1	5.544 85
1 thermie per litre ^c th/litre	=	4 185.5 × 10 ⁶	0.999 690 × 10 ⁶	1.000 36 × 10 ⁶	1 × 10 ⁶	0.112 335 × 10 ⁶	0.180 347	1

NOTE In this table, where gases are concerned, it is assumed that the volumes involved in the conversion are measured under the same conditions of temperature, pressure and humidity. (See also Table 43 and Table 44.)

^a This is the International Table kilocalorie. For a description of the three calories mentioned see 37.2.

^b 1 therm/ft³ = 10⁵ Btu/ft³.

^c The litre here = 1 dm³. See 5.3.

Table 43 — Calorific value of gases, volume basis (with differing reference conditions)

	Btu/ft ³ (60 °F, 30 inHg wet)	kJ/m ³ (0 °C, 760 mmHg dry)	kJ/m ³ (15 °C, 760 mmHg dry)	kcal/m ³ (0 °C, 760 mmHg dry)	kcal/m ³ (0 °C, 760 mmHg wet)	kcal/m ³ (15 °C, 760 mmHg dry)	kcal/m ³ (15 °C, 760 mmHg wet)	kcal/m ³ (60 °F, 30 inHg wet)								
1 Btu/ft ³ (60 °F, 30 inHg, wet) =	1	39.972	37.891	9.547 1	9.489 5	9.050 1	8.898 0	8.899 1								
1 kJ/m ³ (0 °C, 760 mmHg, dry) =	0.025 018	<p>The values of the vapour pressure of water used in the computation of the factors given in this table are as follows:</p> <table style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th>Temperature</th> <th>Vapour pressure mmHg</th> </tr> </thead> <tbody> <tr> <td>0 °C</td> <td>4.581</td> </tr> <tr> <td>15 °C</td> <td>12.771</td> </tr> <tr> <td>60 °F</td> <td>13.235</td> </tr> </tbody> </table> <p>For definitions of pressure units mmHg and inHg see 33.2.</p>							Temperature	Vapour pressure mmHg	0 °C	4.581	15 °C	12.771	60 °F	13.235
Temperature	Vapour pressure mmHg															
0 °C	4.581															
15 °C	12.771															
60 °F	13.235															
1 kJ/m ³ (15 °C, 760 mmHg, dry) =	0.026 392															
1 kcal/m ³ (0 °C, 760 mmHg, dry) =	0.104 74															
1 kcal/m ³ (0 °C, 760 mmHg, wet) =	0.105 38															
1 kcal/m ³ (15 °C, 760 mmHg, dry) =	0.110 50															
1 kcal/m ³ (15 °C, 760 mmHg, wet) =	0.112 38															
1 kcal/m ³ (60 °F, 30 inHg, wet) =	0.112 37															
<p>NOTE 1 In this table the designation “wet” means “saturated with water vapour at the temperature stated”.</p> <p>NOTE 2 760 mmHg = 1 atm = 1 013.25 mbar = 101.325 kPa 30 inHg = 762 mmHg = 1 015.92 mbar = 101.592 kPa</p> <p>NOTE 3 For historical and legal reasons, the reference bases of conversion factors for calorific value once used by the United Kingdom Gas Industry differ in some respects from those in this table. (See Table 44 and the notes to that table.)</p>																

Table 44 — Conversion factors previously used by the UK Gas Industry

The following information and conversion factors were extracted from the booklet *SI Units and conversion factors for use in the British Gas Industry* issued by the Gas Council and Society of British Gas Industries, May 1972 edition.

The conversion factor for converting from British thermal units per cubic foot (measured at 60 °F, 30 inches Hg (@ 60 °F, latitude 53 °N) and saturated with water) to megajoules per standard cubic metre (measured at 15 °C 1 013.25 mbar and dry) is: 1 Btu/ft ³ = 0.037 96 MJ/m ³ 1 MJ/m ³ = 26.34 Btu/ft ³ Other conversion factors for various conditions are given in the table below. Whereas two bases for the British thermal unit are included, for practical purposes there is no significant difference between them. In the gas industry the British thermal unit was based on the 15° calorie.							
		MJ/m ³ 15 °C 1 013.25 mbar		Btu/ft ³ 60 °F, 30 inches Hg (IT calorie)		Btu/ft ³ 60 °F, 30 inches Hg (15 °C calorie)	
		Dry	Sat.	Dry	Sat.	Dry	Sat.
MJ/m ³ 15 °C 1 013.25 mbar	Dry	1	0.983 2	26.80	26.33	26.81	26.34
	Sat.	1.017	1	27.26	26.78	27.27	26.79
Btu/ft ³ 60 °F 30 inches Hg (International calorie basis)	Dry	0.037 31	0.036 69	1	0.982 6	1.000	0.982 9
	Sat.	0.037 97	0.037 34	1.018	1	1.018	1.000
Btu/ft ³ 60 °F 30 inches Hg (15° calorie basis)	Dry	0.037 30	0.036 67	0.999 7	0.982 3	1	0.982 6
	Sat.	0.037 96	0.037 32	1.017	0.999 7	1.018	1

NOTE (This note is for users of BS 350 and is not part of the GC/SBGI booklet.)

Attention is called to the following differences in the reference bases of Table 43 and Table 44.

- a) In Table 43 the only British thermal unit used is the one corresponding to the International Table calorie (see 37.2). In Table 44 this Btu is indicated by the parenthesis (International calorie basis) or (IT calorie), and factors relating to a Btu based on the 15 °C calorie are stated in the preamble above the table and included in the table.
- b) The pressure “30 inches Hg” as shown in, and under the conditions stated in, Table 44, is given as being equal to 1 013.740 5 mbar. The pressure 30 inHg shown in Table 43 (and as defined in 33.2) is, approximately, 1 015.916 6 mbar.

42 Specific heat capacity³⁰⁾ [heat/(mass × temperature interval)]

42.1 The SI unit of specific heat capacity is the joule per kilogram kelvin [J/(kg·K)].

42.2 The degree Celsius is often used in the expression of the above unit:

$$1 \text{ J}/(\text{kg} \cdot ^\circ\text{C}) = 1 \text{ J}/(\text{kg} \cdot \text{K})$$

This is also the case with other units including the kelvin, given in 42.3 and 42.4.

42.3 Other metric units are:

kilocalorie per kilogram kelvin [kcal/(kg·K)] (See 37.2 for the various calories.)

kilogram-force metre per kilogram kelvin [kgf·m/(kg·K)]

$$1 \text{ kcal}_{\text{IT}}/(\text{kg} \cdot \text{K}) = 4 \text{ 186.8 J}/(\text{kg} \cdot \text{K})$$

$$1 \text{ kcal}_{\text{th}}/(\text{kg} \cdot \text{K}) = 4 \text{ 184 J}/(\text{kg} \cdot \text{K})$$

$$1 \text{ kcal}_{15}/(\text{kg} \cdot \text{K}) = 4 \text{ 185.5 J}/(\text{kg} \cdot \text{K})$$

$$1 \text{ kgf} \cdot \text{m}/(\text{kg} \cdot \text{K}) = 9.806 \text{ 65 J}/(\text{kg} \cdot \text{K})$$

³⁰⁾ The older and simpler term “specific heat” referred to heat capacities, usually on a mass basis but sometimes on a volume basis. It is now preferred to reserve “specific” for the meaning “per unit mass”.

42.4 Corresponding imperial units are:

British thermal unit per pound degree Fahrenheit [Btu/(lb·°F)]

foot pound-force per pound degree Fahrenheit [ft·lbf/(lb·°F)]

$$1 \text{ Btu}/(\text{lb}\cdot^{\circ}\text{F}) = 4 \text{ 186.8 J}/(\text{kg}\cdot\text{K})$$

$$1 \text{ ft}\cdot\text{lbf}/(\text{lb}\cdot^{\circ}\text{F}) = 5.380 \text{ 32 J}/(\text{kg}\cdot\text{K})$$

For interconversion factors for the above units see Table 45.

43 Specific entropy [heat/(mass × thermodynamic temperature)]

Table 45 may also be used for the conversion of values of specific entropy, expressed in joules per kilogram kelvin, J/(kg·K), in kilocalories per kilogram kelvin, kcal/(kg·K), or in British thermal units per pound degree Rankine, Btu/(lb·°R).

44 Heat capacity, volume basis³¹⁾ [heat/(volume × temperature interval)]

44.1 The SI unit of heat capacity, volume basis, is the joule per cubic metre kelvin [J/(m³·K)].

44.2 The degree Celsius is often used in the expression of the above unit:

$$1 \text{ J}/(\text{m}^3\cdot^{\circ}\text{C}) = 1 \text{ J}/(\text{m}^3\cdot\text{K})$$

This is also the case with other units including the kelvin, given in **44.3** and **44.4**.

44.3 Other metric units are:

kilocalorie per cubic metre kelvin [kcal/(m³·K)] (See **37.2** for the various calories.)

$$1 \text{ kcal}_{\text{IT}}/(\text{m}^3\cdot\text{K}) = 4 \text{ 186.8 J}/(\text{m}^3\cdot\text{K})$$

$$1 \text{ kcal}_{\text{th}}/(\text{m}^3\cdot\text{K}) = 4 \text{ 184 J}/(\text{m}^3\cdot\text{K})$$

$$1 \text{ kcal}_{15}/(\text{m}^3\cdot\text{K}) = 4 \text{ 185.5 J}/(\text{m}^3\cdot\text{K})$$

44.4 The corresponding imperial unit is:

British thermal unit per cubic foot degree Fahrenheit [Btu/(ft³·°F)]

$$1 \text{ Btu}/(\text{ft}^3\cdot^{\circ}\text{F}) = 67 \text{ 066.1 J}/(\text{m}^3\cdot\text{K})$$

For interconversion factors for the above units see Table 46.

In Table 46 and in the above conversion factors it is assumed that, for gases, the volumes involved in the conversions are measured under the same conditions of temperature, pressure and humidity.

³¹⁾ This is sometimes known as “specific heat, volume basis” but see footnote to Clause **42**.

45 Heat flux density [heat/(area × time)]

45.1 The SI unit for this quantity, which is sometimes known as intensity of heat flow rate, and commonly appears, for example, in calculations of heat losses from surfaces, is the watt per square metre (W/m^2).

45.2 Other metric units are:

calorie per square centimetre second [$\text{cal}/(\text{cm}^2 \cdot \text{s})$] (See **37.2** for the various calories.)

kilocalorie per square metre hour [$\text{kcal}/(\text{m}^2 \cdot \text{h})$]

$$1 \text{ cal}_{\text{IT}}/(\text{cm}^2 \cdot \text{s}) = 41\,868 \text{ W}/\text{m}^2$$

$$1 \text{ kcal}_{\text{IT}}/(\text{m}^2 \cdot \text{h}) = 1.163 \text{ W}/\text{m}^2$$

45.3 Corresponding units in the imperial system are:

British thermal unit per square foot hour [$\text{Btu}/(\text{ft}^2 \cdot \text{h})$]

watt per square inch (W/in^2)

$$1 \text{ Btu}/(\text{ft}^2 \cdot \text{h}) = 3.154\,59 \text{ W}/\text{m}^2$$

$$1 \text{ W}/\text{in}^2 = 1\,550.00 \text{ W}/\text{m}^2$$

For conversion factors for the above see Table 47.

Table 45 — Specific heat, mass basis

Exact values are printed in bold type

	joule per kilogram kelvin ^a J/(kg·K)	kilocalorie ^b per kilogram kelvin kcal _{IT} /(kg·K)	thermochemical kilocalorie per kilogram kelvin kcal _{th} /(kg·K)	15 °C kilocalorie per kilogram kelvin kcal ₁₅ /(kg·K)	British thermal unit per pound degree Fahrenheit Btu/(lb·°F)	foot pound-force per pound degree Fahrenheit ft·lbf/(lb·°F)	kilogram-force metre per kilogram kelvin kgf·m/(kg·K)
1 joule per kilogram kelvin ^a J/(kg·K)	1	0.238 846 × 10 ⁻³	0.239 006 × 10 ⁻³	0.238 920 × 10 ⁻³	0.238 846 × 10 ⁻³	0.185 863	0.101 972
1 kilocalorie ^b per kilogram kelvin kcal _{IT} /(kg·K)	4 186.8	1	1.000 67	1.000 31	1	778.169	426.935
1 thermochemical kilocalorie per kilogram kelvin kcal _{th} /(kg·K)	4 184	0.999 331	1	0.999 642	0.999 331	777.649	426.649
1 15 °C kilocalorie per kilogram kelvin kcal ₁₅ /(kg·K)	4 185.5	0.999 690	1.000 36	1	0.999 690	777.928	426.802
1 British thermal unit per pound degree Fahrenheit Btu/(lb·°F)	4 186.8	1	1.000 67	1.000 31	1	778.169	426.935
1 foot pound-force per pound degree Fahrenheit ft·lbf/(lb·°F)	5.380 32	1.285 07 × 10 ⁻³	1.285 93 × 10 ⁻³	1.285 47 × 10 ⁻³	1.285 07 × 10 ⁻³	1	0.548 64
1 kilogram-force metre per kilogram kelvin kgf·m/(kg·K)	9.806 65	2.342 28 × 10 ⁻³	2.343 85 × 10 ⁻³	2.343 01 × 10 ⁻³	2.342 28 × 10 ⁻³	1.822 69	1

NOTE This table may also be used for interconversion of values of specific entropy e.g. J/(kg·K), kcal/(kg·K), Btu/(lb·°R).

^a Wherever the kelvin (K) occurs in this table it may be replaced by the degree Celsius (°C) e.g. J/(kg·K) is often shown as J/(kg·°C).

^b This is the International Table kilocalorie. For a description of the three calories mentioned see **37.2**.

Table 46 — Heat capacity, volume basis

Exact values are printed in bold type

	joule per cubic metre kelvin ^a J/(m ³ ·K)	kilocalorie ^b per cubic metre kelvin kcal _{IT} /(m ³ ·K)	thermochemical kilocalorie per cubic metre kelvin kcal _{th} /(m ³ ·K)	15 °C kilocalorie per cubic metre kelvin kcal ₁₅ /(m ³ ·K)	British thermal unit per cubic foot degree Fahrenheit Btu/(ft ³ ·°F)
1 joule per cubic metre kelvin ^a J/(m ³ ·K)	1	0.238 846 × 10 ⁻³	0.239 006 × 10 ⁻³	0.238 920 × 10 ⁻³	14.910 7 × 10 ⁻⁶
1 kilocalorie ^b per cubic metre kelvin kcal _{IT} /(m ³ ·K)	4 186.8	1	1.000 67	1.000 31	0.062 428 0
1 thermochemical kilocalorie per cubic metre kelvin kcal _{th} /(m ³ ·K)	4 184	0.999 331	1	0.999 642	0.062 386 2
1 15 °C kilocalorie per cubic metre kelvin kcal ₁₅ /(m ³ ·K)	4 185.5	0.999 690	1.000 36	1	0.062 408 6
1 British thermal unit per cubic foot degree Fahrenheit Btu/(ft ³ ·°F)	67 066.1	16.018 5	16.029 2	16.023 4	1

NOTE In this table it is assumed that, for gases, the volumes involved in the conversions are measured under the same conditions of temperature, pressure and humidity.

^a Wherever the kelvin occurs in this table it may be replaced by the degree Celsius (°C), e.g. J/(m³·K) is often shown as J/(m³·°C).
^b This is the International Table kilocalorie. For a description of the three calories mentioned see 37.2.

Table 47 — Heat flux density, intensity of heat flow rate (e.g. heat loss from surfaces)

Exact values are printed in bold type

	W/m ²	W/in ²	cal _{IT} /(cm ² ·s)	kcal _{IT} /(m ² ·h)	Btu/(ft ² ·h)
1 watt per square metre W/m ²	1	6.451 6 × 10 ⁻⁴	0.238 846 × 10 ⁻⁴	0.859 845	0.316 998
1 watt per square inch W/in ²	1 550.00	1	3.702 12 × 10 ⁻²	1 332.76	491.348
1 calorie ^a per square centimetre second cal _{IT} /(cm ² ·s)	41 868	27.011 6	1	36 000	13 272.1
1 kilocalorie ^a per square metre hour kcal _{IT} /(m ² ·h)	1.163	7.503 21 × 10 ⁻⁴	2.777 78 × 10 ⁻⁵	1	0.368 669
1 British thermal unit per square foot hour Btu/(ft ² ·h)	3.154 59	2.035 22 × 10 ⁻³	7.534 61 × 10 ⁻⁵	2.712 46	1

^a This refers to the International Table calorie. For other calories see 37.2.

46 Thermal conductance (heat transfer coefficient)

[heat/(area × time × temperature difference)] or [power/(area × temperature difference)]³²⁾

46.1 The SI unit is the watt per square metre kelvin [$W/(m^2 \cdot K)$]

46.2 The degree Celsius ($^{\circ}C$) is often used in the expression of the above unit:

$$1 W/(m^2 \cdot ^{\circ}C) = 1 W/(m^2 \cdot K)$$

This is also the case with other units including the kelvin, given in **46.3** and **46.4**.

46.3 Other metric units are:

calorie per square centimetre second kelvin [$cal/(cm^2 \cdot s \cdot K)$]

kilocalorie per square metre hour kelvin [$kcal/(m^2 \cdot h \cdot K)$]

(The conversion factors given below refer to the International Table calorie; see **37.2** for other calories.)

$$1 cal/(cm^2 \cdot s \cdot K) = 41\,868 W/(m^2 \cdot K)$$

$$1 kcal/(m^2 \cdot h \cdot K) = 1.163 W/(m^2 \cdot K)$$

46.4 The imperial unit is:

British thermal unit per square foot hour degree Fahrenheit [$Btu/(ft^2 \cdot h \cdot ^{\circ}F)$]

$$1 Btu/(ft^2 \cdot h \cdot ^{\circ}F) = 5.678\,26 W/(m^2 \cdot K)$$

For interconversion factors for the above units see Table 48.

Table 48 — Thermal conductance

Exact values are printed in bold type

	$W/(m^2 \cdot K)^a$	$cal^b/(cm^2 \cdot s \cdot K)$	$kcal^b/(m^2 \cdot h \cdot K)$	$Btu/(ft^2 \cdot h \cdot ^{\circ}F)$
1 watt per square metre kelvin ^a $W/(m^2 \cdot K)$	= 1	$0.238\,846 \times 10^{-4}$	0.859 845	0.176 110
1 calorie ^b per square centimetre second kelvin $cal/(cm^2 \cdot s \cdot K)$	= 41 868	1	36 000	7 373.38
1 kilocalorie ^b per square metre hour kelvin $kcal/(m^2 \cdot h \cdot K)$	= 1.163	$2.777\,78 \times 10^{-5}$	1	0.204 816
1 British thermal unit per square foot hour degree Fahrenheit $Btu/(ft^2 \cdot h \cdot ^{\circ}F)$	= 5.678 26	$1.356\,23 \times 10^{-4}$	4.882 43	1

^a Wherever the kelvin occurs in this table it may be replaced by the degree Celsius: e.g. $W/(m^2 \cdot K)$ is often shown as $W/(m^2 \cdot ^{\circ}C)$
^b This refers to the International Table calorie. For other calories see **37.2**.

³²⁾ Also corresponds to (heat flux density/temperature difference).

47 Thermal conductivity [heat × length/(area × time × temperature difference)]

47.1 The SI unit is the watt per metre kelvin [W/(m·K)]

47.2 The degree Celsius (°C) is often used in the expression of the above unit:

$$1 \text{ W}/(\text{m}\cdot^{\circ}\text{C}) = 1 \text{ W}/(\text{m}\cdot\text{K})$$

This is also the case with other units including the kelvin, given in **47.3** and **47.4**.

47.3 Other metric units are:

calorie per centimetre second kelvin [cal/(cm·s·K)]

kilocalorie per metre hour kelvin [kcal/(m·h·K)]

(The conversion factors given below refer to the International Table calorie; see **37.2** for other calories.)

$$1 \text{ cal}/(\text{cm}\cdot\text{s}\cdot\text{K}) = \mathbf{418.68} \text{ W}/(\text{m}\cdot\text{K})$$

$$1 \text{ kcal}/(\text{m}\cdot\text{h}\cdot\text{K}) = \mathbf{1.163} \text{ W}/(\text{m}\cdot\text{K})$$

47.4 Two imperial units are:

British thermal unit per foot hour degree Fahrenheit [Btu/(ft·h·°F)]

British thermal unit inch per square foot hour degree Fahrenheit [Btu·in/(ft²·h·°F)]

$$1 \text{ Btu}/(\text{ft}\cdot\text{h}\cdot^{\circ}\text{F}) = 1.730\ 73 \text{ W}/(\text{m}\cdot\text{K})$$

$$1 \text{ Btu}\cdot\text{in}/(\text{ft}^2\cdot\text{h}\cdot^{\circ}\text{F}) = 0.144\ 228 \text{ W}/(\text{m}\cdot\text{K})$$

For interconversion factors for the above units see Table 49.

Table 49 — Thermal conductivity

Exact figures are printed in bold type

	W/(m·K) ^a	cal/(cm·s·K) ^b	kcal/(m·h·K)	Btu/(ft·h·°F)	Btu·in/(ft ² ·h·°F)
1 watt per metre kelvin ^a W/(m·K)	1	0.238 846 × 10 ⁻²	0.859 845	0.577 789	6.933 47
1 calorie ^b per centimetre second kelvin cal/(cm·s·K)	418.68	1	360	241.909	2 902.91
1 kilocalorie ^a per metre hour kelvin kcal/(m·h·K)	1.163	2.777 78 × 10 ⁻³	1	0.671 969	8.063 63
1 British thermal unit per foot hour degree Fahrenheit Btu/(ft·h·°F)	1.730 73	4.133 79 × 10 ⁻³	1.488 16	1	12
1 British thermal unit inch per square foot hour degree Fahrenheit Btu·in/(ft ² ·h·°F)	0.144 228	3.444 82 × 10 ⁻⁴	0.124 014	0.083 333 3	1

^a Wherever the kelvin occurs in this table it may be replaced by the degree Celsius (°C) e.g. W/(m·K) is often shown as W/(m·°C).
^b This refers to the International Table calorie. For other calories see **37.2**.

48 Thermal resistivity [area × time × temperature difference/(heat × length)]

The SI unit of thermal resistivity (the inverse of thermal conductivity) is the metre kelvin per watt ($\text{m}\cdot\text{K}/\text{W}$).

Interconversion factors between the above and some other units are given in Table 50.

Similar comments concerning the use of the degree Celsius and the other calories apply as in Clause 47 and Table 49.

Table 50 — Thermal resistivity

Exact figures are printed in bold type

	$\text{m}\cdot\text{K}/\text{W}$	$\text{cm}\cdot\text{s}\cdot\text{K}/\text{cal}^{\text{a}}$	$\text{m}\cdot\text{h}\cdot\text{K}/\text{kcal}^{\text{a}}$	$\text{ft}\cdot\text{h}\cdot^{\circ}\text{F}/\text{Btu}$	$\text{ft}^2\cdot\text{h}\cdot^{\circ}\text{F}/(\text{Btu}\cdot\text{in})$
$1 \text{ m}\cdot\text{K}/\text{W} =$	1	418.68	1.163	1.730 73	0.144 228
$1 \text{ cm}\cdot\text{s}\cdot\text{K}/\text{cal}^{\text{a}} =$	$0.238\ 846 \times 10^{-2}$	1	$2.777\ 78 \times 10^{-3}$	$4.133\ 79 \times 10^{-3}$	$3.444\ 82 \times 10^{-4}$
$1 \text{ m}\cdot\text{h}\cdot\text{K}/\text{kcal}^{\text{a}} =$	0.859 845	360	1	1.488 16	0.124 014
$1 \text{ ft}\cdot\text{h}\cdot^{\circ}\text{F}/\text{Btu} =$	0.577 789	241.909	0.671 969	1	0.083 333 3
$1 \text{ ft}^2\cdot\text{h}\cdot^{\circ}\text{F}/(\text{Btu}\cdot\text{in}) =$	6.933 47	2 902.91	8.063 63	12	1
NOTE For thermal conductivity, see Table 49, the notes to which also apply here.					
^a This refers to the International Table calorie. For other calories see 37.2.					

49 Heat release rate (e.g. as used in connection with furnaces) [heat/(volume × time)], or (power/volume)

49.1 The SI unit for this quantity is the watt per cubic metre (W/m^3).

49.2 Other metric units are:

calorie per cubic centimetre second [$\text{cal}/(\text{cm}^3 \cdot \text{s})$]

kilocalorie per cubic metre hour [$\text{kcal}/(\text{m}^3 \cdot \text{h})$]

The conversion factors given below refer to the International Table calorie. (See 37.2 for other calories.)

$$1 \text{ cal}/(\text{cm}^3 \cdot \text{s}) = 4.186 8 \times 10^6 \text{ W}/\text{m}^3$$

$$1 \text{ kcal}/(\text{m}^3 \cdot \text{h}) = 1.163 \text{ W}/\text{m}^3$$

49.3 A similar imperial unit is:

British thermal unit per cubic foot hour [$\text{Btu}/(\text{ft}^3 \cdot \text{h})$]

$$1 \text{ Btu}/(\text{ft}^3 \cdot \text{h}) = 10.349 7 \text{ W}/\text{m}^3$$

For interconversion factors for the above units see Table 51.

Table 51 — Heat release rate

Exact values are printed in bold type

	watt per cubic metre W/m^3	calorie ^a /cubic centimetre second $\text{cal}/(\text{cm}^3 \cdot \text{s})$	kilocalorie ^a /cubic metre hour $\text{kcal}/(\text{m}^3 \cdot \text{h})$	British thermal unit/cubic foot hour $\text{Btu}/(\text{ft}^3 \cdot \text{h})$
1 watt per cubic metre W/m^3	1	$0.238 846 \times 10^{-6}$	0.859 845	$9.662 11 \times 10^{-2}$
1 calorie ^a per cubic centimetre second $\text{cal}/(\text{cm}^3 \cdot \text{s})$	$4.186 8 \times 10^6$	1	3.6×10^6	$4.045 33 \times 10^5$
1 kilocalorie ^a per cubic metre hour $\text{kcal}/(\text{m}^3 \cdot \text{h})$	1.163	$2.777 78 \times 10^{-7}$	1	0.112 370
1 British thermal unit per cubic foot hour $\text{Btu}/(\text{ft}^3 \cdot \text{h})$	10.349 7	$2.471 99 \times 10^{-6}$	8.899 15	1

^a This refers to the International Table calorie. For other calories see 37.2.

1 $\text{W}/\text{cm}^3 = 10^6 \text{ W}/\text{m}^3 = 1 \text{ MW}/\text{m}^3$.

50 Thermal diffusivity (area/time)

The SI unit of thermal diffusivity (which is thermal conductivity divided by heat capacity per unit volume) is the metre squared per second (m^2/s).

Since kinematic viscosity has the same dimensions as thermal diffusivity, for units and conversion factors reference can be made to Clause 36 and Table 36.

Annex A (informative)

Commentary on imperial and metric systems of measurement and units

A.1 Development of units

In the past, units have evolved in a haphazard manner to meet the basic measurement requirements of early and often unconnected societies. With improvement in communications and extension of trade it became necessary to standardize the units in use and also to establish the relationship between existing units used to measure the same physical quantities. Often, as this latter process developed, the numerical factors relating one such unit to another were cumbersome and difficult to use in calculations (for example, the mile is 1 760 yards and the UK ton is 2 240 pounds). Moreover, while one physical quantity might be a simple derivative of another, there was often no correspondingly simple relationship between their respective units, (for example, area and volume are simple derivatives of length, but 1 acre is 4 840 square yards and 1 UK gallon is 0.160 544 cubic feet).

As science and technology developed, many new and complex units were required. Inevitably these were derived from the available units in common usage and the result was a muddled conglomeration of technical units involving many awkward factors which were difficult to remember and inconvenient to use. The learning of these factors was once a necessary part of scientific and engineering education.

A.2 Unit systems and coherence

The various physical quantities used in science and technology are related to one another by certain mathematical or physical laws. For example, area equals length multiplied by length, velocity equals length divided by time, force equals mass multiplied by acceleration, momentum equals mass multiplied by velocity.

In a coherent system of units, the units used to measure the various physical quantities are consistent with these physical laws. A minimum number of independent physical quantities are arbitrarily selected and base units are defined for these. Units for all other physical quantities can then be derived in accordance with the physical laws, preserving a unity relationship in terms of the base units. Thus, if unit area results when unit length is multiplied by unit length, the units are coherent with the particular physical law expressing the relationship between length and area, and no factors are involved in calculations concerned with this relationship.

With further development of science and technology, certain “systems” of units came into use, (for example the foot-pound-second system and the centimetre-gram-second system). While the base units concerned were clearly defined, the total extent of each of these systems and also units for some physical quantities were in certain respects vague. The units comprising these systems were coherent with respect to some of the physical laws, but not to others.

A.3 The metric system of measurement

The SI (Système International d’Unités) is now the most widely used source of units in the UK. The SI is the latest development of the metric system. Only a few imperial measures are still permitted.

In Britain, units of length, weight (mass) and so on have been standardized for a long time, but prior to the adoption of the metric system this was not the case in France and in other mainland European countries. When the metric system was introduced it met two main requirements. The first was the standardization and definition of the important units of measurement, the metre for length and the gram for mass, from which other units then required for general use and for trade were derived. The second was the provision of a convenient and systematic relationship between different-sized units for the same quantity. These were related by powers of ten and a system of prefixes developed to indicate these powers. This gave a flexible means of expression for a wide range of magnitudes, avoiding the need for very large or very small numerical values, and enabled the different-sized units to be memorized and converted with ease.

In technology, probably the most widely used metric system is one in which the base units for length, mass and time are the metre, kilogram and second respectively, but in which there is a non-coherent relationship between the units for mass, force and acceleration. The unit of force used is the kilogram-force (sometimes described as a “metric technical unit of force”), and because this force acting on a mass of 1 kilogram produces an acceleration of “*g*” (9.81 m/s² approximately) the factor 9.81 is introduced in an awkward manner into many engineering calculations. The SI has now largely superseded this system.

There are other metric systems still in use in some sectors of industry and science, which are dynamically coherent, and from the first two of which the development of the SI can directly be traced. Some are shown in Table A.1.

Table A.1 — Other metric systems

Description and abbreviation	mass unit	force unit
centimetre-gram-second (CGS)	gram	dyne
metre-kilogram-second (MKS)	kilogram	newton
metre-tonne-second (MTS)	tonne	sthène

Scientists were quick to recognize the convenience of the metric approach in the CGS system and this system was developed by them to meet their immediate needs, according to their knowledge at the time, and among other things it served in the development of electrostatics and electromagnetism. Although it gained considerable usage in industrial technology, many of the associated units were inconveniently sized for this purpose. It was also clear that more than the three base units provided in the CGS system were required in the framework of the metric system to deal adequately with the physical quantities required in science and technology. Also, some of the subsidiary units that had come into use in conjunction with the CGS system were not coherent.

These factors led in due course to the evolution of the MKS system, thence to the MKSA system, incorporating the independent quantity electric current and the base unit ampere. This system embodied the joule as the derived and coherent unit of energy in all its forms, and the watt as the unit of power.

A.4 The International System of units (SI)

This “modern metric system” expanded on the MKS system to include a total of seven base units and two supplementary units which, in conjunction with derived units, meet all known needs for a coherent system of units both in science and technology.

The base and supplementary quantities, and their units (defined in BS 5555:1993) are shown in Table A.2.

Table A.2 — Base and supplementary quantities, units and symbols in the SI system

Quantity	Name of unit	Symbol
Base		
Length	metre	m
Mass	kilogram	kg
Time	second	s
Electric current	ampere	A
Thermodynamic temperature	kelvin	K
Amount of substance	mole	mol
Luminous intensity	candela	cd
Supplementary		
Plane angle	radian	rad
Solid angle	steradian	sr

There are, for practical applications or for everyday life, certain other units, some metric and some non-metric, which at present are authorized for use in conjunction with the International System. Such units are listed in categories in BS 5555:1993 and their use introduces an element of non-coherence.

The metric prefixes, which now form part of the International System, are shown in detail in 2.1. As is evident from the foregoing description of a coherent system, the use of multiples in the form of a prefix also introduces non-coherence, but in the SI the prefixed units still retain a simple decimal relationship one with the other. This is an important feature, which is not sacrificed by the fact that the base unit for mass is the *kilogram*. Recent discussions were aimed at renaming the kilogram to avoid the use of the prefix, but it now appears no change will be made in the immediate future.

A.5 Imperial systems

Before the advent of metric systems in the UK, the most widely used British system was one in which the base units for length, mass and time were the foot, pound and second respectively. But, in this system there is a non-coherent relationship between the units used for mass, force and acceleration i.e. there is not “dynamic coherence”. The unit of force used is the pound-force (sometimes described as a “technical unit of force”), and, because this force acting on a mass of one pound produces an acceleration of “*g*” ($= 32.2 \text{ ft/s}^2$ approximately), the factor 32.2 was introduced in an awkward manner into many engineering calculations.

There are two other systems based on imperial units formerly used in some sections of industry which are dynamically coherent. The first is a variant of the foot-pound-second system which has the poundal as its force unit. The poundal acting on a mass of one pound produces an acceleration of unity (1 ft/s^2). The other is the foot-slug-second system in which the mass unit is the slug ($= 32.2 \text{ lb}$ approximately) and the force unit the pound-force. Again, the acceleration produced by the pound-force on the slug is unity.

In the above, only dynamic coherence has been mentioned. While this is of vital importance in mechanics, there are many other important physical quantities and laws; further base units had to be introduced and the corresponding units that came about in conjunction with these imperial systems were frequently non-coherent. Furthermore, in dealing with the foot-pound-second, and foot-slug-second systems, there are the practical complications in calculations caused by the awkward relationships between the foot, inch and yard, and the pound and ton. In the measurement sense these units all form part of the imperial system.

A.6 Comparison of United Kingdom (UK or imperial) and United States systems of measurement

The yard has the same value in both the UK and US systems and is defined in terms of the SI base unit of length, the metre. Similarly, the pound has the same value in the UK and US systems and is defined in terms of the SI base unit of mass, the kilogram.

The UK Weights and Measures Act, 1985 [1], defines these units in the UK. In the USA the same definitions are valid for all purposes except for coast and geodetic surveys within the USA, for which the foot previously adopted there will continue temporarily to be used under the name “US survey foot”.

Most of the subsidiary units of length are identical in both the UK and US systems. There are marked differences between some subsidiary units of mass used in the UK and US systems, notably in the “long” and “short” tons and hundredweights. These differences arise from different whole number relationships between units.

There are also marked differences between the subsidiary units of capacity used in the UK and US systems. These differences arise both from some different whole number relationships between units and also from different definitions of capacity in the two systems.

In order to avoid confusion where those differences occur, the units to be distinguished are denoted by the use of prefixes, for example:

UK gallon, symbolized by UKgal

US gallon, symbolized by USgal

This notation is similar to one adopted by the International Organization for Standardization (ISO). In certain contexts the qualification “imperial” or “imp” is also used to make it clear that the unit qualified by UK does in fact belong to the imperial system of units.

Index of symbols and abbreviations

Symbol or abbreviation	Name of unit or prefix, where appropriate	Textual reference
A	ampere	A.4
Å	ångström	3.3
a	are	4.3
a	atto (prefix)	2.1
a	year	Clause 10, Note 2
at	technical atmosphere	33.3
ata	technical atmosphere (absolute, German)	33.4
atm	standard atmosphere	33.3
atü	technical atmosphere (gauge, German)	33.4
AU	astronomical unit	3.3
bar	bar	33.1.2
bbl	barrel (US, for petroleum)	5.10
bbl (dry)	dry barrel (US)	5.11
Btu	British thermal unit	37.2
Btu _{mean}	mean British thermal unit	37.2
Btu _{60/61}	60 °F British thermal unit	37.2
bu	bushel (US)	5.9, 5.11
c	centi (prefix)	2.1
c	cycle	13.1
°C	degree Celsius	39.2
cal	calorie	37.2
cal _{IT}	International Table calorie	37.2
cal _{th}	thermochemical calorie	37.2
cal ₁₅	15 °C calorie	37.2
cd	candela	A.4
ch	cheval vapeur (metric horsepower) (French)	38.2, footnote
C.H.U.	Centigrade heat unit	37.2
cl	centilitre	5.3
cm	centimetre	3.2
cP	centipoise	35.2
cSt	centistokes	36.2
ctl	cental	15.5
cumec	cubic metre per second	24.1, footnote
cusec	cubic foot per second	24.3, footnote
CV	cheval vapeur (metric horsepower) (French)	38.2, footnote
cwt	hundredweight	15.5
d	day	10.3
d	deci (prefix)	2.1
da	deca (prefix)	2.1

Symbol or abbreviation	Name of unit or prefix, where appropriate	Textual reference
deg	(to indicate temperature interval)	39.5
dm	decimetre	3.2
dr	dram (avoirdupois)	15.5
dry qt	dry quart (US)	5.11
dyn	dyne	29.2
E	exa (prefix)	2.1
erg	erg	37.1.2
°F	degree Fahrenheit	39.3
f	femto (prefix)	2.1
f	fermi	3.3
fl dr	fluid drachm (UK)	5.8
fl dr	fluid dram (US)	5.10
fl oz	fluid ounce	5.8
fm	fermi	3.3
Fm	Festmeter (German)	5.5
ft	foot	3.5
ftH ₂ O	conventional foot of water	33.2
G	giga (prefix)	2.1
g	gram	15.2
g	acceleration due to gravity	14.4
g _n	standard acceleration due to gravity ("standard gravity")	14.4
g	grade	8.3
Gal	galileo (or gal)	14.2
gal	gallon	5.8
gi	gill (US)	5.10
gon	gon (German)	8.3
gr	grain	15.5
h	hecto (prefix)	2.1
h	hour	10.3
ha	hectare	4.3
hbar	hectobar	Clause 34
hl	hectolitre	5.3
hp	horsepower	38.3
hp·h	horsepower hour	37.1.5
Hz	hertz	13.1
in	inch	3.5
inHg	conventional inch of mercury	33.2
inH ₂ O	conventional inch of water	33.2

Symbol or abbreviation	Name of unit or prefix, where appropriate	Textual reference
J	joule	37.1.1
K	kelvin	39.1
k	kilo (prefix)	2.1
kg	kilogram	15.1
kgf	kilogram-force	29.2
kip	1 000 pounds-force (US)	29.3
km	kilometre	3.2
kn	knot (international)	11.5
kp	kilopond (kilogram-force, German)	29.2
k.s.i.	kips per square inch (US)	33.1.4
kW	kilowatt	38.1
kW·h	kilowatt hour	37.1.3
l	litre	5.3
lb	pound	15.4
lbf	pound-force	29.3
liq dr	liquid dram (US)	5.10, footnote
liq oz	liquid ounce (US)	5.10, footnote
liq pt	liquid pint (US)	5.10
liq qt	liquid quart (US)	5.10
l.y.	light year	3.3
M	mega (prefix)	2.1
m	metre	3.1
m	milli (prefix)	2.1
mb	millibar	33.1.2, footnote
mil	(of area)	4.5, and Clause 4, Note 2
mil	(of angle)	Clause 8, Note 2
mil	(of length)	3.6, and Clause 3, Note 8
mil	(of volume)	5.5
Mg	megagram	15.2
mg	milligram	15.2
mGal	milligal	14.2
mH ₂ O	conventional metre of water	33.2
min	minute (of time)	10.3
min	minim	5.8
ml	millilitre	5.3
mm	millimetre	3.2
mmHg	conventional millimetre of mercury	33.2
mmH ₂ O	conventional millimetre of water	33.2
mol	mole	21.1, A.4
ms	millisecond	Clause 10, Note 1

Symbol or abbreviation	Name of unit or prefix, where appropriate	Textual reference
N	newton	29.1
n	nano (prefix)	2.1
n mile	nautical mile (international)	3.3
ns	nanosecond	Clause 10, Note 1
oz	ounce	15.5
oz ap	apothecaries' ounce (US)	15.5
oz apoth	apothecaries' ounce (UK)	15.5
ozf	ounce-force	29.3
oz t	ounce troy (US)	15.5
oz tr	ounce troy (UK)	15.5
P	peta (prefix)	2.1
P	poise	35.2
p	pico (prefix)	2.1
Pa	pascal	33.1
pc	parsec	3.3
pdl	poundal	29.3
pk	peck (US)	5.11
Pl	poiseuille (French)	35.1
PS	Pferdestärke (metric horsepower) (German)	38.2, footnote
p.s.i.	pound-force per square inch	33.1.4
p.s.i.a.	pound-force per square inch (absolute)	33.4
p.s.i.g.	pound-force per square inch (gauge)	33.4
pt	pint	5.8
pz	pièze (French)	33.1.2
q	quintal	15.3
qr	quarter	15.5
qt	quart	5.8
°R	degree Rankine	39.4
r	revolution	12.2
rad	radian	8.1
rev	revolution	12.2
reyn	(viscosity unit)	35.3, Note 1
Rm	Raummeter (German)	5.5
s	second (of time)	10.1
sh cwt	short hundredweight (US)	15.5
sh ton	short ton (US)	15.5
sn	sthène (French)	29.2
sr	steradian	Clause 9
St	stokes	36.2
st	stère (French)	5.5

Symbol or abbreviation	Name of unit or prefix, where appropriate	Textual reference
T	tera (prefix)	2.1
t	tonne	15.2
th	thermie	37.2
ton	ton (UK)	15.5
tonf	ton-force	29.3
u	atomic mass unit	15.3
UKgal	gallon (UK)	5.8
UKpt	pint (UK)	5.8
UKqt	quart (UK)	5.8
USgal	gallon (US)	5.10
W	watt	38.1
y	yocto (prefix)	2.1
Y	yotta (prefix)	2.1
yd	yard	3.4, 3.5
z	zepto (prefix)	2.1
Z	zetta (prefix)	2.1
°	degree (of angle)	8.2
°	degree (temperature)	Clause 39
'	minute (of angle)	8.2
"	second (of angle)	8.2
γ	gamma (microgram)	Clause 15, Note 1
μ	micro (prefix)	2.1
μ	micron (micrometre) (abrogated)	3.2
μ	micron (micrometre of mercury) (incorrect)	33.2
μg	microgram	15.2
μin	micro-inch	3.6
μl	microlitre	5.3
μm	micrometre	3.2
μmHg	micrometre of mercury	33.2
μs	microsecond	Clause 10, Note 1
⊥	right angle	8.1

Index of terms

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
absolute pressure	—	33.4	—
acceleration	—	Clause 14	Table 14
acceleration, standard	—	14.4	—
acceleration due to gravity	g	14.4	—
acceleration due to gravity, standard	g_n	14.4	Table 14
acre	—	4.4	Table 4
acre per pound	acre/lb	18.3	Table 20
ampere	A	A.4	—
angle, plane	—	Clause 8	Table 11
angle, right	\perp	8.1	Table 11
angle, solid	—	Clause 9	—
ångström	Å	3.3	—
angular momentum	—	Clause 28	—
angular velocity	—	Clause 12	Table 13
apothecaries' units	—	15.5	—
are	a	4.3	—
area	—	Clause 4	—
area, first moment of	—	Clause 6	Table 6
area, second moment of	—	Clause 7	Table 10
area per unit capacity	—	Clause 19	Table 21
area per unit mass	—	Clause 18	Table 20
areic mass	—	Clause 17	Table 19
assay ton (UK)	—	15.6 and Clause 15 , Note 3	—
assay ton (US)	—	15.6 and Clause 15 , Note 4	—
astronomical unit	AU	3.3	—
atmosphere, standard	atm	33.3	Table 33
atmosphere, technical	at	33.3	Table 32
atmosphere, technical (absolute, German)	ata	33.4	—
atmosphere, technical (gauge, German)	atü	33.4	—
atomic mass unit	u	15.3	—
atto (prefix)	a	2.1	Table 1
avoirdupois units	—	15.5	Table 15, Table 16, Table 17
bar	bar	33.1.2	Table 33
barn	—	4.3	—
barrel (beer, UK)	—	5.8	—
barrel (cranberry, US)	—	Clause 5 , Note 2	—
barrel (wine, UK)	—	Clause 5 , Note 1	—
barrel (petroleum, US)	bbl	5.10	—
barrel, dry (US)	bbl (dry)	5.11	—
barye	—	33.1.2	—
billion	—	2.3	Table 2
board foot	—	Clause 5 , Note 4	—
British thermal unit	Btu	37.2	Table 38
British thermal unit, mean	Btu _{mean}	37.2	—
British thermal unit, 60 °F	Btu _{60/61}	37.2	—

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
British thermal unit inch per square foot hour degree Fahrenheit	Btu·in/(ft ² ·h·°F)	47.4	Table 49
British thermal unit per cubic foot	Btu/ft ³	41.3	Table 42, Table 43, Table 44
British thermal unit per cubic foot degree Fahrenheit	Btu/(ft ³ ·°F)	44.4	Table 46
British thermal unit per cubic foot hour	Btu/(ft ³ ·h)	49.3	Table 51
British thermal unit per foot hour degree Fahrenheit	Btu/(ft·h·°F)	47.4	Table 49
British thermal unit per hour	Btu/h	38.4	Table 39
British thermal unit per pound	Btu/lb	40.3	Table 41
British thermal unit per pound degree Fahrenheit	Btu/(lb·°F)	42.4	Table 45
British thermal unit per pound degree Rankine	Btu/(lb·°R)	Clause 43	Table 45, Note
British thermal unit per square foot hour	Btu/(ft ² ·h)	45.3	Table 47
British thermal unit per square foot hour degree Fahrenheit	Btu/(ft ² ·h·°F)	46.4	Table 48
bushel (UK)	—	5.8 and Clause 5, Note 3	Table 6, Table 9
bushel (US)	bu	5.9, 5.11 and Clause 5, Note 3	Table 6, Table 9
bushel, international corn	—	15.6 and Clause 15, Note 6	—
cable-length	—	3.6 and Clause 3, Note 16	—
calendar year	—	10.4, footnote	—
calorie	cal	37.2	Table 38
calorie, dietitians'	—	see 37.2	use Table 38
calorie, International Table	cal _{IT}	37.2	Table 38
calorie, kilogram-	—	see 37.2	use Table 38
calorie, thermochemical	cal _{th}	37.2	Table 38
calorie, tonne-	—	see 37.2	use Table 38
calorie, 15 °C	cal ₁₅	37.2	Table 38
calorie per centimetre second kelvin	cal/(cm·s·K)	47.3	Table 49
calorie per cubic centimetre second	cal/(cm ³ ·s)	49.2	Table 51
calorie per second	cal/s	38.4	Table 39
calorie per square centimetre second	cal/(cm ² ·s)	45.2	Table 47
calorie per square centimetre second kelvin	cal/(cm ² ·s·K)	46.3	Table 48
calorific value, gases, volume basis, with differing reference conditions	—	41.4	Table 43, Table 44
calorific value, mass basis	—	40.1	Table 41
calorific value, volume basis	—	Clause 41	Table 42, Table 43, Table 44
candela	cd	A.4	—
capacity	—	Clause 5	Table 6, Table 7, Table 8, Table 9
carat, metric	CM (see Clause 15, Note 2)	15.3	Table 16
Celsius, degree	°C	39.2	Table 40
cental	ctl	15.5	—

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
centi (prefix)	c	2.1	Table 1
Centigrade	—	<i>see</i> 39.2	—
Centigrade heat unit	C.H.U.	37.2	—
centilitre	cl	5.3	<i>use</i> Table 6, Table 7, Table 8
centimetre	cm	3.2	<i>use</i> Table 3
centimetre cubed (modulus of section)	cm ³	6.2	<i>use</i> Table 6
centimetre per second squared	cm/s ²	14.2	<i>use</i> Table 14
centimetre second kelvin per calorie	cm·s·K/cal	—	Table 50
centimetre to the fourth	cm ⁴	Clause 7	Table 10
centipoise	cP	35.2	Table 35
centistokes	cSt	36.2	Table 36
chain	—	3.5	Table 3
chain, engineer's	—	3.6	—
chain, Gunter's	—	3.5 and Clause 3 , Note 5	Table 3
cheval vapeur (metric horsepower, French)	CV, ch	38.2 , footnote	Table 39
coefficient, heat transfer	—	Clause 46	Table 48
concentration, mass	—	Clause 21	Table 23
conductance, thermal	—	Clause 46	Table 48
conductivity, thermal	—	Clause 47	Table 49
cord	—	Clause 5 , Note 4	—
corn bushel, international	—	15.6 and Clause 15 , Note 6	—
cran	—	Clause 5 , Note 5	—
cubic centimetre	cm ³	5.2	Table 8
cubic decimetre	dm ³	5.2	Table 6, Table 7
cubic foot	ft ³	5.6	Table 6, Table 7
cubic foot per hour	ft ³ /h	24.3	Table 26
cubic foot per pound	ft ³ /lb	22.3	Table 24
cubic foot per second	ft ³ /s	24.3	Table 26
cubic foot per UK ton	ft ³ /UK ton	22.3	Table 24
cubic inch	in ³	5.6	Table 6, Table 7, Table 8
cubic inch per pound	in ³ /lb	22.3	Table 24
cubic metre	m ³	5.1	Table 6, Table 7
cubic metre per hour	m ³ /h	24.2	Table 26
cubic metre per kilogram	m ³ /kg	22.1	Table 24
cubic metre per second	m ³ /s	24.1	Table 26
cubic millimetre	mm ³	5.2	Table 8, Note
cubic yard	yd ³	5.6	Table 6
cumec	—	24.1 , footnote	—
cusec	—	24.3 , footnote	—
cycle	c	13.1	—
cycle per second	c/s	13.1	—
day	d	10.3	—
deca (prefix)	da	2.1	Table 1

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
deci (prefix)	d	2.1	Table 1
decimetre	dm	3.2	use Table 3
deg	deg	39.5	—
degree Celsius	°C	39.2	Table 40
degree Fahrenheit	°F	39.3	Table 40
degree (of angle)	°	8.2	Table 11
degree per minute	°/min	12.2	Table 13
degree per second	°/s	12.2	Table 13
degree Rankine	°R	39.4	Table 40
denier	—	16.2	—
density	—	Clause 20	Table 22
density, linear	—	Clause 16	Table 18
density, relative	—	Clause 20, footnote	—
diffusivity, thermal	—	Clause 50	see Table 36
drachm (apothecaries', UK)	—	15.5	Table 16
drachm, fluid (UK)	UK fl dr	5.8	Table 8, Table 9
dram (apothecaries', US)	—	15.5	—
dram (avoirdupois)	dr	15.5	Table 16
dram, fluid (US)	fl dr	5.10	Table 9
dram, liquid (US)	liq dr	5.10, footnote	—
dynamic viscosity	—	Clause 35	Table 35
dyne	dyn	29.2	Table 30, Note
dyne per centimetre	dyn/cm	32.2	—
dyne per square centimetre	dyn/cm ²	33.1.2	—
em	—	3.6 and Clause 3, Note 13	—
energy	—	Clause 37	Table 37, Table 38
energy, specific	—	Clause 40	Table 41
engineer's chain	—	3.6	—
Engler degrees	—	36.5	—
enthalpy, specific	—	40.1	Table 41
entropy, specific	—	Clause 43	see Table 45
ephemeris second	—	10.2	—
erg	erg	37.1.2	—
exa (prefix)	E	2.1	Table 1
Fahrenheit, degree	°F	39.3	Table 40
fathom	—	3.6	Table 3
femto (prefix)	f	2.1	Table 1
fermi	f, fm	3.3	—
Festmeter (German)	Fm	5.5	—
firkin	—	5.8	—
flow rate, mass	—	Clause 23	Table 25
flow rate, volume	—	Clause 24	Table 26
flux density, heat	—	Clause 45	Table 47
foot	ft	3.5	Table 3
foot, board	—	Clause 5, Note 4	—
foot, US survey	—	3.6	—
foot cubed (modulus of section)	ft ³	6.3	use Table 6

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
foot hour degree Fahrenheit per British thermal unit	ft·h·°F/Btu	—	Table 50
foot hour degree Fahrenheit per British thermal unit inch	ft·h·°F/(Btu·in)	—	Table 50
foot of water	ftH ₂ O	33.2	Table 34
foot per minute	ft/min	11.4	Table 12
foot per second	ft/s	11.4	Table 12
foot per second squared	ft/s ²	14.3	Table 14
foot poundal	ft·pdl	37.1.5	Table 37
foot pound-force	ft·lbf	37.1.5	Table 37, Table 38
foot pound-force per pound	ft·lbf/lb	40.3	Table 41
foot pound-force per pound degree Fahrenheit	ft·lbf/(lb·°F)	42.4	Table 45
foot pound-force per second	ft·lbf/s	38.3	Table 39
foot squared per hour	ft ² /h	36.4	Table 36
foot squared per second	ft ² /s	36.4	Table 36
foot to the fourth	ft ⁴	Clause 7	Table 10
force	—	Clause 29	Table 30
force per unit length	—	Clause 32	—
frequency	—	Clause 13	—
frigorie	—	37.2	<i>use</i> Table 38
furlong	—	3.5	Table 3
gal	Gal	14.2	Table 14, Note
galileo	Gal	14.2	Table 14, Note
gallon (UK)	UKgal	5.8	Table 7, Table 9
gallon (UK) per hour	UKgal/h	24.3	Table 26
gallon (UK) per mile	UKgal/mile	<i>see</i> Clause 25	Table 27
gallon (UK) per minute	UKgal/min	24.3	Table 26
gallon (UK) per pound	UKgal/lb	22.3	Table 24
gallon (UK) per second	UKgal/s	24.3	Table 26
gallon (US)	USgal	5.9, 5.10	Table 7, Table 9
gallon (US) per mile	USgal/mile	—	Table 27
gamma	γ	Clause 15, Note 1	—
gauge pressure	—	33.4	—
geometrical moment of inertia	—	Clause 7	Table 10
giga (prefix)	G	2.1	Table 1
gill (UK)	—	5.8	Table 9
gill (US)	gi	5.10	Table 9
Gon (German)	gon	8.3	Table 11
grade	g	8.3	Table 11
grain	gr	15.5	Table 16
grain per cubic foot	gr/ft ³	21.3	Table 23
grain per UK gallon	gr/UKgal	21.3	Table 23
grain per US gallon	gr/USgal	21.3	Table 23
gram	g	15.2	Table 16
gram centimetre squared	g·cm ²	26.2	Table 29, Note
gram per cubic centimetre	g/cm ³	20.2	Table 22
gram per cubic decimetre	g/dm ³	21.2	Table 23
gram per litre	g/l	21.2	Table 23
gram per millilitre	g/ml	20.2	Table 22

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
gram per square metre	g/m ²	17.2	Table 19, footnote
gravity, specific	—	Clause 20, footnote	—
gravity, standard	g_n	14.4	Table 14
gross ton (US)	—	15.5	—
hand	—	3.6 and Clause 3, Note 14	—
heat	—	37.2	Table 38
heat capacity (volume basis)	—	Clause 44	Table 46
heat content (volume basis)	—	Clause 41	Table 42, Table 43, Table 44
heat flow rate, intensity of	—	45.1	Table 47
heat flux density	—	Clause 45	Table 47
heat release rate	—	Clause 49	Table 51
heat transfer coefficient	—	Clause 46	Table 48
hectare	ha	4.3	Table 4
hectare per kilogram	ha/kg	18.2	Table 20
hecto (prefix)	h	2.1	Table 1
hectobar	hbar	Clause 34	Table 32
hectolitre	hl	5.3	see Table 7, Table 8
hertz	Hz	13.1	—
hogshead	—	5.8	—
horsepower	hp	38.3	Table 39
horsepower, metric	(see 38.2, footnote)	38.2	Table 39
horsepower hour	hp·h	37.1.5	Table 37, Table 38
hour	h	10.3	—
hundredweight	cwt	15.5	Table 17
hundredweight, long (US)	—	15.5	—
hundredweight, short (US)	sh cwt	15.5	Table 17
imperial system, commentary on	—	Annex A	—
inch	in	3.5	Table 3
inch cubed (modulus of section)	in ³	6.3	use Table 6
inch of mercury, conventional	inHg	33.2	Table 33 Table 34
inch of water, conventional	inH ₂ O	33.2	Table 34
inch per second	in/s	11.4	Table 12
inch squared per hour	in ² /h	36.4	Table 36
inch squared per second	in ² /s	36.4	Table 36
inch to the fourth	in ⁴	Clause 7	Table 10
inertia, geometrical moment of	—	Clause 7	Table 10
inertia, moment of	—	Clause 26	Table 29
intensity of heat flow rate	—	45.1	Table 47
international corn bushel	—	15.6 and Clause 15, Note 6	—
international nautical mile	n mile	3.3	Table 3
International Practical Temperature Scale of 1968	—	39.6	—
International Temperature Scale of 1990	—	39.6	—
inverse second	s ⁻¹	13.1, 13.2	—

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
IPTS - 68	—	39.6	—
ITS - 90	—	39.6	—
iron	—	3.6 and Clause 3 , Note 10	—
joule	J	37.1.1	Table 37, Table 38
joule, absolute	J	37.1.1	—
joule, international	—	37.1.1	—
joule per cubic metre	J/m ³	41.1	Table 42, <i>see</i> Table 43, Table 44
joule per cubic metre degree Celsius	J/(m ³ ·°C)	44.2	<i>use</i> Table 46
joule per cubic metre kelvin	J/(m ³ ·K)	44.1	Table 46
joule per kilogram	J/kg	40.1	Table 41
joule per kilogram degree Celsius	J/(kg·°C)	42.2	<i>use</i> Table 45
joule per kilogram kelvin	J/(kg·K)	42.1	Table 45
kelvin	K	39.1	Table 40
kilderkin	—	5.8	—
kilo (prefix)	k	2.1	Table 1
kilocalorie per cubic metre	kcal/m ³	41.2	Table 42, Table 43
kilocalorie per cubic metre hour	kcal/(m ³ ·h)	49.2	Table 51
kilocalorie per cubic metre kelvin	kcal/(m ³ ·K)	44.3	Table 46
kilocalorie per hour	kcal/h	38.4	Table 39
kilocalorie per kilogram	kcal/kg	40.2	Table 41
kilocalorie per kilogram kelvin	kcal/(kg·K)	42.3	Table 45
kilocalorie per metre hour kelvin	kcal/(m·h·K)	47.3	Table 49
kilocalorie per square metre hour	kcal/(m ² ·h)	45.2	Table 47
kilocalorie per square metre hour kelvin	kcal/(m ² ·h·K)	46.3	Table 48
kilogram	kg	15.1	Table 15
kilogram-calorie	—	<i>see</i> 37.2	<i>use</i> Table 38
kilogram-force	kgf	29.2	Table 30
kilogram-force metre (energy)	kgf·m	37.1.4	Table 37
kilogram-force metre (torque)	kgf·m	31.2	Table 31
kilogram-force metre per kilogram	kgf·m/kg	40.2	Table 41
kilogram-force metre per kilogram kelvin	kgf·m/(kg·K)	42.3	Table 45
kilogram-force metre per second	kgf·m/s	38.2	Table 39
kilogram-force per square centimetre	kgf/cm ²	33.1.3	Table 32, Table 33
kilogram-force per square metre	kgf/m ²	33.1.3	Table 34
kilogram-force per square millimetre	kgf/mm ²	—	Table 32, footnote
kilogram-force second per square metre	kgf·s/m ²	35.3	Table 35
kilogram metre squared	kg·m ²	26.1	Table 29
kilogram millimetre squared	kg·mm ²	26.2	Table 29, footnote
kilogram per cubic metre (density)	kg/m ³	20.1	Table 22
kilogram per cubic metre (concentration)	kg/m ³	21.2	Table 23
kilogram per hectare	kg/ha	17.2	Table 19
kilogram per hour	kg/h	23.2	Table 25
kilogram per metre	kg/m	16.1	Table 18
kilogram per metre second	kg/(m·s)	35.1	<i>see</i> Table 35
kilogram per second	kg/s	23.1	Table 25

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
kilogram per square metre	kg/m ²	17.1	Table 19
kilojoule	kJ	<i>see</i> 37.2	<i>see</i> Table 43
kilometre	km	3.2	<i>see</i> Table 3
kilometre per hour	km/h	11.3	Table 12
kilometre per litre	km/l	<i>see</i> Clause 25	Table 28
kilopascal	kPa	33.1.1	<i>see</i> Table 32, Table 33, Table 34
kilopond	kp	29.2	Table 30
kilopond metre (energy)	kp·m	37.1.4 , footnote	—
kilopond metre (torque)	kp·m	31.2	Table 31, footnote
kilopond per square centimetre	kp/cm ²	33.1.3	Table 32, footnote
kilopond per square metre	kp/m ²	<i>see</i> 33.1.3	—
kilowatt	kW	38.1	<i>use</i> Table 39
kilowatt hour	kW·h	37.1.3	Table 37, Table 38
kinematic viscosity	—	Clause 36	Table 36
kip (US)	—	29.3	Table 30, footnote
knot (international)	kn	11.5	Table 12
knot (UK)	—	11.5	Table 12
latent heat, specific	—	40.1	Table 41
length	—	Clause 3	Table 3
light year	l.y.	3.3	—
ligne	—	3.6 and Clause 3 , Note 12	—
line	—	3.6 and Clause 3 , Note 11 and Note 12	—
linear density	—	Clause 16	Table 18
linear velocity	—	Clause 11	Table 12
lineic mass	—	Clause 16	Table 18
link	—	3.6	—
litre	l	5.3	Table 6, Table 7, Table 8, footnote
litre (1901)	—	5.3	Table 6, Table 7, Table 8, footnote
litre atmosphere	—	37.1.4	Table 37
litre per hour	l/h	24.2	Table 26
litre per hundred kilometres	l/100 km	Clause 25	—
litre per kilogram	l/kg	22.2	Table 24
litre per kilometre	l/km	Clause 25	Table 27
litre per mile	—	<i>see</i> Clause 25	Table 27
litre per minute	l/min	24.2	Table 26
litre per second	l/s	24.2	Table 26
long hundredweight (US)	—	15.5	—
long ton (US)	—	15.5	—
mass	—	Clause 15	Table 15, Table 16, Table 17
mass, areic	—	Clause 17	Table 19
mass, lineic	—	Clause 16	Table 18

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
mass, volumic	—	Clause 20	Table 22
mass concentration	—	Clause 21	Table 23
mass per unit area	—	Clause 17	Table 19
mass per unit length	—	Clause 16	Table 18
mass rate of flow	—	Clause 23	Table 25
mass unit, atomic	u	15.3	—
mega (prefix)	M	2.1	Table 1
megagram	Mg	15.2	Table 17
megajoule	MJ	see 37.2	see Table 44
megapascal	MPa	33.1.1	Table 32
metre	m	3.1	Table 3
metre cubed (modulus of section)	m ³	6.1	use Table 6
metre hour kelvin per kilocalorie	m·h·K/kcal	—	Table 50
metre kelvin per watt	m·K/W	Clause 48	Table 50
metre of water, conventional	mH ₂ O	33.2	Table 32, footnote
metre per second	m/s	11.1	Table 12
metre per second squared	m/s ²	14.1	Table 14
metre squared per hour	m ² /h	36.3	Table 36
metre squared per second	m ² /s	36.3, Clause 50	Table 36
metre to the fourth	m ⁴	Clause 7	Table 10
metric carat	CM (see Clause 15, Note 2)	15.3	Table 16
metric horsepower	(see 38.2, footnote)	38.2	Table 39
metric system, commentary on	—	Annex A	—
metric ton	—	15.2	—
micro (prefix)	μ	2.1	Table 1
microgram	μg	15.2	use Table 16
micro-inch	μin	3.6	use Table 3
microlitre	μl	5.3	use Table 8
micrometre	μm	3.2	use Table 3
micron (length)	—	see 3.2	—
micron (pressure)	μmHg	33.2	use Table 34
microsecond	μs	Clause 10, Note 1	—
mil, circular (area)	—	4.5 and Clause 4, Note 2	Table 5
mil (angle)	mil	Clause 8, Note 2	—
mil (length)	mil	3.6 and Clause 3, Note 8	Table 2
mil (volume)	—	5.5	—
mile	mile	3.5	Table 3
mile, international nautical	n mile	3.3	Table 3
mile (statute)	mile	3.5 and Clause 3, Note 6	Table 3
mile, telegraph nautical	—	3.6	—
mile, UK nautical	—	3.6 and Clause 3, Note 16	Table 3
mile per gallon (UK)	mile/UKgal	Clause 25	Table 28
mile per gallon (US)	mile/USgal	see Clause 25	Table 28
mile per hour	mile/h	11.4	Table 12
milli (prefix)	m	2.1	Table 1
millibar	mbar (mb)	33.1.2 and footnote	Table 33, Table 34

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
milligal	mGal	14.2	Table 14, Note
milligram	mg	15.2	use Table 16
milligram per square centimetre	mg/cm ²	17.2	Table 19, footnote
milligram per square millimetre	mg/mm ²	17.2	Table 19, footnote
millilitre	ml	5.3	Table 8
millimetre	mm	3.2	use Table 3
millimetre cubed (modulus of section)	mm ³	6.2	use Table 6
millimetre of mercury, conventional	mmHg	33.2	Table 34
millimetre of water, conventional	mmH ₂ O	33.2	Table 34, footnote
millimetre to the fourth	mm ⁴	Clause 7	Table 10
million	—	2.2	Table 2
millisecond	ms	Clause 10, Note 1	—
minim (UK)	UKmin	5.8	Table 8, Table 9
minim (US)	—	5.10	Table 9
minute (angle)	'	8.2	Table 11
minute (time)	min	10.3	—
modulus of section	—	Clause 6	Table 6
mole	mol	21.1, A.4	—
moment of area, first	—	Clause 6	Table 6
moment of area, second	—	Clause 7	Table 10
moment of force	—	Clause 31	Table 31
moment of inertia	—	Clause 26	Table 29
moment of inertia, geometrical	—	Clause 7	Table 10
momentum, angular	—	Clause 28	—
momentum (linear)	—	Clause 27	—
month	—	10.4	—
nano (prefix)	n	2.1	Table 1
nanosecond	ns	Clause 10, Note 1	—
nautical mile, international	n mile	3.3	Table 3
nautical mile, telegraph	—	3.6	—
nautical mile, UK	—	3.6	Table 3
newton	N	29.1	Table 30
newton metre	N·m	31.1, 37.1.1	Table 31
newton per metre	N/m	32.1	—
newton per square metre	N/m ²	33.1.1	Table 32, Table 33, Table 34
newton per square millimetre	N/mm ²	Clause 34	Table 32
newton second per square metre	N·s/m ²	35.1	use Table 35
number	—	Clause 2	Table 1, Table 2
ounce (apothecaries' UK)	oz apoth	15.5	Table 16
ounce (apothecaries' US)	oz ap	15.5	Table 16
ounce (avoirdupois)	oz	15.5	Table 16
ounce, fluid (UK)	UKfl oz	5.8	Table 8, Table 9
ounce, fluid (US)	USfl oz	5.10	Table 8, Table 9
ounce, liquid (US)	liq oz	5.10, footnote	—
ounce-force	ozf	29.3	Table 30

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
ounce-force inch	ozf·in	31.3	Table 31
ounce inch squared	oz·in ²	26.3	Table 29
ounce per square foot	oz/ft ²	17.3	Table 19
ounce per square yard	oz/yd ²	17.3	Table 19
ounce per UK gallon	oz/UKgal	21.3	Table 23
ounce per US gallon	oz/USgal	21.3	Table 23
ounce troy (UK)	oz tr	15.5	Table 16
ounce troy (US)	oz t	15.5	Table 16
parsec	pc	3.3	—
pascal	Pa	33.1.1, Clause 34	Table 32, Table 33, Table 34
pascal second	Pa·s	35.1	Table 35
peck (UK)	—	5.8	Table 9
peck (US)	pk	5.11	Table 9
perch	—	3.6 and Clause 3, Note 15	—
peta (prefix)	P	2.1	Table 1
Petrograd standard	—	Clause 5, Note 4	—
Pferdestärke (metric horsepower, German)	PS	38.2, footnote	Table 39
pico (prefix)	p	2.1	Table 1
pièze	pz	33.1.2	—
pin	—	5.8	—
pint (UK)	UKpt	5.8	Table 7, Table 9
pint, dry (US)	—	5.11	Table 6, Table 9
pint, liquid (US)	liq pt	5.10	Table 7, Table 9
plane angle	—	Clause 8	Table 11
point	—	3.6 and Clause 3, Note 9	—
poise	P	35.2	—
poiseuille (French)	Pl	35.1	—
pole	—	3.6 and Clause 3, Note 15	—
pound	lb	15.4	Table 15, Table 17
pound foot squared	lb·ft ²	26.3	Table 29
pound-force	lbf	29.3	Table 30
pound-force foot	lbf·ft	31.3	Table 31
pound-force hour per square foot	lbf·h/ft ²	35.3	Table 35
pound-force inch	lbf·in	31.3	Table 31
pound-force per square foot	lbf/ft ²	33.1.4	Table 34
pound-force per square inch (pressure)	lbf/in ² (p.s.i.)	33.1.4	Table 32, Table 33
pound-force per square inch (absolute)	p.s.i.a	33.4	—
pound-force per square inch (gauge)	p.s.i.g	33.4	—
pound-force per square inch (stress)	lbf/in ²	Clause 34	Table 32, Table 33
pound-force second per square inch	lbf·s/in ²	35.3	—
pound-force second per square foot	lbf·s/ft ²	35.3	Table 35
pound inch squared	lb·in ²	26.3	Table 29

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
pound per acre	lb/acre	17.3	Table 19
pound per cubic foot	lb/ft ³	20.3	Table 22
pound per cubic inch	lb/in ³	20.3	Table 22
pound per foot	lb/ft	16.3	Table 18
pound per foot hour	lb/(ft·h)	35.3	Table 35, Note
pound per foot second	lb/(ft·s)	35.3	Table 35
pound per (UK) gallon	lb/UKgal	20.3	Table 22
pound per (US) gallon	lb/USgal	20.3	Table 22
pound per hour	lb/h	23.3	Table 25
pound per inch	lb/in	16.3	Table 18
pound per mile	lb/mile	16.3	Table 18
pound per second	lb/s	23.3	Table 25
pound per thousand square feet	lb/1 000 ft ²	17.3	Table 19
pound per yard	lb/yd	16.3	Table 18
pound troy (US)	—	15.5	—
poundal	pdl	29.3	Table 30
poundal foot	pdl·ft	31.3	Table 31
poundal per square foot	pdl/ft ²	33.1.4	Table 32
poundal second per square foot	pdl·s/ft ²	35.3	Table 35
power	—	Clause 38	Table 39
pressure	—	Clause 33	Table 32, Table 33, Table 34
pressure, absolute	—	33.4	—
pressure, gauge	—	33.4	—
quadrillion	—	2.2	Table 2
quart (UK)	UKqt	5.8	Table 9
quart, dry (US)	dry qt	5.11	Table 9
quart, liquid (US)	liq qt	5.10	Table 9
quarter	qr	15.5	—
quintal	q	15.3	—
radian	rad	8.1	Table 11
radian per minute	rad/min	12.2	Table 13
radian per second	rad/s	12.1	Table 13
Rankine, degree	°R	39.4	Table 40
Raummeter (German)	Rm	5.5	—
Redwood second	—	36.5	—
refrigeration, ton of	—	38.4	—
relative density	—	Clause 20, footnote	—
release rate, heat	—	Clause 49	Table 51
resistivity, thermal	—	Clause 48	Table 50
revolution per minute (angular velocity)	rev/min r/min	12.2	Table 13
revolution per minute (rotational frequency)	rev/min r/min	13.3	—
revolution per second (angular velocity)	rev/s r/s	12.2	Table 13
revolution per second (rotational frequency)	rev/s r/s	13.2	—
reyn	—	35.3, footnote	—

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
right angle	└	8.1	Table 11
rod	—	3.6 and Clause 3, Note 15	—
rood	—	4.4 and Clause 4, Note 1	Table 4
rotation, speed of	—	Clause 12, footnote	—
rotational speed	—	Clause 12, footnote	—
rotational velocity	—	Clause 12, footnote	—
Saybolt universal scale	—	36.5	—
Scale, International Practical Temperature of 1968	IPTS - 68	39.6	—
Scale, International Temperature of 1990	ITS - 90	39.6	—
scruple (apothecaries')	—	15.5	—
second (angle)	"	8.2	Table 11
second (time)	s	10.1	—
second, ephemeris	—	10.2	—
second, inverse	s ⁻¹	13.1, 13.2	—
second moment of area	—	Clause 10	Table 10
section, modulus of	—	Clause 6	use Table 6
short hundredweight (US)	sh cwt	15.5	Table 17
short ton (US)	sh ton	15.5	Table 17
slug	—	15.6 and Clause 15, Note 5	Table 15
slug hour per foot second squared	slug·h/(ft·s ²)	35.3	use Table 35
slug per foot second	slug/(ft·s)	35.3	use Table 35
solid angle	—	Clause 9	—
specific energy	—	Clause 40	Table 41
specific enthalpy	—	40.1	Table 41
specific entropy	—	Clause 43	see Table 45
specific gravity	—	Clause 20, footnote	—
specific heat	—	Clause 42, footnote	—
specific heat, volume basis	—	Clause 44, footnote	—
specific heat capacity	—	Clause 42	Table 45
specific latent heat	—	40.1	Table 41
specific surface	—	Clause 18	Table 20
specific volume	—	Clause 22	Table 24
speed	—	Clause 11	Table 12
speed of rotation	—	Clause 12, footnote	—
speed, rotational	—	Clause 12, footnote	—
square centimetre	cm ²	4.2	use Table 4
square centimetre per milligram	cm ² /mg	18.2	use Table 20
square decimetre	dm ²	4.2	use Table 4
square foot	ft ²	4.4	Table 4
square foot hour degree Fahrenheit per British thermal unit inch	ft ² ·h·°F/(Btu·in)	—	Table 50
square foot per gallon	ft ² /gal	Clause 19	Table 21
square foot per ounce	ft ² /oz	18.3	Table 20
square foot (thousand) per pound	1 000 ft ² /lb	18.3	Table 20
square inch	in ²	4.4	Table 4, Table 5
square kilometre	km ²	4.2	use Table 4
square metre	m ²	4.1	Table 4

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
square metre per gram	m ² /g	18.2	<i>use</i> Table 20
square metre per kilogram	m ² /kg	18.1	Table 20
square metre per litre	m ² /l	Clause 19	Table 21
square mile	mile ²	4.4	Table 4
square mile per UK ton	mile ² /ton	18.3	Table 20
square millimetre	mm ²	4.2	Table 5, <i>use also</i> Table 4
square millimetre per milligram	mm ² /mg	18.2	Table 20, footnote
square yard	yd ²	4.4	Table 4
square yard per gallon	yd ² /gal	Clause 19	Table 21
square yard per ounce	yd ² /oz	18.3	Table 20
standard, Petrograd	—	Clause 5 , Note 4	—
standard atmosphere	atm	33.3	Table 33
standard gravity	<i>g_n</i>	14.4	Table 14
steradian	sr	Clause 9	—
stère (French)	st	5.5	—
sthène (French)	sn	29.2	Table 30, Note
stokes	St	36.2	<i>use</i> Table 36
stone	—	15.5	—
stress	—	Clause 34	Table 32, Table 33, Table 34
surface, specific	—	Clause 18	Table 20
technical atmosphere	at	33.3	Table 32, footnote
technical atmosphere, absolute (German)	ata	33.4	—
technical atmosphere, gauge (German)	atü	33.4	—
temperature	—	Clause 39	Table 40
temperature, thermodynamic	—	Clause 39	—
temperature difference	—	Clause 39	—
temperature interval	—	Clause 39	—
tera (prefix)	T	2.1	Table 1
tex	—	16.2	—
therm	—	37.2	—
therm per UK gallon	therm/UKgal	41.3	Table 42
thermal conductance	—	Clause 46	Table 48
thermal conductivity	—	Clause 47	Table 49
thermal diffusivity	—	Clause 50	<i>see</i> Table 36
thermal resistivity	—	Clause 48	Table 50
thermie	th	37.2	Table 38, footnote
thermie per litre	th/litre	41.2	Table 42
thermodynamic temperature	—	Clause 39	—
thou	thou	3.6 and Clause 3 , Note 7	—
time	—	Clause 10	—
ton (UK)	ton	15.5	Table 17
ton, assay (UK)	—	15.6 and Clause 15 , Note 3	—
ton, assay (US)	—	15.6 and Clause 15 , Note 4	—
ton, gross (US)	—	15.5	—
ton, long (US)	—	15.5	—
ton, metric	—	15.2	—

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
ton, short (US)	sh ton	15.5	Table 17
ton-force	tonf	29.3	Table 30
ton-force (US)	—	29.3	use Table 30
ton-force foot	tonf·ft	31.3	Table 31
ton-force per square foot	tonf/ft ²	33.1.4	Table 32
ton-force per square inch	tonf/in ²	33.1.4, Clause 34	Table 32
ton mile	UKton·mile	Clause 25	—
ton mile per gallon	UKton·mile/UKgal	Clause 25	—
ton of refrigeration	—	38.4	—
ton per cubic yard	UKton/yd ³	20.3	Table 22
ton per hour	UKton/h	23.3	Table 25
ton per mile	ton/mile	16.3	Table 18
ton per square mile	ton/mile ²	17.3	Table 19
ton per thousand yards	ton/1 000 yd	16.3	Table 18
tonne	t	15.2	Table 17
tonne-calorie	—	37.2	use Table 38
tonne kilometre	t·km	Clause 25	—
tonne kilometre per litre	t·km/l	Clause 25	—
torque	—	Clause 31	Table 31
torr	—	33.2	Table 33
traffic factors	—	Clause 25	Table 27, Table 28
trillion	—	2.2	Table 2
tropical year	—	10.4, footnote	—
troy units	—	15.5	—
unit, atomic mass	u	15.3	—
vacuum values	—	33.4	—
velocity, angular	—	Clause 12	Table 13
velocity, linear	—	Clause 11	Table 12
velocity, rotational	—	Clause 12, footnote	—
viscosity, dynamic	—	Clause 35	Table 35
viscosity, kinematic	—	Clause 36	Table 36
volume	—	Clause 5	Table 6, Table 7, Table 8
volume, specific	—	Clause 22	Table 24
volume rate of flow	—	Clause 24	Table 26
Vollwinkel (German)	—	Clause 8, Note 3	—
watt	W	38.1	Table 39
watt per cubic metre	W/m ³	49.1	Table 51
watt per metre degree Celsius	W/(m·°C)	47.2	Table 49, footnote
watt per metre kelvin	W/(m·K)	47.1	Table 49
watt per square inch	W/in ²	45.3	Table 47
watt per square metre	W/m ²	45.1	Table 47
watt per square metre degree Celsius	W/(m ² ·°C)	46.2	Table 48, footnote
watt per square metre kelvin	W/(m ² ·K)	46.1	Table 48
week	—	10.4	—

Term	Symbol or abbreviation	Textual reference and important notes	Table reference
weight	—	Clause 30	Table 15, Table 16, Table 17, Table 30
work	—	Clause 37	Table 37, Table 38
yard	yd	3.4	Table 3
year	a	10.4	—
year, calendar	—	10.4, footnote	—
year, light	l.y.	3.3	—
year, tropical	—	10.4, footnote	—
yocto (prefix)	y	2.1	Table 1
yotta (prefix)	Y	2.1	Table 1
zepto (prefix)	z	2.1	Table 1
zetta (prefix)	Z	2.1	Table 1

Bibliography

Standards publications

- BS 718:1991, *Specification for density hydrometers.*
- BS 874-1:1986, *Methods for determining thermal insulating properties — Part 1: Introduction, definitions and principles of measurement.*
- BS 947:1970, *Specification for a universal system for designating linear density of textiles (Tex system).*
- BS 1797:1987, *Schedule for tables for use in the calibration of volumetric glassware.*
- BS 2520:1983, *Specification for barometer conventions and tables, their application and use.*
- BS 2856:1973, *Precise conversion of inch and metric sizes on engineering drawings.*
- BS 5555:1993, *Specification for SI units and recommendations for the use of their multiples and of certain other units.*
- BS 5775-0:1993, *Specification for quantities, units and symbols — Part 0: General principles.*
- BS 5775-1:1993, *Specification for quantities, units and symbols — Part 1: Space and time.*
- BS 5775-2:1993, *Specification for quantities, units and symbols — Part 2: Periodic and related phenomena.*
- BS 5775-3:1993, *Specification for quantities, units and symbols — Part 3: Mechanics.*
- BS 5775-4:1993, *Specification for quantities, units and symbols — Part 4: Heat.*
- BS 5775-5:1993, *Specification for quantities, units and symbols — Part 5: Electricity and magnetism.*
- BS 5775-6:1993, *Specification for quantities, units and symbols — Part 6: Light and related electromagnetic radiations.*
- BS 5775-7:1993, *Specification for quantities, units and symbols — Part 7: Acoustics.*
- BS 5775-8:1993, *Specification for quantities, units and symbols — Part 8: Physical chemistry and molecular physics.*
- BS 5775-9:1993, *Specification for quantities, units and symbols — Part 9: Atomic and nuclear physics.*
- BS 5775-10:1993, *Specification for quantities, units and symbols — Part 10: Nuclear reactions and ionizing radiations.*
- BS 5775-11:1993, *Specification for quantities, units and symbols — Part 11: Mathematical signs and symbols for use in the physical sciences and technology.*
- BS 5775-12:1993, *Specification for quantities, units and symbols — Part 12: Characteristic numbers.*
- BS 5775-13:1993, *Specification for quantities, units and symbols — Part 13: Solid state physics.*
- BS EN ISO 18265:2003, *Metallic materials — Conversion of hardness values.*

Other publications

- [1] GREAT BRITAIN. The Weights and Measures Act 1985, as amended. London: The Stationery Office.
- [2] GREAT BRITAIN. The Units of Measurement Regulations 1995, SI No. 1995/1804. London: The Stationery Office.
- [3] INTERNATIONAL BUREAU OF WEIGHTS AND MEASURES. *The International System of Units (SI)*. 7th edition 1998³³.
- [4] ESDU. Item No. 68036 *Introductory memorandum on the viscosity of liquids and the classification of lubricating oils*³⁴.
- [5] GREAT BRITAIN. NATIONAL PHYSICAL LABORATORY. *Changing to the metric system. Conversion factors, symbols and definitions*, 5th edition 1979. London: HMSO³⁵.

³³) Available from International Bureau of Weights and Measures at www.bipm.org

³⁴) Obtainable from: ESDU International plc, 27 Corsham Street, London, N1 6UA. www.esdu.com

³⁵) Out of print, but may be available to order. Further information obtainable from www.tso.co.uk

BSI — British Standards Institution

BSI is the independent national body responsible for preparing British Standards. It presents the UK view on standards in Europe and at the international level. It is incorporated by Royal Charter.

Revisions

British Standards are updated by amendment or revision. Users of British Standards should make sure that they possess the latest amendments or editions.

It is the constant aim of BSI to improve the quality of our products and services. We would be grateful if anyone finding an inaccuracy or ambiguity while using this British Standard would inform the Secretary of the technical committee responsible, the identity of which can be found on the inside front cover.
Tel: +44 (0)20 8996 9000. Fax: +44 (0)20 8996 7400.

BSI offers members an individual updating service called PLUS which ensures that subscribers automatically receive the latest editions of standards.

Buying standards

Orders for all BSI, international and foreign standards publications should be addressed to Customer Services. Tel: +44 (0)20 8996 9001.
Fax: +44 (0)20 8996 7001. Email: orders@bsi-global.com. Standards are also available from the BSI website at <http://www.bsi-global.com>.

In response to orders for international standards, it is BSI policy to supply the BSI implementation of those that have been published as British Standards, unless otherwise requested.

Information on standards

BSI provides a wide range of information on national, European and international standards through its Library and its Technical Help to Exporters Service. Various BSI electronic information services are also available which give details on all its products and services. Contact the Information Centre.
Tel: +44 (0)20 8996 7111. Fax: +44 (0)20 8996 7048. Email: info@bsi-global.com.

Subscribing members of BSI are kept up to date with standards developments and receive substantial discounts on the purchase price of standards. For details of these and other benefits contact Membership Administration.
Tel: +44 (0)20 8996 7002. Fax: +44 (0)20 8996 7001.
Email: membership@bsi-global.com.

Information regarding online access to British Standards via British Standards Online can be found at <http://www.bsi-global.com/bsonline>.

Further information about BSI is available on the BSI website at <http://www.bsi-global.com>.

Copyright

Copyright subsists in all BSI publications. BSI also holds the copyright, in the UK, of the publications of the international standardization bodies. Except as permitted under the Copyright, Designs and Patents Act 1988 no extract may be reproduced, stored in a retrieval system or transmitted in any form or by any means – electronic, photocopying, recording or otherwise – without prior written permission from BSI.

This does not preclude the free use, in the course of implementing the standard, of necessary details such as symbols, and size, type or grade designations. If these details are to be used for any other purpose than implementation then the prior written permission of BSI must be obtained.

Details and advice can be obtained from the Copyright & Licensing Manager.
Tel: +44 (0)20 8996 7070. Fax: +44 (0)20 8996 7553.
Email: copyright@bsi-global.com.